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PASSWORD:

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* * * * * * * * * *
                     Welcome to STN International
                 Web Page for STN Seminar Schedule - N. America
NEWS
         DEC 01
NEWS
                 ChemPort single article sales feature unavailable
NEWS
         FEB 02
                 Simultaneous left and right truncation (SLART) added
                 for CERAB, COMPUAB, ELCOM, and SOLIDSTATE
NEWS
         FEB 02
                 GENBANK enhanced with SET PLURALS and SET SPELLING
NEWS
         FEB 06
                 Patent sequence location (PSL) data added to USGENE
NEWS
         FEB 10
                 COMPENDEX reloaded and enhanced
NEWS
      7
         FEB 11
                 WTEXTILES reloaded and enhanced
NEWS
      8 FEB 19
                 New patent-examiner citations in 300,000 CA/CAplus
                 patent records provide insights into related prior
                 art
NEWS
      9
         FEB 19
                 Increase the precision of your patent queries -- use
                 terms from the IPC Thesaurus, Version 2009.01
NEWS 10
         FEB 23
                 Several formats for image display and print options
                 discontinued in USPATFULL and USPAT2
         FEB 23
                 MEDLINE now offers more precise author group fields
NEWS 11
                 and 2009 MeSH terms
                 TOXCENTER updates mirror those of MEDLINE - more
NEWS 12
         FEB 23
                 precise author group fields and 2009 MeSH terms
NEWS 13
         FEB 23
                 Three million new patent records blast AEROSPACE into
                 STN patent clusters
NEWS 14
         FEB 25
                 USGENE enhanced with patent family and legal status
                 display data from INPADOCDB
         MAR 06
                 INPADOCDB and INPAFAMDB enhanced with new display
NEWS 15
                 formats
NEWS 16
         MAR 11
                 EPFULL backfile enhanced with additional full-text
                 applications and grants
         MAR 11
                 ESBIOBASE reloaded and enhanced
NEWS 17
                 CAS databases on STN enhanced with new super role
NEWS 18
         MAR 20
                 for nanomaterial substances
                 CA/CAplus enhanced with more than 250,000 patent
NEWS 19
         MAR 23
                 equivalents from China
NEWS 20
         MAR 30
                 IMSPATENTS reloaded and enhanced
NEWS 21
         APR 03
                 CAS coverage of exemplified prophetic substances
                  enhanced
NEWS 22
         APR 07
                 STN is raising the limits on saved answers
NEWS 23
         APR 24
                 CA/CAplus now has more comprehensive patent assignee
                  information
                 USPATFULL and USPAT2 enhanced with patent
NEWS 24
         APR 26
                 assignment/reassignment information
NEWS 25
         APR 28
                 CAS patent authority coverage expanded
NEWS 26
         APR 28
                 ENCOMPLIT/ENCOMPLIT2 search fields enhanced
NEWS 27
         APR 28
                 Limits doubled for structure searching in CAS
                 REGISTRY
NEWS 28 MAY 08
                 STN Express, Version 8.4, now available
NEWS 29
         MAY 11
                 STN on the Web enhanced
```

NEWS 30 MAY 11 BEILSTEIN substance information now available on STN Easy

NEWS 31 MAY 14 DGENE, PCTGEN and USGENE enhanced with increased limits for exact sequence match searches and introduction of free HIT display format

NEWS 32 MAY 15 INPADOCDB and INPAFAMDB enhanced with Chinese legal status data

NEWS EXPRESS JUNE 27 08 CURRENT WINDOWS VERSION IS V8.3, AND CURRENT DISCOVER FILE IS DATED 06 APRIL 2009.

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=> file caplus
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SINCE FILE TOTAL ENTRY SESSION 0.66 0.66

FULL ESTIMATED COST

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FILE COVERS 1907 - 21 May 2009 VOL 150 ISS 21
FILE LAST UPDATED: 20 May 2009 (20090520/ED)
REVISED CLASS FIELDS (/NCL) LAST RELOADED: Feb 2009
USPTO MANUAL OF CLASSIFICATIONS THESAURUS ISSUE DATE: Feb 2009

CAplus now includes complete International Patent Classification (IPC) reclassification data for the third quarter of 2008.

CAS Information Use Policies apply and are available at:

http://www.cas.org/legal/infopolicy.html

This file contains CAS Registry Numbers for easy and accurate

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=> e us20070039665/pn
E1 1 US20070039661/PN
E2
                  US20070039662/PN
            1
E3
            1 --> US20070039665/PN
E4
            1 US20070039666/PN
E5
            1
                  US20070039667/PN
            1
                 US20070039668/PN
Ε7
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                  US20070039671/PN
            1
E10
            1
                 US20070039672/PN
E11
            1
                 US20070039674/PN
E12
                  US20070039675/PN
=> s e3;d all
    1 US20070039665/PN
T.1
L1
    ANSWER 1 OF 1 CAPLUS COPYRIGHT 2009 ACS on STN
ΑN
     2004:847649 CAPLUS
DN
     141:353637
ED
     Entered STN: 15 Oct 2004
     Pretreatment of Ag-alloy surface with organosulfur compounds for
ΤI
     tarnishing prevention
ΙN
     Johns, Peter Gammon; Harrison, Clare Elizabeth
PA
    Middlesex Silver Co. Limited, UK
SO
    PCT Int. Appl., 43 pp.
    CODEN: PIXXD2
DT
    Patent
    English
LA
TC
     ICM C23F011-16
     56-6 (Nonferrous Metals and Alloys)
CC
FAN.CNT 1
    PATENT NO.
                        KIND
                                DATE
                                           APPLICATION NO.
                                                                 DATE
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                                            _____
                                           WO 2004-GB1373 20040330
     WO 2004087996
                         A1
                                20041014
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             CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, EG, ES, FI, GB, GD,
             GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC,
             LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NA, NI,
             NO, NZ, OM, PG, PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, SY,
             TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, YU, ZA, ZM, ZW
         RW: BW, GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZM, ZW, AM, AZ,
             BY, KG, KZ, MD, RU, TJ, TM, AT, BE, BG, CH, CY, CZ, DE, DK, EE,
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             SK, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN,
             TD, TG
     AU 2004225693
                                            AU 2004-225693
                          Α1
                                20041014
                                                                    20040330
     CA 2520807
                          Α1
                                20041014
                                            CA 2004-2520807
                                                                    20040330
                                           EP 2004-724313
     EP 1611267
                         Α1
                                20060104
                                                                    20040330
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     CN 1780937
                                20060531
                                            CN 2004-80011375
                         Α
                                                                    20040330
     JP 2006523266
                          Τ
                                            JP 2006-506057
                                20061012
                                                                    20040330
                    A 20070831
A 20060510
A1 20070222
A 20030331
W 20040330
     IN 2005DN04346
                         Α
                                20070831
                                            IN 2005-DN4346
                                                                   20050926
                                                                   20050928
     MX 2005010452
                                            MX 2005-10452
     US 20070039665
                                            US 2005-551476
                                                                   20050929 <--
PRAI GB 2003-7290
     WO 2004-GB1373
CLASS
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PATENT NO. CLASS PATENT FAMILY CLASSIFICATION CODES
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               ____
WO 2004087996
              ICM
                      C23F011-16
               IPCI C23F0011-16 [ICM, 7]; C23F0011-10 [ICM, 7, C*]
               IPCR C23F0011-10 [I,C*]; C23F0011-16 [I,A]
               ECLA C23F011/16; C23F011/16B
               IPCI C23F0011-16 [ICM, 7]; C23F0011-10 [ICM, 7, C*]
AU 2004225693
               IPCR C23F0011-10 [I,C*]; C23F0011-16 [I,A]
                ECLA C23F011/16; C23F011/16B
CA 2520807
               IPCI C23F0011-16 [ICM, 7]; C23F0011-10 [ICM, 7, C*]
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               ECLA C23F011/16; C23F011/16B
EP 1611267
               IPCI C23F0011-16 [ICM, 7]; C23F0011-10 [ICM, 7, C*]
               ECLA C23F011/16; C23F011/16B
               CN 1780937
               ECLA C23F011/16; C23F011/16B
                      C23F0011-00 [I,A]; C22C0005-06 [I,A]; C22C0005-08 [I,A]
JP 2006523266
               IPCI
               IPCR
                      C23F0011-00 [I,C]; C23F0011-00 [I,A]; C22C0005-06
                      [I,C]; C22C0005-06 [I,A]; C22C0005-08 [I,A];
                      C23F0011-10 [I,C*]; C23F0011-16 [I,A]
                FTERM 4K062/AA01; 4K062/BB21; 4K062/BC22; 4K062/FA16
                      C23F0011-16 [ICM, 7]; C23F0011-10 [ICM, 7, C*]
IN 2005DN04346
               IPCI
MX 2005010452
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                      C23F0011-16 [ICM, 7]; C23F0011-10 [ICM, 7, C*]
                ECLA
                      C23F011/16; C23F011/16B
US 20070039665 IPCI
                      C23G0001-00 [I,A]; C23C0022-58 [I,A]; C23C0022-05
                      [I,C*]
               NCL
                      148/271.000; 134/002.000
AB
    The Ag alloys containing minor Ge (especially Ag-Cu-Ge alloys) to decrease the
fire
    stain discoloration are pretreated on the surface with an alkanethiol,
    alkyl thioglycollate, dialkyl sulfide, or dialkyl disulfide to prevent
    tarnishing. The treatment with organosulfur compds. is suitable for
    and the subsequent extended display without special packaging. The
    Ag-alloy surface is optionally treated with aqueous solution containing an
    alkanethiol, alkyl thioglycollate, dialkyl sulfide, or dialkyl disulfide,
    as well as a mixture of anionic surfactant and amphoteric or nonionic
```

manufactured Ag-alloy articles to prevent tarnished appearance during transit surfactant to solubilize the treatment agent. The typical ternary alloy

contains Ag 80-96, Cu 1-19.9, and Ge 0.1-5%.

ST silver copper germanium alloy tarnishing prevention organosulfur

ΙT Surfactants

> (anionic, in tarnishing prevention; Ag-alloy surface treated with organosulfur compds. for tarnishing prevention)

ΤT Surfactants

> (in tarnishing prevention; Ag-alloy surface treated with organosulfur compds. for tarnishing prevention)

ΙT Surfactants

> (nonionic, in tarnishing prevention; Ag-alloy surface treated with organosulfur compds. for tarnishing prevention)

ΙT

(prevention of; Ag-alloy surface treated with organosulfur compds. for tarnishing prevention)

Thioethers ΙT

Thiols, uses

RL: TEM (Technical or engineered material use); USES (Uses) (tarnishing prevention by; Ag-alloy surface treated with organosulfur compds. for tarnishing prevention)

7440-56-4, Germanium, uses ΤТ

RL: MOA (Modifier or additive use); USES (Uses)

(Ag alloys containing, tarnishing prevention on; Ag-alloy surface treated with organosulfur compds. for tarnishing prevention)

IT 106-94-5, n-Propyl bromide

RL: TEM (Technical or engineered material use); USES (Uses) (solvent, in tarnishing prevention; Ag-alloy surface treated with organosulfur compds. for tarnishing prevention)

IT 2885-00-9, Octadecyl mercaptan 2917-26-2, Cetyl mercaptan RL: CPS (Chemical process); PEP (Physical, engineering or chemical process); PROC (Process)

(tarnishing prevention by; Ag-alloy surface treated with organosulfur compds. for tarnishing prevention)

IT 39282-03-6, Sterling silver 103221-24-5 476614-10-5 476614-12-7 476614-13-8

RL: CPS (Chemical process); PEP (Physical, engineering or chemical process); PROC (Process)

(tarnishing prevention on; Ag-alloy surface treated with organosulfur compds. for tarnishing prevention)

IT 9080-17-5, Ammonium polysulfide

RL: CPS (Chemical process); PEP (Physical, engineering or chemical process); PROC (Process)

(test solution with, for tarnishing; Ag-alloy surface treated with organosulfur compds. for tarnishing prevention)

RE.CNT 9 THERE ARE 9 CITED REFERENCES AVAILABLE FOR THIS RECORD RE

- (1) Carlton, C; US 3503883 A 1970
- (2) Carpenter, J; US 3398003 A 1968 CAPLUS
- (3) Gamon, J; EP 0729398 A 1996 CAPLUS
- (4) Gamon, J; WO 02095082 A 2002 CAPLUS
- (5) Goddard & Sons Ltd J; GB 1070384 A 1967 CAPLUS
- (6) Goddard & Sons Ltd J; GB 1130540 A 1968
- (7) Han, S; JOURNAL OF THE AMERICAN CHEMICAL SOCIETY 2001, V123, P2422 CAPLUS
- (8) Metaleurop Rech; GB 2255348 A 1992 CAPLUS
- (9) Nippon Germanium Lab Co Ltd; EP 1130124 A 2001 CAPLUS

=> file reg;s 106-94-5/rn;d;s 2885-00-9/rn;d;s 2917-26-2/rn;d COST IN U.S. DOLLARS SINCE FILE TOTAL ENTRY SESSION FULL ESTIMATED COST 9.12 9.78 DISCOUNT AMOUNTS (FOR QUALIFYING ACCOUNTS) SINCE FILE TOTAL ENTRY SESSION CA SUBSCRIBER PRICE -0.82-0.82

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STRUCTURE FILE UPDATES: 20 MAY 2009 HIGHEST RN 1147939-89-6 DICTIONARY FILE UPDATES: 20 MAY 2009 HIGHEST RN 1147939-89-6

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TSCA INFORMATION NOW CURRENT THROUGH January 9, 2009.

Please note that search-term pricing does apply when conducting ${\tt SmartSELECT}$ searches.

REGISTRY includes numerically searchable data for experimental and predicted properties as well as tags indicating availability of

experimental property data in the original document. For information on property searching in REGISTRY, refer to:

http://www.cas.org/support/stngen/stndoc/properties.html

L2 1 106-94-5/RN

```
L2
    ANSWER 1 OF 1 REGISTRY COPYRIGHT 2009 ACS on STN
    106-94-5 REGISTRY
RN
ED
    Entered STN: 16 Nov 1984
CN
    Propane, 1-bromo- (CA INDEX NAME)
OTHER NAMES:
CN
    1-Bromopropane
    1-Propyl bromide
CN
    Ascusol MC
CN
CN
    Drysolv
CN
    Leksol
CN
     n-Propyl bromide
CN
     Propyl bromide
MF
     C3 H7 Br
CI
     COM
LC
     STN Files:
                  AGRICOLA, ANABSTR, AQUIRE, BEILSTEIN*, BIOSIS, BIOTECHNO, CA,
       CAPLUS, CASREACT, CBNB, CHEMCATS, CHEMINFORMRX, CHEMLIST, CHEMSAFE, CIN,
       CSCHEM, CSNB, DETHERM*, EMBASE, GMELIN*, HSDB*, IFICDB, IFIPAT, IFIUDB,
       IPA, MEDLINE, MRCK*, MSDS-OHS, NAPRALERT, PIRA, PROMT, PS, RTECS*,
       SPECINFO, SYNTHLINE, TOXCENTER, USPAT2, USPATFULL, USPATOLD
         (*File contains numerically searchable property data)
                    DSL**, EINECS**, TSCA**
     Other Sources:
         (**Enter CHEMLIST File for up-to-date regulatory information)
```

Br-CH2-CH2-CH3

PROPERTY DATA AVAILABLE IN THE 'PROP' FORMAT

4951 REFERENCES IN FILE CA (1907 TO DATE)
50 REFERENCES TO NON-SPECIFIC DERIVATIVES IN FILE CA
4963 REFERENCES IN FILE CAPLUS (1907 TO DATE)

L3 1 2885-00-9/RN

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L3
     ANSWER 1 OF 1 REGISTRY COPYRIGHT 2009 ACS on STN
     2885-00-9 REGISTRY
RN
ED
     Entered STN: 16 Nov 1984
     1-Octadecanethiol (CA INDEX NAME)
CN
OTHER NAMES:
    1-Mercaptooctadecane
CN
CN
    1-Octadecyl mercaptan
CN
    n-Octadecanethiol
CN
    n-Octadecyl mercaptan
CN
    NSC 5545
CN
    Octadecanethiol
CN
    Octadecyl mercaptan
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Octadecylthiol
CN
     Stearyl mercaptan
CN
MF
    C18 H38 S
CI
     STN Files:
                  AGRICOLA, ANABSTR, BEILSTEIN*, BIOSIS, CA, CAPLUS, CASREACT,
LC.
       CHEMCATS, CHEMINFORMRX, CHEMLIST, CIN, CSCHEM, DETHERM*, IFICDB, IFIPAT,
       IFIUDB, MEDLINE, MSDS-OHS, PIRA, PROMT, SPECINFO, TOXCENTER, USPAT2,
      USPATFULL, USPATOLD
        (*File contains numerically searchable property data)
                     EINECS**, NDSL**, TSCA**
     Other Sources:
         (**Enter CHEMLIST File for up-to-date regulatory information)
HS-(CH_2)_{17}-Me
**PROPERTY DATA AVAILABLE IN THE 'PROP' FORMAT**
            2003 REFERENCES IN FILE CA (1907 TO DATE)
             221 REFERENCES TO NON-SPECIFIC DERIVATIVES IN FILE CA
            2011 REFERENCES IN FILE CAPLUS (1907 TO DATE)
T.4
            1 2917-26-2/RN
    ANSWER 1 OF 1 REGISTRY COPYRIGHT 2009 ACS on STN
T. 4
RN
    2917-26-2 REGISTRY
    Entered STN: 16 Nov 1984
ED
    1-Hexadecanethiol (CA INDEX NAME)
OTHER CA INDEX NAMES:
    Hexadecanethiol (6CI)
CN
OTHER NAMES:
CN
    1-Mercaptohexadecane
CN
  Cetyl mercaptan
CN Hexadecyl mercaptan
CN Hexadecylthiol
CN n-Hexadecanethiol
CN
    n-Hexadecyl mercaptan
CN
   n-Hexadecylthiol
CN
    NSC 229611
    NSC 57866
CN
MF
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CI
    COM
                AGRICOLA, ANABSTR, BEILSTEIN*, BIOSIS, CA, CAPLUS, CASREACT,
LC
       CHEMCATS, CHEMINFORMRX, CHEMLIST, CIN, CSCHEM, ENCOMPLIT, ENCOMPLIT2,
       ENCOMPPAT, ENCOMPPAT2, IFICDB, IFIPAT, IFIUDB, MEDLINE, PIRA, SPECINFO,
       SYNTHLINE, TOXCENTER, USPAT2, USPATFULL, USPATOLD
         (*File contains numerically searchable property data)
                     EINECS**, NDSL**, TSCA**
         (**Enter CHEMLIST File for up-to-date regulatory information)
HS-(CH_2)_{15}-Me
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=> e stearyl thioglycollate/cn
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Ε2
            1
                 STEARYL THIOGLYCOLATE/CN
Е3
            0 --> STEARYL THIOGLYCOLLATE/CN
E4
                 STEARYL TITANATE/CN
           1
                 STEARYL TOSYLATE/CN
           1
                STEARYL TRIHYDROXY SILANE/CN
           1
E7
           1
                STEARYL TRIHYDROXYETHYL PROPYLENEDIAMINE DIHYDROFLUORIDE/CN
E8
           1
                STEARYL UNDECENOATE/CN
                 STEARYL UROCANATE/CN
E9
           1
                 STEARYL URSOLATE/CN
E10
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                STEARYL VINYL ETHER/CN
E11
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E12
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                 STEARYL VINYL ETHER HOMOPOLYMER/CN
=> s e3;d
L5
            0 "STEARYL THIOGLYCOLLATE"/CN
L5 HAS NO ANSWERS
L_5
             O SEA FILE=REGISTRY "STEARYL THIOGLYCOLLATE"/CN
=> s e2;d
            1 "STEARYL THIOGLYCOLATE"/CN
L6
   ANSWER 1 OF 1 REGISTRY COPYRIGHT 2009 ACS on STN
1.6
   10220-46-9 REGISTRY
RN
   Entered STN: 16 Nov 1984
ED
   Acetic acid, 2-mercapto-, octadecyl ester (CA INDEX NAME)
OTHER CA INDEX NAMES:
    Acetic acid, mercapto-, octadecyl ester (7CI, 8CI, 9CI)
OTHER NAMES:
CN
   NSC 65478
    Octadecyl mercaptoacetate
CN
CN Octadecyl thioglycolate
CN
    Stearyl thioglycolate
CN
    Thioglycolate octadecyl ester
CN
    Thioglycolic acid octadecyl ester
MF
    C20 H40 O2 S
CI
    COM
LC
    STN Files: BIOSIS, CA, CAPLUS, CHEMCATS, CHEMLIST, CSCHEM, GMELIN*,
      IFICDB, IFIPAT, IFIUDB, MEDLINE, TOXCENTER, USPAT2, USPATFULL, USPATOLD
         (*File contains numerically searchable property data)
    Other Sources: EINECS**, NDSL**, TSCA**
         (**Enter CHEMLIST File for up-to-date regulatory information)
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 $Me^-(CH_2)_{17}^-O^-C^-CH_2^-SH$

PROPERTY DATA AVAILABLE IN THE 'PROP' FORMAT

1614 REFERENCES IN FILE CA (1907 TO DATE)

1621 REFERENCES IN FILE CAPLUS (1907 TO DATE)

179 REFERENCES TO NON-SPECIFIC DERIVATIVES IN FILE CA

50 REFERENCES IN FILE CA (1907 TO DATE)

4 REFERENCES TO NON-SPECIFIC DERIVATIVES IN FILE CA

50 REFERENCES IN FILE CAPLUS (1907 TO DATE)

=> s thioglycollate

L7 15 THIOGLYCOLLATE

=> d 1-15

L7 ANSWER 1 OF 15 REGISTRY COPYRIGHT 2009 ACS on STN

RN 74162-83-7 REGISTRY

ED Entered STN: 16 Nov 1984

CN 8-0xa-3,5-dithia-4-stannadocosanoic acid, 4-octyl-7-oxo-4-[[2-oxo-2-(tetradecyloxy)ethyl]thio]-, tetradecyl ester (CA INDEX NAME)

OTHER NAMES:

CN Octyltin tris(tetradecathioglycollate)

MF C56 H110 O6 S3 Sn

LC STN Files: CA, CAPLUS, CHEMLIST, TOXCENTER, USPATFULL Other Sources: EINECS**

(**Enter CHEMLIST File for up-to-date regulatory information)

PROPERTY DATA AVAILABLE IN THE 'PROP' FORMAT

7 REFERENCES IN FILE CA (1907 TO DATE)

7 REFERENCES IN FILE CAPLUS (1907 TO DATE)

L7 ANSWER 2 OF 15 REGISTRY COPYRIGHT 2009 ACS on STN

RN 72259-65-5 REGISTRY

ED Entered STN: 16 Nov 1984

CN 8-Oxa-3,5-dithia-4-stannadocosanoic acid, 4-butyl-7-oxo-4-[[2-oxo-2-(tetradecyloxy)ethyl]thio]-, tetradecyl ester (CA INDEX NAME)

OTHER CA INDEX NAMES:

CN Acetic acid, 2,2',2''-[(butylstannylidyne)tris(thio)]tris-, tritetradecyl ester (9CI)

OTHER NAMES:

CN Butyltin tris(tetradecathioglycollate)

MF C52 H102 O6 S3 Sn

LC STN Files: CA, CAPLUS, CHEMLIST, TOXCENTER, USPATFULL Other Sources: EINECS**, NDSL**, TSCA**

(**Enter CHEMLIST File for up-to-date regulatory information)

$$\begin{array}{c} \text{O} & \text{S-CH}_2-\text{C-O-(CH}_2)_{13}-\text{Me} \\ \text{Me-(CH}_2)_{13}-\text{O-C-CH}_2-\text{S-Sn-Bu-n} \\ & \text{S-CH}_2-\text{C-O-(CH}_2)_{13}-\text{Me} \\ & \text{S-CH}_2-\text{C-O-(CH}_2)_{13}-\text{Me} \\ & \text{O} \end{array}$$

PROPERTY DATA AVAILABLE IN THE 'PROP' FORMAT

- 7 REFERENCES IN FILE CA (1907 TO DATE)
- 7 REFERENCES IN FILE CAPLUS (1907 TO DATE)
- L7 ANSWER 3 OF 15 REGISTRY COPYRIGHT 2009 ACS on STN
- RN 55400-47-0 REGISTRY
- ED Entered STN: 16 Nov 1984
- CN Acetic acid, mercapto-, 2-ethyl-2-[(mercaptoacetyl)oxy]-1,3-propanediyl ester, polymer with $(all-Z)-\alpha,\alpha',\alpha''-1,2,3-$ propanetriyltris[ω -[(3-carboxy-1-oxo-2-propenyl)oxy]poly[oxy(methyl-1,2-ethanediyl)]] (9CI) (CA INDEX NAME)

OTHER CA INDEX NAMES:

CN Poly[oxy(methyl-1,2-ethanediyl)], α,α',α'' , α'' ,

OTHER NAMES:

- CN Polypropylene glycol glycerol triether tris(hydrogen maleate)-1,1,1-trimethylolpropane trithioglycollate copolymer
- MF (C12 H20 O6 S3 . (C3 H6 O)n (C3 H6 O)n (C3 H6 O)n C15 H14 O12) \times
- CI PMS
- PCT Polyester, Polyether, Polythioester, Polythioester formed, Polythioether, Polythioether formed, Polyvinyl
- LC STN Files: CA, CAPLUS

CM 1

CRN 52297-16-2

CMF (C3 H6 O)n (C3 H6 O)n (C3 H6 O)n C15 H14 O12

CCI IDS, PMS

$$HO_2C-CH=CH-C-O-(C_3H_6)-O-(C_1H_2)$$
 $HO_2C-CH=CH-C-O-(C_3H_6)-O-(C_1H_2)$

PAGE 1-A

CM 2

CRN 10193-96-1 CMF C12 H20 O6 S3

1 REFERENCES IN FILE CA (1907 TO DATE)
1 REFERENCES IN FILE CAPLUS (1907 TO DATE)

L7 ANSWER 4 OF 15 REGISTRY COPYRIGHT 2009 ACS on STN

RN 52486-43-8 REGISTRY

ED Entered STN: 16 Nov 1984

CN Acetic acid, mercapto-, 1,2,3-propanetriyl ester, polymer with $(all-Z)-\alpha,\alpha',\alpha''-1,2,3$ -propanetriyltris[$\omega-[(3-carboxy-1-oxo-2-propenyl)oxy]poly[oxy(methyl-1,2-ethanediyl)]] (9CI) (CAINDEX NAME)$

OTHER CA INDEX NAMES:

CN Poly[oxy(methyl-1,2-ethanediyl)], $\alpha,\alpha',\alpha''-1,2,3-$ propanetriyltris[ω -[(3-carboxy-1-oxo-2-propenyl)oxy]-, (all-Z)-, polymer with 1,2,3-propanetriyl tris(mercaptoacetate) (9CI) OTHER NAMES:

CN Glycerol tris(mercaptoacetate)- α , α ', α ''-1, 2, 3-propanetriyltris(polypropylene glycol) maleate (1:3) polymer

CN Glycerol trithioglycollate-polypropylene glycol glycerol triether tris(hydrogen maleate) copolymer

MF (C9 H14 O6 S3 . (C3 H6 O)n (C3 H6 O)n (C3 H6 O)n C15 H14 O12)x

CI PMS

PCT Polyester, Polyether, Polythioester, Polythioester formed, Polythioether, Polythioether formed, Polyvinyl

LC STN Files: CA, CAPLUS

CM 1

CRN 52297-16-2

CMF (C3 H6 O)n (C3 H6 O)n (C3 H6 O)n C15 H14 O12

CCI IDS, PMS

$$HO_2C-CH=CH-C-O-(C_3H_6)-O-D-CH_2$$
 $HO_2C-CH=CH-C-O-(C_3H_6)-O-D-CH_2-CH-O-CH_2$

PAGE 1-B

CM 2

CRN 14974-53-9 CMF C9 H14 O6 S3

2 REFERENCES IN FILE CA (1907 TO DATE)

2 REFERENCES IN FILE CAPLUS (1907 TO DATE)

L7 ANSWER 5 OF 15 REGISTRY COPYRIGHT 2009 ACS on STN

RN 52080-69-0 REGISTRY

ED Entered STN: 16 Nov 1984

CN D-Valine, 3-mercapto-, mixt. with mercaptoacetic acid bismuth(3+) sodium salt (3:1:3) (9CI) (CA INDEX NAME)

OTHER CA INDEX NAMES:

CN Acetic acid, mercapto-, bismuth(3+) sodium salt (3:1:3), mixt. contg.
 (9CI)

OTHER NAMES:

CN d-Dimethylcysteine-sodium bismuth thioglycollate mixture

FS STEREOSEARCH

MF C5 H11 N O2 S . C2 H4 O2 S . 1/3 Bi . Na

CI MXS

LC STN Files: CA, CAPLUS

CM 1

CRN 150-49-2 (68-11-1)

CMF C2 H4 O2 S . 1/3 Bi . Na

$$\begin{array}{c} \text{O} \\ || \\ \text{HO-C-CH}_2\text{-SH} \end{array}$$

●1/3 Bi(III)

Na

CM 2

CRN 52-67-5

CMF C5 H11 N O2 S

Absolute stereochemistry.

1 REFERENCES IN FILE CA (1907 TO DATE)

1 REFERENCES IN FILE CAPLUS (1907 TO DATE)

L7 ANSWER 6 OF 15 REGISTRY COPYRIGHT 2009 ACS on STN

RN 42249-01-4 REGISTRY

ED Entered STN: 16 Nov 1984

CN Acetic acid, mercapto-, 1,4-butanediyl ester, homopolymer (9CI) (CA INDEX NAME)

OTHER NAMES:

CN 1,4-Butanediol bis(thioglycollate) polymer

MF (C8 H14 O4 S2)x

CI PMS

PCT Polyester, Polysulfide, Polysulfide formed, Polythioether, Polythioether formed

LC STN Files: CA, CAPLUS

CM 1

CRN 10193-95-0 CMF C8 H14 O4 S2

1 REFERENCES IN FILE CA (1907 TO DATE)

1 REFERENCES IN FILE CAPLUS (1907 TO DATE)

```
ANSWER 7 OF 15 REGISTRY COPYRIGHT 2009 ACS on STN
T.7
    42249-00-3 REGISTRY
RN
ΕD
    Entered STN: 16 Nov 1984
    Acetic acid, mercapto-, oxydi-2,1-ethanediyl ester, homopolymer (9CI) (CA
CN
    INDEX NAME)
OTHER NAMES:
    Diethyleneglycolbis(thioglycollate)polymer
MF
    (C8 H14 O5 S2)x
CI
    PMS
PCT Polyester, Polyether, Polysulfide, Polysulfide formed, Polythioether,
    Polythioether formed
LC
    STN Files: CA, CAPLUS
    CM
         1
    CRN 14974-52-8
    CMF C8 H14 O5 S2
        0
HS-CH2-C-O-CH2-CH2-O-CH2-CH2-O-C-CH2-SH
               1 REFERENCES IN FILE CA (1907 TO DATE)
               1 REFERENCES IN FILE CAPLUS (1907 TO DATE)
L7
    ANSWER 8 OF 15 REGISTRY COPYRIGHT 2009 ACS on STN
RN
    40544-92-1 REGISTRY
    Entered STN: 16 Nov 1984
ED
    Acetic acid, mercapto-, 1,2-ethanediylbis(oxy-2,1-ethanediyl) ester,
    homopolymer (9CI) (CA INDEX NAME)
OTHER NAMES:
    Triethylene glycol bis(thioglycollate) prepolymer
CN
CN
    Triethyleneglycolbisthioglycollate polymer
MF
    (C10 H18 O6 S2)x
CI
    PMS
PCT Polyester, Polyether, Polysulfide, Polysulfide formed, Polythioether,
    Polythioether formed
LC
    STN Files: CA, CAPLUS
    CM
         1
    CRN 10193-94-9
    CMF C10 H18 O6 S2
        0
HS-CH2-C-O-CH2-CH2-O-CH2-CH2-O-CH2-CH2-O-CH2-SH
               1 REFERENCES IN FILE CA (1907 TO DATE)
              1 REFERENCES IN FILE CAPLUS (1907 TO DATE)
    ANSWER 9 OF 15 REGISTRY COPYRIGHT 2009 ACS on STN
L7
    37019-53-7 REGISTRY
RN
ED
    Entered STN: 16 Nov 1984
    Acetic acid, 2-[(1,4-dihydro-3-methyl-1,4-dioxo-2-naphthalenyl)thio]-,
CN
    sodium salt (1:1) (CA INDEX NAME)
OTHER CA INDEX NAMES:
    Acetic acid, [(1,4-dihydro-3-methyl-1,4-dioxo-2-naphthalenyl)thio]-,
```

sodium salt (9CI)

OTHER NAMES:

CN 2-Methyl-1:4-naphthaquinone-3-thioglycollate sodium salt

MF C13 H10 O4 S . Na

LC STN Files: CA, CAPLUS

CRN (6325-58-2)

Na

1 REFERENCES IN FILE CA (1907 TO DATE)

1 REFERENCES IN FILE CAPLUS (1907 TO DATE)

L7 ANSWER 10 OF 15 REGISTRY COPYRIGHT 2009 ACS on STN

RN 36118-61-3 REGISTRY

ED Entered STN: 16 Nov 1984

CN Acetic acid, 2,2',2''-[(phenylstannylidyne)tris(thio)]tris-, triisooctyl ester (9CI) (CA INDEX NAME)

OTHER NAMES:

CN Monophenyltintris(isooctyl)thioglycollate

CN Phenyltin tris(isooctyl thioglycolate)

MF C36 H62 O6 S3 Sn

CI IDS

LC STN Files: CA, CAPLUS, IFICDB, IFIPAT, IFIUDB, USPATFULL, USPATOLD

3 REFERENCES IN FILE CA (1907 TO DATE)

2 REFERENCES TO NON-SPECIFIC DERIVATIVES IN FILE CA

3 REFERENCES IN FILE CAPLUS (1907 TO DATE)

L7 ANSWER 11 OF 15 REGISTRY COPYRIGHT 2009 ACS on STN

RN 6780-12-7 REGISTRY

ED Entered STN: 16 Nov 1984

CN Acetic acid, mercapto-, triester with thioantimonic acid (H3SbS3), trisodium salt (8CI) (CA INDEX NAME)

OTHER NAMES:

CN Sodium antimonylthioglycollate

MF C6 H9 O6 S3 Sb . 3 Na

CRN (736072-12-1)

●3 Na

```
L7
    ANSWER 12 OF 15 REGISTRY COPYRIGHT 2009 ACS on STN
     5421-46-5 REGISTRY
ED
    Entered STN: 16 Nov 1984
    Acetic acid, 2-mercapto-, ammonium salt (1:1) (CA INDEX NAME)
OTHER CA INDEX NAMES:
   Acetic acid, mercapto-, monoammonium salt (8CI, 9CI)
OTHER NAMES:
CN
    Ammonium mercaptoacetate
CN
    Ammonium thioglycolate
    Ammonium thioglycollate
CN
CN
    Thiofaco A-50
CN
     Thioglycolic acid ammonium salt
     860540-22-3, 8046-21-7, 67124-12-3, 34316-71-7
DR
MF
    C2 H4 O2 S . H3 N
CI
     COM
LC
                 AQUIRE, BIOSIS, CA, CAPLUS, CASREACT, CBNB, CHEMCATS,
     STN Files:
       CHEMLIST, CIN, CSCHEM, CSNB, EMBASE, HSDB*, IFICDB, IFIPAT, IFIUDB, IPA,
      MSDS-OHS, PROMT, RTECS*, TOXCENTER, ULIDAT, USPAT2, USPATFULL, USPATOLD
         (*File contains numerically searchable property data)
     Other Sources: DSL**, EINECS**, TSCA**
         (**Enter CHEMLIST File for up-to-date regulatory information)
CRN
    (68-11-1)
```

● NH3

PROPERTY DATA AVAILABLE IN THE 'PROP' FORMAT

542 REFERENCES IN FILE CA (1907 TO DATE) 3 REFERENCES TO NON-SPECIFIC DERIVATIVES IN FILE CA 542 REFERENCES IN FILE CAPLUS (1907 TO DATE) ANSWER 13 OF 15 REGISTRY COPYRIGHT 2009 ACS on STN 814-71-1 REGISTRY

Entered STN: 16 Nov 1984
Acetic acid, 2-mercapto-, calcium salt (2:1) (CA INDEX NAME) CN

OTHER CA INDEX NAMES:

Acetic acid, mercapto-, calcium salt (2:1) (8CI, 9CI) CN OTHER NAMES:

Calcium thioglycolate CN

CN Calcium thioglycollate

CN Depil

L7RN

```
CN
    Ebacream
CN
    Jully
CN
    Surgex
CN
    Vikor
    C2 H4 O2 S . 1/2 Ca
MF
CI
                  AGRICOLA, BIOSIS, CA, CAPLUS, CHEMCATS, CHEMLIST, CIN,
LC
       CSCHEM, EMBASE, HSDB*, IFICDB, IFIPAT, IFIUDB, IPA, MRCK*, MSDS-OHS,
       PROMT, TOXCENTER, USPATFULL, USPATOLD
         (*File contains numerically searchable property data)
     Other Sources: DSL**, EINECS**, TSCA**
         (**Enter CHEMLIST File for up-to-date regulatory information)
CRN
    (68-11-1)
   0
HO-C-CH_2-SH
  ●1/2 Ca
             139 REFERENCES IN FILE CA (1907 TO DATE)
               1 REFERENCES TO NON-SPECIFIC DERIVATIVES IN FILE CA
             140 REFERENCES IN FILE CAPLUS (1907 TO DATE)
    ANSWER 14 OF 15 REGISTRY COPYRIGHT 2009 ACS on STN
L7
     645-74-9 REGISTRY
RN
     Entered STN: 16 Nov 1984
ED
CN
     Aurate(1-), [mercaptoacetato(2-)-0,S]-, calcium (2:1) (9CI) (CA INDEX
     NAME)
OTHER CA INDEX NAMES:
    Acetic acid, mercapto-, calcium gold(1+) salt (2:1:2) (8CI)
    Acetic acid, mercapto-, gold complex
    Calcium aurothioglycolate
    Calcium aurothioglycollate
CN
CN
    Myoral
CN
    Nedaurine
DR
    16925-54-5
    C2 H4 O2 S . Au . 1/2 Ca
MF
     STN Files: CA, CAPLUS, CHEMLIST
LC
     Other Sources: EINECS**
         (**Enter CHEMLIST File for up-to-date regulatory information)
    (68-11-1)
CRN
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Au(I)

●1/2 Ca

1 REFERENCES IN FILE CA (1907 TO DATE)
1 REFERENCES IN FILE CAPLUS (1907 TO DATE)

L7 ANSWER 15 OF 15 REGISTRY COPYRIGHT 2009 ACS on STN

RN 539-54-8 REGISTRY

ED Entered STN: 16 Nov 1984

CN Acetic acid, 2-[(5-oxo-1,3,2-oxathiastibolan-2-y1)thio]-, sodium salt (1:1) (CA INDEX NAME)

OTHER CA INDEX NAMES:

CN 1,3,2-Oxathiastibolane, acetic acid deriv.

CN Acetic acid, [(5-oxo-1,3,2-oxathiastibolan-2-yl)thio]-, sodium salt (8CI, 9CI)

OTHER NAMES:

CN Antimony sodium thioacetate

CN Antimony sodium thioglycollate

DR 1186-45-4

MF C4 H5 O4 S2 Sb . Na

LC STN Files: CA, CAPLUS, MRCK*, TOXCENTER, USAN (*File contains numerically searchable property data)

CRN (1843-43-2)

● Na

- 3 REFERENCES IN FILE CA (1907 TO DATE)
- 3 REFERENCES IN FILE CAPLUS (1907 TO DATE)

```
=> e octadecyl thioglycollate/cn
Ε1
            1
                  OCTADECYL THIOCYANATE/CN
E2
                  OCTADECYL THIOGLYCOLATE/CN
Е3
            0 --> OCTADECYL THIOGLYCOLLATE/CN
E4
            1
               OCTADECYL THIOPEROXYDIPHOSPHATE/CN
E5
            1
                  OCTADECYL THIOPYROPHOSPHATE/CN
Ε6
            1
                 OCTADECYL THIOPYROPHOSPHATE, (C18H37O)4P2O2S/CN
                 OCTADECYL THIOPYROPHOSPHATE, COMPD. WITH HGI2/CN
E7
            1
E8
            1
                 OCTADECYL THIOSILANE ((C18H37S)4SI)/CN
E9
            1
                 OCTADECYL TITANATE/CN
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E10
            1
                  OCTADECYL TITANATE(IV) ((C18H37O)(HO)3TI)/CN
E11
            1
                  OCTADECYL TITANATE(IV) ((C18H37O)4TI)/CN
E12
                   OCTADECYL TITANATE, (C18H37O)4TI/CN
             1
=> s e2;d
             1 "OCTADECYL THIOGLYCOLATE"/CN
L8
    ANSWER 1 OF 1 REGISTRY COPYRIGHT 2009 ACS on STN
L8
    10220-46-9 REGISTRY
    Entered STN: 16 Nov 1984
CN
    Acetic acid, 2-mercapto-, octadecyl ester (CA INDEX NAME)
OTHER CA INDEX NAMES:
    Acetic acid, mercapto-, octadecyl ester (7CI, 8CI, 9CI)
OTHER NAMES:
    NSC 65478
CN
CN
    Octadecyl mercaptoacetate
CN
    Octadecyl thioglycolate
CN
     Stearyl thioglycolate
CN
     Thioglycolate octadecyl ester
CN
     Thioglycolic acid octadecyl ester
MF
     C20 H40 O2 S
CI
     COM
LC
     STN Files: BIOSIS, CA, CAPLUS, CHEMCATS, CHEMLIST, CSCHEM, GMELIN*,
       IFICDB, IFIPAT, IFIUDB, MEDLINE, TOXCENTER, USPAT2, USPATFULL, USPATOLD
         (*File contains numerically searchable property data)
     Other Sources: EINECS**, NDSL**, TSCA**
         (**Enter CHEMLIST File for up-to-date regulatory information)
Me^-(CH_2)_{17}-O^-C^-CH_2^-SH
**PROPERTY DATA AVAILABLE IN THE 'PROP' FORMAT**
              50 REFERENCES IN FILE CA (1907 TO DATE)
               4 REFERENCES TO NON-SPECIFIC DERIVATIVES IN FILE CA
              50 REFERENCES IN FILE CAPLUS (1907 TO DATE)
=> e octadecylthioglycollate/cn
                  OCTADECYLTHIO/CN
E1
            1
                  OCTADECYLTHIO-1, 4-BENZOQUINONE/CN
E2
             1
             0 --> OCTADECYLTHIOGLYCOLLATE/CN
E3
                 OCTADECYLTHIOHYDROQUINONE/CN
E4
             1
                  OCTADECYLTHIOL/CN
E5
             1
Ε6
             1
                  OCTADECYLTHIOSEMICARBAZIDE/CN
Ε7
             1
                  OCTADECYLTHIOUREA/CN
             1
                  OCTADECYLTITANIUM TRICHLORIDE/CN
E8
             1
                  OCTADECYLTOLUENESULFONIC ACID ETHANOLAMINE SALT/CN
Ε9
                  OCTADECYLTRI-DECYLSILANE/CN
E10
             1
E11
             1
                  OCTADECYLTRIACETOXYSILANE/CN
E12
             1
                  OCTADECYLTRIBUTYLAMMONIUM/CN
=> e cetyl thioglycolate/cn
            1
                  CETYL SULFONE/CN
F.1
E.2
             1
                  CETYL TETRAETHYLPHOSPHORODIAMIDITE/CN
Е3
             1 --> CETYL THIOGLYCOLATE/CN
```

```
1 CETYL TITANATE/CN
E.4
E.5
            1
                 CETYL TOSYLATE/CN
E6
            1
                 CETYL TRIFLATE/CN
E.7
            1
                 CETYL VINYL ETHER/CN
E.8
                  CETYL VINYL ETHER POLYMER/CN
            1
                  CETYL VINYL ETHER-2,2,4-TRIMETHYL-1,3-PENTANEDIOL 1-ISOBUTYR
E9
             1
                  ATE 3-METHACRYLATE-VINYL CHLORIDE COPOLYMER/CN
E10
             1
                  CETYL VINYL ETHER-2,2,4-TRIMETHYL-1,3-PENTANEDIOL 3-ACRYLATE
                   1-ISOBUTYRATE-VINYL CHLORIDE COPOLYMER/CN
             1
                   CETYL VINYL ETHER-DODECYL VINYL ETHER-MYRISTYL VINYL ETHER C
E11
                   OPOLYMER/CN
E12
                   CETYL VINYL ETHER-ETHYLENE-METHYL METHACRYLATE-VINYL CHLORID
             1
                   E COPOLYMER/CN
=> s e3;d
             1 "CETYL THIOGLYCOLATE"/CN
T.9
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ANSWER 1 OF 1 REGISTRY COPYRIGHT 2009 ACS on STN 1.9 RN 22811-02-5 REGISTRY EDEntered STN: 16 Nov 1984 Acetic acid, 2-mercapto-, hexadecyl ester (CA INDEX NAME) OTHER CA INDEX NAMES: Acetic acid, mercapto-, hexadecyl ester (8CI, 9CI) OTHER NAMES: CN Cetyl thioglycolate CN Hexadecyl mercaptoacetate MF C18 H36 O2 S LC STN Files: BEILSTEIN*, BIOSIS, CA, CAPLUS, CHEMLIST, CSCHEM, GMELIN*, MEDLINE, TOXCENTER, USPATFULL (*File contains numerically searchable property data) EINECS** Other Sources:

(**Enter CHEMLIST File for up-to-date regulatory information)

PROPERTY DATA AVAILABLE IN THE 'PROP' FORMAT

6 REFERENCES IN FILE CA (1907 TO DATE)

1 REFERENCES TO NON-SPECIFIC DERIVATIVES IN FILE CA

6 REFERENCES IN FILE CAPLUS (1907 TO DATE)

```
=> file plus
'PLUS' IS NOT A VALID FILE NAME
SESSION CONTINUES IN FILE 'REGISTRY'
Enter "HELP FILE NAMES" at an arrow prompt (=>) for a list of files
that are available. If you have requested multiple files, you can
specify a corrected file name or you can enter "IGNORE" to continue
accessing the remaining file names entered.

=> s 106-94-5 and (2885-00-9 or 2917-26-2 or 22811-02-5 or 10220-46-9)
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1 106-94-5 (106-94-5/RN) 1 2885-00-9 (2885-00-9/RN)

1 2917-26-2 (2917-26-2/RN) 1 22811-02-5 (22811-02-5/RN)1 10220-46-9 (10220-46-9/RN)0 106-94-5 AND (2885-00-9 OR 2917-26-2 OR 22811-02-5 OR 10220-46-9 L10

=> file caplus COST IN U.S. DOLLARS SINCE FILE TOTAL ENTRY SESSION FULL ESTIMATED COST 77.96 87.74 DISCOUNT AMOUNTS (FOR QUALIFYING ACCOUNTS) SINCE FILE TOTAL ENTRY SESSION CA SUBSCRIBER PRICE 0.00 -0.82

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CAplus now includes complete International Patent Classification (IPC) reclassification data for the third quarter of 2008.

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This file contains CAS Registry Numbers for easy and accurate

 \Rightarrow s 106-94-5 and (2885-00-9 or 2917-26-2 or 22811-02-5 or 10220-46-9)REG1stRY INITIATED

Substance data SEARCH and crossover from CAS REGISTRY in progress... Use DISPLAY HITSTR (or FHITSTR) to directly view retrieved structures.

REG1stRY INITIATED

Substance data SEARCH and crossover from CAS REGISTRY in progress... Use DISPLAY HITSTR (or FHITSTR) to directly view retrieved structures.

L14 50 L13

REG1stRY INITIATED

Substance data SEARCH and crossover from CAS REGISTRY in progress... Use DISPLAY HITSTR (or FHITSTR) to directly view retrieved structures.

L16 6 L15

REG1stRY INITIATED

Substance data SEARCH and crossover from CAS REGISTRY in progress... Use DISPLAY HITSTR (or FHITSTR) to directly view retrieved structures.

L18 1621 L17

REG1stRY INITIATED

Substance data SEARCH and crossover from CAS REGISTRY in progress... Use DISPLAY HITSTR (or FHITSTR) to directly view retrieved structures.

L20 2011 L19

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=> d
L22 ANSWER 1 OF 1 CAPLUS COPYRIGHT 2009 ACS on STN
    2004:847649 CAPLUS
DN
    141:353637
    Pretreatment of Aq-alloy surface with organosulfur compounds for
ΤI
    tarnishing prevention
    Johns, Peter Gammon; Harrison, Clare Elizabeth
IN
    Middlesex Silver Co. Limited, UK
PA
SO
    PCT Int. Appl., 43 pp.
    CODEN: PIXXD2
DT
    Patent
    English
LA
FAN.CNT 1
                                       APPLICATION NO.
                      KIND
    PATENT NO.
                             DATE
                                                             DATE
                      ____
                                        _____
                             20041014 WO 2004-GB1373
    WO 2004087996
                       A1
                                                             20040330
PΙ
        W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BW, BY, BZ, CA, CH,
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W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BW, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, EG, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NA, NI, NO, NZ, OM, PG, PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, SY, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, YU, ZA, ZM, ZW RW: BW, GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZM, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM, AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IT, LU, MC, NL, PL, PT, RO, SE, SI, SK, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GO, GW, ML, MR, NE, SN, SK, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG AU 2004225693 20041014 AU 2004-225693 Α1 20040330 CA 2520807 20041014 CA 2004-2520807 Α1 20040330 EP 2004-724313 EP 1611267 20060104 20040330 Α1 AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT, IE, SI, LT, LV, FI, RO, MK, CY, AL, TR, BG, CZ, EE, HU, PL, SK CN 1780937 Α 20060531 CN 2004-80011375 20040330 Τ JP 2006523266 20061012 JP 2006-506057 20040330 IN 2005DN04346 20070831 IN 2005-DN4346 20050926 Α MX 2005010452 20060510 MX 2005-10452 20050928 Α US 20070039665 20070222 US 2005-551476 A1 20050929 PRAI GB 2003-7290 Α 20030331 WO 2004-GB1373 W 20040330

RE.CNT 9 THERE ARE 9 CITED REFERENCES AVAILABLE FOR THIS RECORD ALL CITATIONS AVAILABLE IN THE RE FORMAT

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10 L21 NOT L22
L23
=> d 1-10 all
L23 ANSWER 1 OF 10 CAPLUS COPYRIGHT 2009 ACS on STN
     1993:472255 CAPLUS
     119:72255
DN
OREF 119:13013a,13016a
     Entered STN: 21 Aug 1993
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ΤI Preparation of alkanamidoammonium compounds as hair growers

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PΑ Taisho Pharmaceutical Co., Ltd., Japan

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DТ
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LA
    Japanese
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    ICM C07C235-10
    ICS A61K007-06; C07C317-44; C07C323-60
    23-18 (Aliphatic Compounds)
CC
    Section cross-reference(s): 62
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                      KIND DATE
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    WO 9303005
                       A1 19930218 WO 1992-JP1014
                                                              19920807
PΙ
        W: AU, CA, JP, KR, US
        RW: AT, BE, CH, DE, DK, ES, FR, GB, GR, IE, IT, LU, MC, NL, SE
                A 19930223 JP 1991-287374 19910810
    JP 05043529
    AU 9224027
                       Α
                             19930302
                                        AU 1992-24027
                                                              19920807
                      В2
    AU 656625
                            19950209
    CN 1082534
                       A
                             19940223
                                        CN 1993-101376
                                                              19930210
                       A
PRAI JP 1991-287374
WO 1992-JP1014
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                             19920807
CLASS
PATENT NO. CLASS PATENT FAMILY CLASSIFICATION CODES
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 WO 9303005
               ICM
                      C07C235-10
                      A61K007-06; C07C317-44; C07C323-60
                ICS
                IPCI
                      C07C0235-10 [ICM,5]; C07C0235-00 [ICM,5,C*];
                      A61K0007-06 [ICS,5]; C07C0317-44 [ICS,5]; C07C0317-00
                      [ICS,5,C*]; C07C0323-60 [ICS,5]; C07C0323-00 [ICS,5,C*]
                      A61K0008-00 [I,C*]; A61K0008-00 [I,A]; A61K0008-30
                IPCR
                      [I,C*]; A61K0008-40 [I,A]; A61K0008-46 [I,A];
                      A61Q0005-00 [I,C*]; A61Q0005-00 [I,A]; A61Q0007-00
                      [I,C*]; A61Q0007-00 [I,A]; C07C0235-00 [I,C*];
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                      C07C235/10; C07C317/44; C07C323/60; M07C
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                IPCR
                      [I,C*]; A61K0008-40 [I,A]; A61K0008-46 [I,A];
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                      [I,C*]; A61Q0007-00 [I,A]; C07C0235-00 [I,C*];
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                      C07C0317-44 [I,A]; C07C0323-00 [I,C*]; C07C0323-52
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                      C07C235/10; C07C317/44; C07C323/60; M07C
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 CN 1082534
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                      [ICS, 5, C*]; C07C0235-06 [ICS, 5]; C07C0235-26 [ICS, 5];
                      C07C0235-00 [ICS,5,C*]
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[I,C*]; A61K0008-40 [I,A]; A61K0008-46 [I,A];
                       A61Q0005-00 [I,C*]; A61Q0005-00 [I,A]; A61Q0007-00
                       [I,C*]; A61Q0007-00 [I,A]; C07C0235-00 [I,C*];
                       C07C0235-06 [I,A]; C07C0235-10 [I,A]; C07C0235-16
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                       C07C0317-44 [I,A]; C07C0323-00 [I,C*]; C07C0323-52
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                ECLA
                       C07C235/10; C07C317/44; C07C323/60; M07C
OS
    MARPAT 119:72255
AΒ
    [R1A-R2-CONR3-R4-N+R5R6R7].1/mXm-(R1 = C1-22 alkyl, C3-8 cycloalkyl; R2 =
    C1-15 alkylene; R3 = H, C1-5 alkyl; R4 = C2-10 alkylene; R5, R6 = C1-5
    alkyl; R7 = C1-22 alkyl, C2-10 alkenyl, phenyl-C1-5 alkyl; A = O, S(O)n; n
    = 0-2; Xm- = anion where m = integer corresponding to the anion charge
    number) are prepared Thus, thiolation of 11-bromoundecanoic acid with dodecyl
    mercaptan in the presence of NaOH in EtOH at 60^{\circ} and amidation of
    the resulting 11-(dodecylthio)undecanoic acid with
    N, N-dimethyl-1, 3-diaminopropane in refluxing xylene with removal of H2O
    gave N-[3-(dimethylamino)propyl]-11-(dodecylthio)undecanamide which was
    methylated by MeI in EtOH at room temperature for 3 days to give
    N-[3-[11-(dodecylthio)undecanamido]propyl]-N,N,N-trimethylammonium iodide.
    When 2% solution of N-benzyl-N-[3-[11-(decylthio)undecanamido]propyl]-N,N-
    dimethylammonium iodide in EtOH at 0.2 Ml/day for 10 days was applied to a
    mice in the back from which hairs were clipped, the hair-removed part
    showed nearly 100% hair growth vs. 0% for the control animal. A total of
    41 I were prepared
ST
    alkanamidoammonium prepn hair grower
ΙT
    Hair preparations
        (growth stimulants, alkanamidoammonium compds.)
ΙT
    124-38-9, Carbon dioxide, reactions
    RL: RCT (Reactant); RACT (Reactant or reagent)
        (dry ice, reaction of, in preparation of alkanamidoammonium hair grower)
    148414-23-7P 148414-24-8P
                                 148414-25-9P 148414-26-0P 148414-27-1P
ΤТ
    148414-28-2P
                  148414-29-3P
                                 148414-30-6P 148414-31-7P 148414-32-8P
    148414-33-9P 148414-34-0P 148414-35-1P 148414-36-2P 148414-37-3P
    148414-38-4P 148414-39-5P 148414-40-8P 148414-41-9P 148414-42-0P
    148414-43-1P 148414-44-2P 148414-45-3P 148414-46-4P 148414-47-5P
    148414 - 48 - 6P \qquad 148414 - 49 - 7P \qquad 148414 - 50 - 0P \qquad 148414 - 51 - 1P \qquad 148414 - 52 - 2P
    148414-53-3P 148414-54-4P 148754-16-9P 148754-17-0P
                                                               148754-18-1P
    148754-19-2P
                  148754-20-5P 148754-21-6P 148754-22-7P
                                                               148754-23-8P
    148781-06-0P
    RL: SPN (Synthetic preparation); PREP (Preparation)
        (preparation of, as hair grower)
ΙT
    1462-53-9P, 3-(Tetradecylthio)propionic acid
                                                   3062-66-6P,
                                      5454-93-3P, 11-(Dodecylthio)undecanoic
    3-(Octadecylthio)propionic acid
           7031-23-4P, 3-(Methylthio) propionyl chloride
    acid
                                                          83518-31-4P,
                                 103808-51-1P, (Octadecylthio) acetic acid
    1-(4-Bromobutoxy)tridecane
    103808-53-3P, 11-(Decylthio)undecanoic acid
                                                  148429-60-1P,
    11-(Isopentylthio)undecanoic acid
                                        148429-61-2P,
    11-(Cyclohexylthio) undecanoic acid
                                        148429-62-3P,
                                    148429-63-4P, 5-(Hexadecylthio)valeric
    5-(Tetradecylthio) valeric acid
            148429-64-5P, 5-(Octadecylthio)valeric acid 148429-65-6P
    148429-66-7P, 2-(Octadecylthio) propionic acid
                                                    148429-67-8P,
     3-(Tetradecylsulfonyl)propionic acid
                                          148429-68-9P,
    3-(Octadecylsulfonyl)propionic acid
                                          148429-69-0P, 5-(Tridecyloxy) valeric
    acid
            148429-70-3P, 5-(Tetradecyloxy)valeric acid
                                                        148429-71-4P,
     5-(Hexadecyloxy)pentanonitrile 148429-72-5P,
    5-(Octadecyloxy)pentanonitrile
                                     148429-73-6P
                                                    148429-74-7P
    148429-75-8P
                   148429-76-9P 148429-77-0P
                                                148429-78-1P 148429-79-2P
                                 148429-82-7P
    148429-80-5P
                  148429-81-6P
                                                 148429-83-8P
                                                                148429-84-9P
    148429-85-0P 148429-86-1P 148429-87-2P
                                                 148429-88-3P
                                                                148429-89-4P
    148429-90-7P 148429-91-8P 148429-92-9P 148429-93-0P
                                                                148429-94-1P
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A61K0008-00 [I,C*]; A61K0008-00 [I,A]; A61K0008-30

TPCR

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148429-95-2P
                  148429-96-3P 148429-97-4P
                                                 148429-98-5P 148429-99-6P
    RL: SPN (Synthetic preparation); PREP (Preparation)
        (preparation of, as intermediate for alkanamidoammonium hair grower)
    74-83-9, Methyl bromide, reactions 74-88-4, Methyl iodide, reactions
ΙT
    77-78-1 79-08-3, Bromoacetic acid 80-48-8, Methyl p-toluenesulfonate
    80-58-0, 2-Bromobutyric acid 100-36-7, N,N-Diethyl-1,2-ethylenediamine
    100-44-7, Benzyl chloride, reactions 104-78-9,
    N, N-Diethyl-1, 3-diaminopropane 106-94-5, Propyl bromide
    106-95-6, Allyl bromide, reactions 108-00-9,
    N, N-Dimethyl-1, 2-ethylenediamine 109-55-7,
    N, N-Dimethyl-1, 3-diaminopropane 110-52-1, 1,4-Dibromobutane 112-55-0,
    Dodecyl mercaptan
                       112-70-9, 1-Tridecanol 112-72-1, 1-Tetradecanol
    112-92-5, 1-Octadecanol 512-56-1, Trimethyl phosphate
    3-Bromopropionic acid 598-72-1, 2-Bromopropionic acid
                                                              2067-33-6,
    5-Bromovaleric acid 2834-05-1, 11-Bromoundecanoic acid 2885-00-9
     , Octadecyl mercaptan 5414-21-1 36653-82-4, 1-Hexadecanol 53369-71-4
    RL: RCT (Reactant); RACT (Reactant or reagent)
        (reaction of, in preparation of alkanamidoammonium hair grower)
RE.CNT 2
             THERE ARE 2 CITED REFERENCES AVAILABLE FOR THIS RECORD
(1) Anon; JP 50019719 A CAPLUS
(2) Anon; JP 54130509 A CAPLUS
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    1992:263476 CAPLUS
AN
    116:263476
OREF 116:44527a,44530a
    Entered STN: 27 Jun 1992
ED
ΤI
    Liquid heat capacity for 300 organics
ΑU
    Yaws, Carl L.; Pan, Xiang
CS
    Lamar Univ., Beaumont, TX, 77710, USA
    Chemical Engineering (New York, NY, United States) (1992), 99(4), 130-5
SO
    CODEN: CHEEA3; ISSN: 0009-2460
DT
    Journal
    English
LA
CC
    69-2 (Thermodynamics, Thermochemistry, and Thermal Properties)
    Section cross-reference(s): 22
    A correlation was developed for the calcn. of the heat capacities of liquid
AΒ
    organic compds. as functions of temperature (T). Values were calculated for
300
    compds. and the coeffs. of the correlation equation, Cp = A + BT + CT2,
    are listed, with T in °K and the temperature interval for which the
    equation is valid for each compound
ST
    heat capacity org compd correlation equation
ΙT
    Organic compounds, properties
    RL: PRP (Properties)
        (heat capacities of liquid, correlation equation for)
ΙT
    Heat capacity
        (of organic compds. in liquid state, correlation equations for)
    50-00-0, Formaldehyde, properties 56-23-5, Carbon tetrachloride,
ΙT
    properties 60-29-7, Ethyl ether, properties 62-53-3, Aniline,
                 64-17-5, Ethyl alcohol, properties 64-18-6, Formic acid,
    properties
                 64-19-7, Acetic acid, properties 65-85-0, Benzoic acid,
    properties
                 66-25-1, Hexanal
                                   67-56-1, Methanol, properties
                                                                   67-63-0,
    properties
                                   67-64-1, Acetone, properties
                                                                   67 - 66 - 3,
    Isopropyl alcohol, properties
    Chloroform, properties 71-23-8, Propyl alcohol, properties 71-36-3,
    Butyl alcohol, properties 71-41-0, Pentyl alcohol, properties
                                                                      74-83-9,
    Bromomethane, properties 74-87-3, Chloromethane, properties 74-88-4,
                              74-89-5, Methylamine, properties
                                                                74-93-1,
    Iodomethane, properties
    Methanethiol, properties 74-96-4, Bromoethane 75-00-3, Chloroethane
    75-01-4, Chloroethene, properties 75-02-5, Fluoroethene
               75-04-7, Ethylamine, properties 75-05-8, Acetonitrile,
    Iodoethane
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properties 75-07-0, Acetaldehyde, properties 75-08-1, Ethanethiol
75-09-2, Dichloromethane, properties 75-10-5, Difluoromethane 75-11-6,
Diiodomethane 75-15-0, Carbon disulfide, properties 75-18-3, Methyl
sulfide 75-21-8, Ethylene oxide, properties 75-26-3, 2-Bromopropane
75-29-6, 2-Chloropropane 75-30-9, 2-Iodopropane 75-33-2,
                 75-34-3, 1,1-Dichloroethane 75-35-4,
2-Propanethiol
1,1-Dichloroethene, properties 75-36-5, Acetyl chloride 75-37-6,
1,1-Difluoroethane 75-38-7, 1,1-Difluoroethene 75-43-4,
Dichlorofluoromethane 75-45-6, Chlorodifluoromethane 75-46-7,
Trifluoromethane 75-47-8, Triiodomethane 75-50-3, Trimethylamine,
properties 75-52-5, Nitromethane, properties 75-56-9, Propylene oxide,
properties 75-64-9, tert-Butylamine, properties 75-65-0, tert-Butyl
alcohol, properties 75-66-1, 2-Methyl-2-propanethiol 75-69-4,
Trichlorofluoromethane 75-71-8, Dichlorodifluoromethane 75-72-9,
Chlorotrifluoromethane 75-73-0, Carbon tetrafluoride 75-85-4,
tert-Pentyl alcohol 76-01-7, Pentachloroethane 76-13-1,
1,1,2-Trichlorotrifluoroethane 76-14-2, 1,2-Dichlorotetrafluoroethane
76-16-4, Hexafluoroethane 78-75-1, 1,2-Dibromopropane 78-76-2,
2-Bromobutane 78-82-0, Isobutyronitrile 78-86-4, 2-Chlorobutane
78-87-5, 1,2-Dichloropropane 78-92-2, sec-Butyl alcohol 78-93-3,
2-Butanone, properties 79-00-5, 1,1,2-Trichloroethane 79-01-6,
Trichloroethene, properties 79-10-7, Acrylic acid, properties
1,1,2,2-Tetrachloroethane 95-48-7, o-Cresol, properties 95-50-1,
o-Dichlorobenzene 96-18-4, 1,2,3-Trichloropropane 98-08-8
p-Cresol, properties 106-46-7, p-Dichlorobenzene
                                                     106-93-4,
1,2-Dibromoethane 106-94-5, 1-Bromopropane 106-95-6,
3-Bromo-1-propene, properties 107-03-9, 1-Propanethiol
                                                            107-05-1,
3-Chloro-1-propene 107-06-2, 1,2-Dichloroethane, properties
1-Iodopropane 107-10-8, Propylamine, properties 107-12-0,
Propionitrile 107-13-1, Acrylonitrile, properties 107-18-6, Allyl
alcohol, properties 107-21-1, Ethylene glycol, properties 107-84-6,
1-Chloro-3-methylbutane 107-87-9, 2-Pentanone 108-20-3, Isopropyl
ether 108-24-7, Acetic anhydride 108-39-4, m-Cresol, properties 108-86-1, Bromobenzene, properties 108-90-7, Chlorobenzene, properties
108-95-2, Phenol, properties 108-98-5, Benzenethiol, properties
108-99-6, 3-Picoline 109-06-8, 2-Picoline 109-65-9, 1-Bromobutane
109-69-3, 1-Chlorobutane 109-73-9, Butylamine, properties
Butyronitrile 109-79-5, 1-Butanethiol 109-89-7, Diethylamine,
           110-00-9, Furan
                              110-01-0, Thiacyclopentane 110-02-1,
properties
           110-53-2, 1-Bromopentane 110-62-3, Valeraldehyde 110-66-7,
Thiophene
               110-81-6, Ethyl disulfide 110-86-1, Pyridine,
1-Pentanethiol
properties 111-27-3, Hexyl alcohol, properties 111-31-9, 1-Hexanethiol
111-43-3, Propyl ether 111-47-7, Propyl sulfide 111-70-6, Heptyl
                              111-87-5, Octyl alcohol, properties
alcohol 111-71-7, Heptanal
111-88-6, 1-Octanethiol 112-30-1, Decyl-alcohol 112-31-2, Decanal
112-42-5, Undecyl alcohol 112-51-6, Pentyl disulfide
                                                        112-53-8, Dodecyl
alcohol 112-55-0, 1-Dodecanethiol 112-70-9, 1-Tridecanol 112-72-1,
1-Tetradecanol 112-92-5, 1-Octadecanol 115-10-6, Methyl ether
115-25-3, Octafluorocyclobutane 121-44-8, Triethylamine, properties
123-38-6, Propional dehyde, properties 123-72-8, Butyral dehyde
123-75-1, Pyrrolidine, properties 124-13-0, Octanal 124-19-6, Nonanal
124-38-9, Carbon dioxide, properties 124-40-3, Dimethylamine, properties 127-18-4, Tetrachloroethene, properties 141-78-6, Ethyl acetate,
            142-28-9, 1,3-Dichloropropane 142-96-1, Butyl ether
properties
143-08-8, Nonyl alcohol 143-10-2, 1-Decanethiol 151-56-4, Ethylenimine, properties 156-59-2, cis-1,2-Dichloroethene 156-60-5, trans-1,2-Dichloroethene 287-27-4, Thiacyclobutane 352-32-9,
p-Fluorotoluene 352-93-2, Ethylsulfide 353-36-6, Fluoroethane
359-11-5, Trifluoroethene 367-11-3, o-Difluorobenzene 372-18-9,
m-Difluorobenzene 420-12-2, Thiacyclopropane 420-26-8, 2-Fluoropropane
420-46-2, 1,1,1-Trifluoroethane 460-13-9, 1-Fluoropropane 462-06-6,
Fluorobenzene 463-58-1, Carbonyl sulfide 507-19-7,
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    Butyl tridecyl sulfide
                           66826-84-4, Propyl undecyl sulfide
    RL: PRP (Properties)
       (heat capacity of, correlation equation for)
L23 ANSWER 3 OF 10 CAPLUS COPYRIGHT 2009 ACS on STN
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DN
    115:263973
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    Entered STN: 14 Dec 1991
    633 organic chemicals: surface tension data
ΤI
ΑU
    Yaws, Carl L.; Yang, Haur Chung; Pan, Xiang
CS
    Lamar Univ., Beaumont, TX, USA
    Chemical Engineering (New York, NY, United States) (1991), 98(3), 140-2,
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    144, 146, 148, 150
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    Journal
    English
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    66-1 (Surface Chemistry and Colloids)
AΒ
    Surface tension data are tabulated for 633 organic chems., including
    hydrocarbons, alcs., and acids. Exptl. data from the literature were used
    to calculate estimated values for chems. for which no exptl. data exist.
Surface
    tensions at any temperature can be calculated using the Othmer relation, which
is
    discussed.
ST
    surface tension org chem
ΙT
    Surface tension
       (of hydrocarbons and sulfides)
ΤТ
    Organic compounds, properties
    RL: PRP (Properties)
       (surface tension data for)
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     RL: PRP (Properties)
        (surface tension data for)
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2216-30-0, 2,5-Dimethylheptane 2216-32-2, 4-Ethylheptane 2216-33-3,
                2216-34-4, 4-Methyloctane 2243-98-3, 1-Undecyne
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2437-56-1, 1-Tridecene 2532-58-3, cis-1,3-Dimethylcyclopentane
2613-61-8, 2,4,6-Trimethylheptane 2690-08-6, Octyl sulfide
                     2851-83-4, Dodecyl ethyl sulfide 2882-98-6,
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                     2883-02-5, 1-Cyclohexylnonane 2885-00-9,
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                     3074-75-7, 4-Ethyl-2-methylhexane 3074-76-8,
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                     3522-94-9, 2,2,5-Trimethylhexane 3698-89-3,
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Ethyl octyl sulfide 3698-95-1, Methyl octyl sulfide 3741-00-2,
                      3877-15-4, Methyl propyl sulfide 4032-86-4,
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                                                           4032-93-3,
2,3,6-Trimethylheptane 4032-94-4, 2,4-Dimethyloctane 4050-45-7,
trans-2-Hexene 4110-44-5, 3,3-Dimethyloctane 4110-50-3, Ethyl propyl
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4753-80-4, Thiacycloheptane 5171-84-6, 3,3,4,4-Tetramethylhexane 5332-52-5, 1-Undecanethiol 5617-41-4, 1-Cyclohexylheptane 5617-42-5,
1-Cyclopentylheptane
                      5634-30-0, 1-Cyclopentyldodecane
RL: PRP (Properties)
   (surface tension data for)
5881-17-4, 3-Ethyloctane
                           5911-04-6, 3-Methylnonane 6006-33-3,
1-Cyclohexyltridecane 6006-34-4, 1-Cyclopentyltridecane 6006-95-7,
1-Cyclohexylpentadecane 6294-31-1, Hexyl sulfide 6742-54-7,
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1-Cyclopentylundecane 6812-38-0, 1-Cyclohexylhexadecane
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                          6876-18-2, 3-Isopropyl-2-methylhexane
1-Cyclopentylhexadecane
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7154-79-2, 2,2,3,3-Tetramethylpentane 7154-80-5, 3,3,5-Trimethylheptane
7220-26-0, 3-Ethyl-2,4-dimethylhexane 7289-44-3, Methyl undecyl sulfide
7289-45-4, Methyl tetradecyl sulfide 7309-44-6, Ethyl hexyl sulfide
7372-86-3, 2-Ethyl-6-methylnaphthalene 7642-09-3, cis-3-Hexene
7688-21-3, cis-2-Hexene 10496-15-8, Hexyl disulfide 10496-16-9, Heptyl
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2,2-Dimethyloctane
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3,4-Dimethyloctane
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2,4,4-Trimethylhexane
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16747-38-9, 2,3,3,4-Tetramethylpentane 16747-42-5, 2,2,4,5-Tetramethylhexane 16747-44-7, 2,2,3,3,4-Pentamethylpentane
16747-45-8, 2,2,3,4,4-Pentamethylpentane 16789-46-1,
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3-Ethyl-2-methylhexane
                        16967-04-7, Butyl hexyl sulfide 17059-55-1,
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                                                           17302-04-4.
4-Ethyl-4-methylheptane 18435-45-5, 1-Nonadecene 18437-89-3, Butyl
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               19398-77-7, 3,4-Diethylhexane 19484-26-5,
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2,3,5-Trimethylheptane 20278-87-9, 3,3,4-Trimethylheptane
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3,4,4-Trimethylheptane 20278-89-1, 3,4,5-Trimethylheptane
                                                                20291-60-5,
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                                                           20291-91-2,
3-Ethyl-2,2-dimethylhexane 20291-95-6, 2,2,5-Trimethylheptane
22438-39-7, Decyl methyl sulfide 24768-42-1, Butyl pentyl sulfide
24768-43-2, Hexyl propyl sulfide 24768-44-3, Ethyl heptyl sulfide
24768-46-5, Heptyl propyl sulfide 25276-70-4, 1-Pentadecanethiol
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                                                                26186-01-6,
1-Nonadecyne
               26186-02-7, 1-Tridecyne 27563-68-4, Hexadecyl methyl
         31032-94-7, 2-Ethyl-3-methylnaphthalene
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1,2,3,5-Tetraethylbenzene
                            40289-98-3, Methyl octadecyl sulfide
40813-84-1, Butyl heptyl sulfide 41947-84-6, Ethyl octadecyl sulfide
42205-08-3, 1,2,3-Triethylbenzene
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51750-65-3, 2,2,4,4-Tetramethylhexane 52896-87-4, 4-Isopropylheptane
52896-88-5, 4-Ethyl-2-methylheptane 52896-89-6, 4-Ethyl-3-methylheptane
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52896-95-4, 2,3,4-Trimethylheptane 52896-99-8,
4-Ethyl-2,2-dimethylhexane 52897-00-4, 3-Ethyl-2,3-dimethylhexane
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52897-09-3, 2,2,3,5-Tetramethylhexane 52897-10-6, 2,3,3,4-Tetramethylhexane 52897-11-7, 2,3,3,5-Tetramethylhexane
52897-12-8, 2,3,4,4-Tetramethylhexane 52897-15-1, 2,3,4,5-Tetramethylhexane 52897-16-2, 3,3-Diethyl-2-methylpentane
52897-17-3, 3-Ethyl-2,2,3-trimethylpentane 52897-18-4,
3-Ethyl-2,2,4-trimethylpentane 52897-19-5,
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     53193-23-0, 1-Nonadecanethiol 54105-66-7, 1-Cyclohexylundecane
     59973-07-8, Methyl nonyl sulfide 59973-08-9, Ethyl nonyl sulfide
     62103-66-6, Nonyl propyl sulfide 62155-09-3, Methyl tridecyl sulfide
     62155-10-6, Methyl pentadecyl sulfide 62155-11-7, Heptadecyl methyl
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                                                                     66271-82-7,
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     Butyl tridecyl sulfide 66826-84-4, Propyl undecyl sulfide
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        (surface tension data for)
L23 ANSWER 4 OF 10 CAPLUS COPYRIGHT 2009 ACS on STN
     1990:485738 CAPLUS
     113:85738
OREF 113:14355a,14358a
    Entered STN: 01 Sep 1990
     Predict enthalpy of vaporization
    Yaws, C. L.; Yang, H. C.; Cawley, W. A.
    Lamar Univ., Beaumont, TX, USA
     Hydrocarbon Processing, International Edition (1990), 69(6), 87-90
    CODEN: IHPRBS; ISSN: 0018-8190
     Journal
     English
     69-2 (Thermodynamics, Thermochemistry, and Thermal Properties)
     The enthapy of vaporization at a given temperature is related to other
     properties for the major organic compds.
     enthalpy vaporization org compd
     Heat of evaporation and Heat of condensation
        (calcn. of, of organic compds.)
     Organic compounds, properties
     RL: PRP (Properties)
        (heats of evaporation of, calcn. of)
                                        56-23-5, Carbon tetrachloride,
     50-00-0, Formaldehyde, properties
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                 64-17-5, Ethanol, properties 64-18-6, Formic acid,
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                64-19-7, Acetic acid, properties 65-85-0, Benzoic acid,
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     properties
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                  67-72-1, Hexachloroethane 71-23-8, Propyl alcohol,
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                  71-36-3, 1-Butanol, properties 71-41-0, Pentyl alcohol,
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                71-43-2, Benzene, properties 74-82-8, Methane, properties
     properties
     74-83-9, properties 74-84-0, Ethane, properties 74-85-1, Ethene,
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                  74-89-5, Methanamine, properties 74-93-1, Methanethiol,
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     properties 74-96-4, Bromoethane 74-98-6, Propane, properties 74-99-7, 1-Propyne 75-00-3, Chloroethane 75-01-4, properties 75-02-5, Fluoroethene 75-03-6, Iodoethane 75-04-7, Ethanamine,
                  75-05-8, Acetonitrile, properties 75-07-0, Acetaldehyde,
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PROC (Process) (b.p. and critical temperature and f.p. and heat of evaporation of) ΙT 292-64-8, Cyclooctane 352-32-9, p-Fluorotoluene 352-93-2, Ethylsulfide 353-36-6, Fluoroethane 359-11-5, Trifluoroethene 367-11-3, o-Difluorobenzene 372-18-9, m-Difluorobenzene 392-56-3, 420-12-2, Thiirane 420-26-8, 2-Fluoropropane Hexafluorobenzene 420-46-2, 1,1,1-Trifluoroethane 460-12-8, 1,3-Butadiyne 460-13-9, 1-Fluoropropane 460-19-5, Cyanogen 462-06-6, Fluorobenzene 463-49-0, 1,2-Propadiene 463-51-4, Ketene 463-58-1, Carbonyl sulfide 463-82-1, 2,2-Dimethylpropane 464-06-2, 2,2,3-Trimethylbutane 488-23-3, 1,2,3,4-Tetramethylbenzene 493-01-6 493-02-7 503-17-3, 2-Butyne 507-09-5, Thioacetic acid, properties 507-19-7, 2-Bromo-2-methylpropane 507-20-0 513-35-9 513-36-0, 1-Chloro-2-methylpropane 513-44-0, 2-Methyl-1-propanethiol 513-53-1, 2-Butanethiol 526-73-8, 1,2,3-Trimethylbenzene 527-53-7, 1,2,3,5-Tetramethylbenzene 533-98-2. 1,2-Dibromobutane 536-74-3, Ethynylbenzene 538-68-1, Pentylbenzene 540-36-3, p-Difluorobenzene 540-54-5 540-67-0, Ethyl methyl ether 540-84-1, 2,2,4-Trimethylpentane 541-73-1, m-Dichlorobenzene 543-59-9, 1-Chloropentane 544-25-2, 1,3,5-Cycloheptatriene 544-40-1, Butyl sulfide 544-76-3, Hexadecane 554-14-3, 2-Methylthiophene 556-56-9, 3-Iodo-1-propene 557-17-5, Methyl propyl ether 558-17-8, 2-Iodo-2-methylpropane 558-37-2 560-21-4, 2,3,3-Trimethylpentane 562-49-2, 3,3-Dimethylpentane 563-16-6, 3,3-Dimethylhexane 563-45-1, 563-79-1 3-Methyl-1-butene 563-46-2 563-78-0, 2,3-Dimethyl-1-butene 564-02-3, 2,2,3-Trimethylpentane 565-59-3, 2,3-Dimethylpentane 565-75-3, 2,3,4-Trimethylpentane 571-58-4, 1,4-Dimethylnaphthalene 571-61-9, 1,5-Dimethylnaphthalene 573-98-8, 1,2-Dimethylnaphthalene 575-37-1, 1,7-Dimethylnaphthalene 575-41-7, 1,3-Dimethylnaphthalene 575-43-9, 1,6-Dimethylnaphthalene 581-40-8, 2,3-Dimethylnaphthalene 581-42-0, 2,6-Dimethylnaphthalene 582-16-1, 2,7-Dimethylnaphthalene 583-48-2, 3,4-Dimethylhexane 584-94-1, 2,3-Dimethylhexane 589-34-4, 3-Methylhexane 589-43-5, 2,4-Dimethylhexane 589-53-7, 4-Methylheptane 589-81-1, 3-Methylheptane 590-18-1 590-19-2, 1,2-Butadiene 2,2-Dimethylpentane 590-66-9, 1,1-Dimethylcyclohexane 590-73-8, 591-50-4, Iodobenzene 591-76-4, 2-Methylhexane 2,2-Dimethylhexane 591-93-5, 1,4-Pentadiene 591-95-7, 1,2-Pentadiene 591-96-8, 2,3-Pentadiene 592-13-2, 2,5-Dimethylhexane 592-27-8, 2-Methylheptane 592-41-6, 1-Hexene, properties 592-76-7, 1-Heptene 593-45-3, 593-60-2, Bromoethylene 593-70-4, Octadecane 593-53-3, Fluoromethane Chlorofluoromethane 594-20-7, 2,2-Dichloropropane 594-36-5, 2-Chloro-2-methylbutane 594-51-4, 2,3-Dibromo-2-methylbutane 2,2,3,3-Tetramethylbutane 598-23-2, 3-Methyl-1-butyne 598-25-4, 3-Methyl-1,2-butadiene 598-29-8, 1,2-Diiodopropane 598-53-8, Methyl isopropyl ether 598-58-3, Methyl nitrate 600-24-8, 2-Nitrobutane 604-88-6, Hexaethylbenzene 605-01-6, Pentaethylbenzene 609-26-7, 3-Ethyl-2-methylpentane 611-14-3, o-Ethyltoluene 611-15-4 616-12-6 616-44-4, 3-Methylthiophene 617-78-7, 3-Ethylpentane 619-99-8, 3-Ethylhexane 620-14-4, m-Ethyltoluene 622-96-8, p-Ethyltoluene 622-97-9 624-29-3 624-64-6 624-73-7, 1,2-Diiodoethane 624-89-5, Ethyl methyl sulfide 624-91-9, Methyl nitrite 624-92-0, Methyl

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625-80-9, Isopropyl sulfide 627-05-4, 1-Nitrobutane 627-13-4, Propyl
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629-45-8, Butyl disulfide 629-50-5, Tridecane 629-59-4, Tetradecane
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630-08-0, Carbon monoxide, properties
                                        635-81-4,
1,2,4,5-Tetraethylbenzene 638-04-0 638-46-0, Butyl ethyl sulfide
646-04-8 674-76-0
                     689-97-4, 1-Buten-3-yne 691-37-2,
4-Methyl-1-pentene 691-38-3 693-02-7, 1-Hexyne 693-83-4, Decyl
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3-Methyl-1-pentene 760-21-4, 2-Ethyl-1-butene 763-29-1,
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                                                                   765-10-6,
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821-95-4, 1-Undecene 822-27-5, Octyl disulfide 822-35-5, Cyclobutene
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3-Ethyl-2,4-dimethylpentane 1069-53-0, 2,3,5-Trimethylhexane
1070-87-7, 2,2,4,4-Tetramethylpentane 1071-26-7, 2,2-Dimethylheptane 1071-81-4, 2,2,5,5-Tetramethylhexane 1072-05-5, 2,6-Dimethylheptane
1072-16-8, 2,7-Dimethyloctane 1077-16-3, Hexylbenzene 1078-71-3,
1-Phenylheptane 1081-77-2 1120-21-4, Undecane 1120-36-1,
1-Tetradecene 1120-62-3, 3-Methylcyclopentene 1127-76-0,
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1454-85-9, 1-Heptadecanol 1455-21-6, 1-Nonanethiol 1459-09-2,
                    1459-10-5, 1-Phenyltetradecane 1551-21-9, Isopropyl
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                    1638-26-2, 1,1-Dimethylcyclopentane
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1-Heptanethiol
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                    1679-09-0, 2-Methyl-2-butanethiol 1712-64-7,
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                                                        1759-58-6
1759-81-5, 4-Methylcyclopentene 1795-15-9, 1-Cyclohexyloctane
1795-16-0, 1-Cyclohexyldecane 1795-17-1, 1-Cyclohexyldodecane
1795-18-2, 1-Cyclohexyltetradecane 1795-20-6 1795-21-7,
1-Cyclopentyldecane 1795-22-8, 1-Cyclopentyltetradecane 1795-26-2
RL: PEP (Physical, engineering or chemical process); PRP (Properties);
PROC (Process)
   (b.p. and critical temperature and f.p. and heat of evaporation of)
1795-27-3
           2004-70-8 2027-19-2, 2-Propylnaphthalene 2040-95-1,
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2,6-Dimethyloctane 2079-95-0, 1-Tetradecanethiol 2131-18-2, 1-Phenylpentadecane 2189-60-8, 1-Phenyloctane 2207-01-4 2207-03-6
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2437-56-1, 1-Tridecene 2532-58-3 2613-61-8, 2,4,6-Trimethylheptane
2690-08-6, Octyl sulfide 2765-18-6, 1-Propylnaphthalene 2851-83-4,
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2917-26-2, 1-Hexadecanethiol 3074-71-3, 2,3-Dimethylheptane

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3074-75-7, 4-Ethyl-2-methylhexane 3074-76-8, 3-Ethyl-3-methylhexane
3074-77-9, 3-Ethyl-4-methylhexane 3129-90-6, Isothiocyanic acid
3178-29-8, 4-Propylheptane 3221-61-2, 2-Methyloctane 3452-07-1,
1-Eicosene 3452-09-3, 1-Nonyne 3522-94-9, 2,2,5-Trimethylhexane
3698-89-3, Dodecyl methyl sulfide 3698-93-9, Octyl propyl sulfide
3698-94-0, Ethyl octyl sulfide 3698-95-1, Methyl octyl sulfide
3741-00-2 3877-15-4, Methyl propyl sulfide 4032-86-4,
3,3-Dimethylheptane 4032-92-2, 2,4,4-Trimethylheptane 4032-93-3,
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4292-75-5, 1-Cyclohexylhexane 4292-92-6 4457-00-5 4485-77-2, Nonyl
disulfide 4669-01-6, 1-Cyclopentylpentadecane 4753-80-4,
Thiacycloheptane 5171-84-6, 3,3,4,4-Tetramethylhexane 5332-52-5,
1-Undecanethiol 5408-86-6, 2,3-Dibromobutane 5617-41-4 5617-42-5,
1-Cyclopentylheptane 5634-30-0, 1-Cyclopentyldodecane 5881-17-4,
3-Ethyloctane 5911-04-6, 3-Methylnonane 6006-33-3,
1-Cyclohexyltridecane 6006-34-4, 1-Cyclopentyltridecane 6006-95-7,
1-Cyclohexylpentadecane 6163-66-2, tert-Butyl ether 6294-31-1, Hexyl
sulfide 6742-54-7 6765-39-5, 1-Heptadecene 6785-23-5,
1-Cyclopentylundecane 6812-38-0, 1-Cyclohexylhexadecane
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1-Cyclopentylhexadecane 6863-58-7, sec-Butyl ether 6876-18-2,
3-Isopropyl-2-methylhexane 6876-23-9 7146-60-3, 2, 3-Dimethyloctane
7154-79-2 7154-80-5, 3,3,5-Trimethylheptane 7220-26-0,
3-Ethyl-2, 4-dimethylhexane 7289-44-3, Methyl undecyl sulfide
7289-45-4, Methyl tetradecyl sulfide 7309-44-6, Ethyl hexyl sulfide
7372-86-3, 2-Ethyl-6-methylnaphthalene 7642-09-3 7688-21-3 10496-15-8, Hexyl disulfide 10496-16-9, Heptyl disulfide 10496-18-1, Decyl disulfide 13269-52-8 13360-61-7, 1-Pentadecene 13373-97-2,
1-Eicosanethiol 13475-78-0, 5-Ethyl-2-methylheptane 13475-79-1,
2,4-Dimethyl-3-isopropylpentane 13475-81-5, 2,2,3.3-Tetramethylhexane
13952-84-6, sec-Butylamine 14676-29-0, 3-Ethyl-2-methylheptane
14720-74-2, 2,2,4-Trimethylheptane 15869-80-4, 3-Ethylheptane
15869-85-9, 5-Methylnonane 15869-86-0, 4-Ethyloctane
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2,2-Dimethyloctane 15869-89-3, 2,5-Dimethyloctane 15869-92-8,
3,4-Dimethyloctane 15869-93-9, 3,5-Dimethyloctane
                                                       15869-94-0,
3,6-Dimethyloctane 15869-95-1, 4,4-Dimethyloctane 15869-96-2,
4,5-Dimethyloctane 16747-25-4, 2,2,3-Trimethylhexane 16747-26-5,
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16747-31-2, 3,3,4-Trimethylhexane 16747-32-3,
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16747-38-9, 2,3,3,4-Tetramethylpentane 16747-42-5,
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2-Ethyl-7-methylnaphthalene 17301-94-9, 4-Methylnonane
                                                           17302-01-1,
3-Ethyl-3-methylheptane 17302-02-2, 3,3-Diethylhexane
                                                           17302-04-4,
4-Ethyl-4-methylheptane 17348-59-3, Isopropyl tert-butyl ether
18435-45-5, 1-Nonadecene 18437-89-3, Butyl hexadecyl sulfide
19313-57-6, Butyl decyl sulfide 19313-61-2, Decyl ethyl sulfide
19398-77-7, 3,4-Diethylhexane 19484-26-5, 1-Tridecanethiol 20278-84-6,
2,4,5-Trimethylheptane 20278-85-7, 2,3,5-Trimethylheptane 20278-88-0, 3,4,4-Trimethylheptane
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3,4,5-Trimethylheptane 20291-60-5, Hexyl methyl sulfide 202
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20291-95-6, 2,2,5-Trimethylheptane 22438-39-7, Decyl methyl sulfide
24768-42-1, Butyl pentyl sulfide 24768-43-2, Hexyl propyl sulfide
24768-44-3, Ethyl heptyl sulfide 24768-46-5, Heptyl propyl sulfide
25276-70-4, 1-Pentadecanethiol 26158-99-6, Ethyl pentyl sulfide
26186-00-5, 1-Heptadecyne 26186-01-6, 1-Nonadecyne 26186-02-7,
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36653-82-4, 1-Hexadecanol 38842-05-6, 1,2,3,5-Tetraethylbenzene
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40289-98-3, Methyl octadecyl sulfide
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                       42205-08-3, 1,2,3-Triethylbenzene
     octadecyl sulfide
                                                            42841-80-5, Pentyl
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                    51750-65-3, 2,2,4,4-Tetramethylhexane
                                                            52896-87-4,
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                                                                52896-89-6,
     4-Ethyl-3-methylheptane 52896-90-9, 3-Ethyl-5-methylheptane
     52896-91-0, 3-Ethyl-4-methylheptane 52896-92-1, 2,2,3-Trimethylheptane
     52896-93-2, 2,3,3-Trimethylheptane 52896-95-4, 2,3,4-Trimethylheptane
     52896-99-8, 4-Ethyl-2,2-dimethylhexane 52897-00-4,
     3-Ethyl-2,3-dimethylhexane 52897-01-5, 4-Ethyl-2,3-dimethylhexane
     52897-03-7, 4-Ethyl-2, 4-dimethylhexane 52897-04-8,
     3-Ethyl-2,5-dimethylhexane 52897-05-9, 4-Ethyl-3,3-dimethylhexane
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     2,2,3,4-Tetramethylhexane 52897-09-3, 2,2,3,5-Tetramethylhexane
     52897-10-6, 2,3,3,4-Tetramethylhexane 52897-11-7,
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     52897-15-1, 2,3,4,5-Tetramethylhexane 52897-16-2
                                                        52897-17-3
     52897-18-4 52897-19-5, 3-Ethyl-2,3,4-trimethylpentane 53161-72-1,
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     nonyl sulfide 59973-08-9, Ethyl nonyl sulfide 62103-66-6, Nonyl propyl
     sulfide
             62155-09-3, Methyl tridecyl sulfide 62155-10-6, Methyl
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                                                                62155-12-8,
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     66271-54-3, Ethyl tetradecyl sulfide 66271-55-4, Propyl tridecyl sulfide
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     sulfide
             66359-41-9, Hexadecyl propyl sulfide 66359-42-0, Butyl
     pentadecyl sulfide
                         66455-35-4, Heptadecyl propyl sulfide 66577-30-8,
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     Butyl nonyl sulfide 66577-61-5, Propyl tetradecyl sulfide
                                                                   66577-62-6,
     Butyl tridecyl sulfide 66826-84-4, Propyl undecyl sulfide
     RL: PEP (Physical, engineering or chemical process); PRP (Properties);
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        (b.p. and critical temperature and f.p. and heat of evaporation of)
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    111:181192
OREF 111:30033a,30036a
     Entered STN: 10 Nov 1989
     Critical properties of chemicals
     Yaws, C. L.; Chen, D.; Yang, H. C.; Tan, L.; Nico, D.
     Lamar Univ., Beaumont, TX, USA
     Hydrocarbon Processing, International Edition (1989), 68(7), 61-4
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     English
     65-6 (General Physical Chemistry)
     The critical temps. and pressures and vols., f.p., normal b.p. critical
     compressibility and acentric factors are tabulated for 700 organic compds.
     crit property org compd
     Compression and Compressibility
        (critical factor for, of chemical substances)
     Boiling point
     Freezing point
        (of chemical substances)
     Organic compounds, properties
     RL: PRP (Properties)
        (selected values for critical consts. for)
     Critical constant
        (pressure, of chemical substances, selected value for)
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(temperature, of chemical substances, selected value for) ΙT Critical constant (volume, of chemical substances, selected value for) ΙT 50-00-0, Formaldehyde, properties 56-23-5, Carbon tetrachloride, properties 60-29-7, Ethyl ether, properties 62-53-3, Aniline, properties 64-17-5, Ethyl alcohol, properties 64-18-6, Formic acid, properties 64-19-7, Acetic acid, properties 65-85-0, Benzoic acid, properties 66-25-1, Hexanal 67-56-1, Methanol, properties 67-63-0, Isopropyl alcohol, properties 67-64-1, Acetone, properties 67-66-3, Chloroform, properties 67-72-1, Hexachloroethane 71-23-8, Propyl alcohol, properties 71-36-3, Butyl alcohol, properties 71-41-0, Pentyl alcohol, properties 71-43-2, Benzene, properties 74-82-8, Methane, properties 74-83-9, Bromomethane, properties 74-84-0, Ethane, 74-85-1, Ethylene, properties 74-86-2, Ethyne, properties properties 74-87-3, Chloromethane, properties 74-88-4, Iodomethane, properties 74-89-5, Methylamine, properties 74-93-1, Methanethiol, properties 74-96-4, Bromoethane 74-98-6, Propane, properties 74-99-7, 1-Propyne 75-00-3, Chloroethane 75-01-4, Chloroethene, properties 75-02-5, Fluoroethene 75-03-6, Iodoethane 75-04-7, Ethylamine, properties 75-05-8, Acetonitrile, properties 75-07-0, Acetaldehyde, properties 75-08-1, Ethanethiol 75-09-2, Dichloromethane, properties 75-10-5, Difluoromethane 75-11-6, Diiodomethane 75-15-0, Carbon disulfide, properties 75-18-3, Methyl sulfide 75-19-4, Cyclopropane 75-21-8, Ethylene oxide, properties 75-26-3, 2-Bromopropane 75-28-5 75-29-6, 2-Chloropropane 75-30-9, 2-Iodopropane 75-33-2, 2-Propanethiol 75-34-3, 1,1-Dichloroethane 75-35-4, 1,1-Dichloroethene, properties 75-36-5, Acetyl chloride 75-37-6, 1,1-Difluoroethane 75-38-7, 1,1-Difluoroethene 75-43-4, Dichlorofluoromethane 75-45-6, Chlorodifluoromethane 75-46-7, Trifluoromethane 75-47-8, Triiodomethane 75-50-3, Trimethylamine, properties 75-52-5, Nitromethane, properties 75-56-9, Propylene oxide, properties tert-Butylamine, properties 75-65-0, tert-Butyl alcohol, properties 75-66-1, 2-Methyl-2-propanethiol 75-69-4, Trichlorofluoromethane 75-71-8, Dichlorodifluoromethane 75-72-9, Chlorotrifluoromethane 75-73-0, Carbon tetrafluoride 75-83-2, 2,2-Dimethylbutane tert-Pentyl alcohol 76-01-7, Pentachloroethane 76-13-1, 1,1,2-Trichlorotrifluoroethane 76-14-2, 1,2-Dichlorotetrafluoroethane 76-16-4, Hexafluoroethane 78-75-1, 1,2-Dibromopropane 76-15-3 78-76-2, 2-Bromobutane 78-78-4 78-79-5, 2-Methyl-1, 3-butadiene, properties 78-82-0, Isobutyronitrile 78-86-4, 2-Chlorobutane 78-87-5, 1,2-Dichloropropane 78-92-2, sec-Butyl alcohol 78-93-3, 2-Butanone, properties 79-00-5, 1,1,2-Trichloroethane 79-01-6, Trichloroethene, properties 79-10-7, Acrylic acid, properties 79-24-3, 79-29-8, 2,3-Dimethylbutane 79-34-5, Nitroethane 1,1,2,2-Tetrachloroethane 79-46-9, 2-Nitropropane 86-89-5, 1-Pentylnaphthalene 87-85-4, Hexamethylbenzene 90-12-0, 1-Methylnaphthalene 91-20-3, Naphthalene, properties 91-57-6, 2-Methylnaphthalene 92-52-4, Biphenyl, properties 93-22-1, 2-Pentylnaphthalene 95-47-6, o-Xylene, properties 95-48-7, o-Cresol, properties 95-50-1, o-Dichlorobenzene 95-63-6, 1,2,4-Trimethylbenzene 95-93-2, 1,2,4,5-Tetramethylbenzene 96-14-0, 3-Methylpentane 96-18-4, 1,2,3-Trichloropropane 96-37-7, Methylcyclopentane 98-08-8 98-83-9, α -Methylstyrene, properties 100-41-4, Cumene Ethylbenzene, properties 100-42-5, Styrene, properties 100-4 Benzonitrile, properties 100-80-1, m-Methylstyrene 102-25-0, 100-47-0, 1,3,5-Triethylbenzene 103-65-1, Propylbenzene 104-51-8, Butylbenzene 104-72-3, 1-Phenyldecane 105-05-5, p-Diethylbenzene 106-42-3, p-Xylene, properties 106-44-5, p-Cresol, properties 106-46-7, p-Dichlorobenzene 106-93-4, 1,2-Dibromoethane 106-94-5, 1-Bromopropane 106-95-6, 3-Bromo-1-propene, properties 106-97-8, Butane, properties 106-98-9, 1-Butene, properties 106-99-0,

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Critical constant

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1,3-Butadiene, properties 107-00-6, 1-Butyne 107-03-9, 1-Propanethiol
107-05-1, 3-Chloro-1-propene 107-06-2, 1,2-Dichloroethane, properties
107-08-4, 1-Iodopropane 107-10-8, Propylamine, properties 107-12-0
Propionitrile 107-13-1, Acrylonitrile, properties 107-18-6, Allyl
                                                                107-12-0,
alcohol, properties 107-21-1, Ethylene glycol, properties 107-31-3,
Methyl formate 107-83-5, 2-Methylpentane 107-84-6,
1-Chloro-3-methylbutane 107-87-9, 2-Pentanone 108-03-2, 1-Nitropropane
108-08-7, 2,4-Dimethylpentane 108-20-3, Isopropyl ether 108-24-7,
Acetic anhydride 108-38-3, m-Xylene, properties 108-39-4, m-Cresol,
            108-67-8, Mesitylene, properties 108-86-1, Bromobenzene,
properties
             108-87-2, Methylcyclohexane 108-88-3, Toluene, properties
properties
108-90-7, Chlorobenzene, properties 108-93-0, Cyclohexanol, properties
108-94-1, Cyclohexanone, properties 108-95-2, Phenol, properties
108-98-5, Benzenethiol, properties 108-99-6, 3-Picoline 109-06-8,
2-Picoline 109-65-9, 1-Bromobutane 109-66-0, Pentane, properties
109-67-1, 1-Pentene 109-69-3, 1-Chlorobutane 109-73-9, Butylamine,
properties 109-74-0, Butyronitrile 109-79-5, 1-Butanethiol 109-89-7,
Diethylamine, properties 110-00-9, Furan 110-01-0, Thiacyclopentane
110-02-1, Thiophene 110-53-2, 1-Bromopentane 110-54-3, Hexane,
properties 110-62-3, Valeraldehyde 110-66-7, 1-Pentanethiol
110-81-6, Ethyl disulfide 110-82-7, Cyclohexane, properties
                                                                110-83-8,
Cyclohexene, properties 110-86-1, Pyridine, properties 111-27-3, Hexyl
alcohol, properties 111-31-9, 1-Hexanethiol 111-43-3, Propyl ether 111-47-7 111-65-9, Octane, properties 111-66-0, 1-Octene 111-70-6,
Heptyl alcohol 111-71-7, Heptanal 111-84-2, Nonane 111-87-5, Octyl
alcohol, properties 111-88-6, 1-Octanethiol 112-30-1, Decyl alcohol
112-31-2, Decanal 112-40-3, Dodecane 112-41-4, 1-Dodecene 112-42-5,
Undecyl alcohol 112-51-6, Pentyl disulfide 112-53-8, Dodecyl alcohol
112-55-0, 1-Dodecanethiol 112-70-9, 1-Tridecanol 112-72-1, 1-Tetradecanol 112-88-9, 1-Octadecene 112-92-5, 1-Octadecanol
112-95-8, Eicosane 115-07-1, Propene, properties 115-10-6, Methyl
       115-11-7, 2-Methylpropene, properties 115-25-3,
ether
Octafluorocyclobutane 116-14-3, Tetrafluoroethene, properties
118-74-1, Hexachlorobenzene 121-44-8, Triethylamine, properties
123-01-3, 1-Phenyldodecane 123-02-4, 1-Phenyltridecane
                                                            123-38-6,
Propionaldehyde, properties 123-72-8, Butyraldehyde 123-75-1,
                         123-91-1, p-Dioxane, properties 124-11-8,
Pyrrolidine, properties
          124-13-0, Octanal
                              124-18-5, Decane 124-19-6, Nonanal
1-Nonene
124-38-9, Carbon dioxide, properties 124-40-3, Dimethylamine, properties
127-18-4, Tetrachloroethene, properties 135-01-3, o-Diethylbenzene
141-78-6, Ethyl acetate, properties 141-93-5, m-Diethylbenzene
142-28-9, 1,3-Dichloropropane 142-29-0, Cyclopentene 142-82-5,
Heptane, properties 142-96-1, Butyl ether 143-08-8, Nonyl alcohol
143-10-2, 1-Decanethiol 151-56-4, Ethylenimine, properties 156-59-2,
                         156-60-5, trans-1,2-Dichloroethene
                                                               157-40-4,
cis-1,2-Dichloroethene
Spiropentane 275-51-4, Azulene 287-23-0, Cyclobutane 287-27-4,
Thiacyclobutane 287-92-3, Cyclopentane
RL: PRP (Properties)
   (critical consts. of, selected values for)
291-64-5, Cycloheptane 292-64-8, Cyclooctane
                                                  352-32-9, p-Fluorotoluene
                        353-36-6, Fluoroethane 359-11-5,
352-93-2, Ethylsulfide
Trifluoroethene 367-11-3, o-Difluorobenzene
                                                 372-18-9,
m-Difluorobenzene 392-56-3, Hexafluorobenzene 420-12-2, Thiacyclopropane 420-26-8, 2-Fluoropropane 420-46-2,
1,1,1-Trifluoroethane 460-12-8, 1,3-Butadiyne 460-13-9,
1-Fluoropropane 460-19-5, Cyanogen 462-06-6, Fluorobenzene
                                                                   463-49-0,
1,2-Propadiene 463-51-4, Ketene 463-58-1, Carbonyl sulfide
                                                                  463-82-1
464-06-2, 2,2,3-Trimethylbutane 488-23-3, 1,2,3,4-Tetramethylbenzene
493-01-6 493-02-7 503-17-3, 2-Butyne 507-09-5, Thioacetic acid,
properties 507-19-7, 2-Bromo-2-methylpropane 507-20-0,
2-Chloro-2-methylpropane 513-35-9, 2-Methyl-2-butene 513-36-0,
1-Chloro-2-methylpropane 513-44-0, 2-Methyl-1-propanethiol 513-53-1,
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2-Butanethiol 526-73-8, 1,2,3-Trimethylbenzene 527-53-7,
1,2,3,5-Tetramethylbenzene 533-98-2, 1,2-Dibromobutane 536-74-3,
Ethynylbenzene 538-68-1, Pentylbenzene 540-36-3, p-Difluorobenzene
540-54-5, 1-Chloropropane 540-67-0, Ethyl methyl ether 540-84-1,
2,2,4-Trimethylpentane 541-73-1, m-Dichlorobenzene 543-59-9,
1-Chloropentane 544-25-2, 1,3,5-Cycloheptatriene 544-40-1, Butyl
sulfide 544-76-3, Hexadecane 554-14-3, 2-Methylthiophene 556-56-9,
3-Iodo-1-propene 557-17-5, Methyl propyl ether 558-17-8,
2-Iodo-2-methylpropane 558-37-2, 3,3-Dimethyl-1-butene 560-21-4,
2,3,3-Trimethylpentane 562-49-2, 3,3-Dimethylpentane
                                                              563-16-6,
3,3-Dimethylhexane 563-45-1, 3-Methyl-1-butene 563-46-2,
2-Methyl-1-butene 563-78-0, 2,3-Dimethyl-1-butene 563-79-1,
2,3-Dimethyl-2-butene 564-02-3, 2,2,3-Trimethylpentane 565-59-3,
2,3-Dimethylpentane 565-75-3, 2,3,4-Trimethylpentane 571-58-4,
1,4-Dimethylnaphthalene 571-61-9, 1,5-Dimethylnaphthalene 573-98-8,
1,2-Dimethylnaphthalene 575-37-1, 1,7-Dimethylnaphthalene
                                                                    575-41-7,
1,3-Dimethylnaphthalene 575-43-9, 1,6-Dimethylnaphthalene
                                                                    581-40-8,
2,3-Dimethylnaphthalene 581-42-0, 2,6-Dimethylnaphthalene 582-2,7-Dimethylnaphthalene 583-48-2, 3,4-Dimethylnaphthalene 584-94-1,
                                                                    582-16-1,
2,3-Dimethylhexane 589-34-4, 3-Methylhexane 589-43-5, 2,4-Dimethylhexane 589-53-7, 4-Methylheptane 589-81-1, 3-Methylheptane 590-18-1 590-19-2, 1,2-Butadiene 590-35-2, 2,2-Dimethylpentane 590-66-9, 1,1-Dimethylcyclohexane 590-73-8, 2,2-Dimethylhexane
591-50-4, Iodobenzene 591-76-4, 2-Methylhexane 591-93-5,
1,4-Pentadiene 591-95-7, 1,2-Pentadiene 591-96-8, 2,3-Pentadiene
592-13-2, 2,5-Dimethylhexane 592-27-8, 2-Methylheptane 592-41-6,
1-Hexene, properties 592-76-7, 1-Heptene 593-45-3, Octadecane
593-53-3, Fluoromethane 593-60-2, Bromoethylene 593-70-4,
Chlorofluoromethane 594-20-7, 2,2-Dichloropropane 594-36-5,
2-Chloro-2-methylbutane 594-51-4, 2,3-Dibromo-2-methylbutane
                                                                       594-82-1,
2,2,3,3-Tetramethylbutane 598-23-2, 3-Methyl-1-butyne 598-25-4,
3-Methyl-1,2-butadiene 598-29-8, 1,2-Diiodopropane 598-53-8, Methyl
isopropyl ether 598-58-3, Methyl nitrate 600-24-8, 2-Nitrobutane
604-88-6, Hexaethylbenzene 605-01-6, Pentaethylbenzene 609-26-7,
3-Ethyl-2-methylpentane 611-14-3, o-Ethyltoluene 611-15-4,
o-Methylstyrene 616-12-6 616-44-4, 3-Methylthiophene 617-78-7,
                  619-99-8, 3-Ethylhexane 620-14-4, m-Ethyltoluene
3-Ethylpentane
622-96-8
           622-97-9, p-Methylstyrene
                                         624-29-3
                                                     624-64-6 624-73-7,
1,2-Diiodoethane 624-89-5, Ethyl methyl sulfide 624-91-9, Methyl
nitrite 624-92-0, Methyl disulfide
                                         625-27-4, 2-Methyl-2-pentene
625-58-1, Ethyl nitrate 625-80-9, Isopropyl sulfide 627-05-4,
1-Nitrobutane
                 627-13-4, Propyl nitrate 627-19-0, 1-Pentyne 627-20-3
627-21-4, 2-Pentyne 628-29-5, Butyl methyl sulfide 628-71-7, 1-Heptyne
629-05-0, 1-Octyne 629-19-6, Propyl disulfide
                                                       629-20-9,
1,3,5,7-Cyclooctatetraene 629-45-8, Butyl disulfide 629-50-5,
Tridecane 629-59-4, Tetradecane 629-62-9, Pentadecane 629-65-2,
                 629-73-2, 1-Hexadecene 629-74-3, 1-Hexadecyne
Heptyl sulfide
629-76-5, 1-Pentadecanol 629-78-7, Heptadecane 629-89-0, 1-Octadecyne
629-92-5, Nonadecane 629-96-9, 1-Eicosanol 630-08-0, Carbon monoxide, properties 635-81-4, 1,2,4,5-Tetraethylbenzene 638-04-0 638-46-0,
Butyl ethyl sulfide 642-32-0, 1,2,3,4-Tetraethylbenzene 646-04-8
674-76-0 689-97-4, 1-Buten-3-yne 691-37-2, 4-Methyl-1-pentene
           693-02-7, 1-Hexyne 693-83-4, Decyl sulfide 693-89-0,
1-Methylcyclopentene 700-12-9, Pentamethylbenzene
                                                          760-20-3,
3-Methyl-1-pentene 760-21-4, 2-Ethyl-1-butene 763-29-1, 2-Methyl-1-pentene 764-93-2, 1-Decyne 765-03-7, 1-Dodecyne 1-Tetradecyne 765-13-9, 1-Pentadecyne 765-27-5, 1-Eicosyne
                                                                       765-10-6,
821-95-4, 1-Undecene 822-27-5, Octyl disulfide 822-35-5, Cyclobutene 822-50-4 871-83-0, 2-Methylnonane 872-05-9, 1-Decene 872-10-6, Pentyl sulfide 873-66-5 877-44-1, 1,2,4-Triethylbenzene 921-47-1,
2,3,4-Trimethylhexane 922-28-1, 3,4-Dimethylheptane 922-62-3
926-82-9, 3,5-Dimethylheptane 929-98-6, Nonyl sulfide 939-27-5,
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2-Ethylnaphthalene 1067-08-9, 3-Ethyl-3-methylpentane 1067-20-5,
    3,3-Diethylpentane 1068-19-5, 4,4-Dimethylheptane 1068-87-7,
    3-Ethyl-2,4-dimethylpentane 1069-53-0, 2,3,5-Trimethylhexane
    1070-87-7, 2,2,4,4-Tetramethylpentane 1071-26-7, 2,2-Dimethylheptane 1071-81-4, 2,2,5,5-Tetramethylhexane 1072-05-5, 2,6-Dimethylheptane
    1072-16-8, 2,7-Dimethyloctane 1077-16-3, Hexylbenzene 1078-71-3,
    1-Phenylheptane 1081-77-2, 1-Phenylnonane 1120-21-4, Undecane
    1120-36-1, 1-Tetradecene 1120-62-3, 3-Methylcyclopentene
                                                              1127-76-0,
    1-Ethylnaphthalene 1134-62-9, 2-Butylnaphthalene 1186-53-4,
    2,2,3,4-Tetramethylpentane 1189-99-7, 2,5,5-Trimethylpentane
    1190-83-6, 2,2,6-Trimethylheptane 1192-18-3 1454-84-8, 1-Nonadecanol
    1454-85-9, 1-Heptadecanol 1455-21-6, 1-Nonanethiol 1459-09-2,
    1-Phenylhexadecane 1459-10-5, 1-Phenyltetradecane 1551-21-9, Isopropyl
    methyl sulfide 1574-41-0 1613-46-3, Butyl propyl sulfide 1613-51-0,
    Thiacyclohexane 1630-77-9, cis-1,2-Difluoroethene 1630-78-0,
    trans-1,2-Difluoroethene 1634-04-4, Methyl-tert-butyl ether 1634-09-9,
    1-Butylnaphthalene 1638-26-2, 1,1-Dimethylcyclopentane 1639-09-4,
    1-Heptanethiol 1640-89-7, Ethylcyclopentane 1678-91-7,
    Ethylcyclohexane 1678-92-8, Propylcyclohexane 1678-93-9,
                     1679-07-8, Cyclopentanethiol 1679-09-0,
    Butylcyclohexane
    2-Methyl-2-butanethiol 1712-64-7, Isopropyl nitrate 1741-83-9, Methyl
    pentyl sulfide 1759-58-6 1759-81-5, 4-Methylcyclopentene 1795-15-9,
    1-Cyclohexyloctane 1795-16-0, 1-Cyclohexyldecane 1795-17-1,
    1-Cyclohexyldodecane 1795-18-2, 1-Cyclohexyltetradecane 1795-20-6
    1795-21-7, 1-Cyclopentyldecane 1795-22-8, 1-Cyclopentyltetradecane
    RL: PRP (Properties)
       (critical consts. of, selected values for)
ΙT
    1795-26-2 1795-27-3 2004-70-8 2027-19-2, 2-Propylnaphthalene
    2040-95-1, Butylcyclopentane 2040-96-2, Propylcyclopentane 2051-30-1,
    2,6-Dimethyloctane 2074-87-5, Cyanogen 2079-95-0, 1-Tetradecanethiol
    2131-18-2, 1-Phenylpentadecane 2189-60-8, 1-Phenyloctane 2207-01-4
    2207-03-6 2207-04-7 2213-23-2, 2,4-Dimethylheptane
                                                          2216-30-0,
    2,5-Dimethylheptane 2216-32-2, 4-Ethylheptane 2216-33-3,
    3-Methyloctane 2216-34-4, 4-Methyloctane 2243-98-3, 1-Undecyne
    2437-56-1, 1-Tridecene 2532-58-3 2613-61-8, 2,4,6-Trimethylheptane
    2690-08-6, Octyl sulfide 2765-18-6, 1-Propylnaphthalene 2851-83-4,
                          2882-98-6, 1-Cyclopentylnonane 2883-02-5,
    Dodecyl ethyl sulfide
    1-Cyclohexylnonane 2885-00-9, 1-Octadecanethiol
    2917-26-2, 1-Hexadecanethiol
                                 3074-71-3
                                              3074-75-7,
                           3074-76-8, 3-Ethyl-3-methylhexane
    4-Ethyl-2-methylhexane
    3-Ethyl-4-methylhexane 3129-90-6, Isothiocyanic acid 3178-29-8,
    4-Propylheptane 3221-61-2, 2-Methyloctane 3452-07-1, 1-Eicosene
    3452-09-3, 1-Nonyne 3522-94-9, 2,2,5-Trimethylhexane 3698-89-3,
    Dodecyl methyl sulfide 3698-93-9, Octyl propyl sulfide 3698-94-0,
                        3698-95-1, Methyl octyl sulfide 3741-00-2
    Ethyl octyl sulfide
    3877-15-4, Methyl propyl sulfide 4032-86-4, 3,3-Dimethylheptane
    4032-92-2, 2,4,4-Trimethylheptane 4032-93-3, 2,3,6-Trimethylheptane
    4032-94-4, 2,4-Dimethyloctane 4050-45-7 4110-44-5, 3,3-Dimethyloctane
    4110-50-3, Ethyl propyl sulfide 4292-75-5, 1-Cyclohexylhexane
    4292-92-6, Pentylcyclohexane 4457-00-5 4485-77-2, Nonyl disulfide
    4669-01-6, 1-Cyclopentylpentadecane 4753-80-4, Thiacycloheptane
    5171-84-6, 3,3,4,4-Tetramethylhexane 5332-52-5, 1-Undecanethiol
    5408-86-6, 2,3-Dibromobutane 5617-41-4 5617-42-5, 1-Cyclopentylheptane
    5634-30-0, 1-Cyclopentyldodecane 5881-17-4, 3-Ethyloctane 5911-04-6,
    3-Methylnonane 6006-33-3, 1-Cyclohexyltridecane
                                                      6006-34-4,
    1-Cyclopentyltridecane 6006-95-7, 1-Cyclohexylpentadecane 6163-66-2,
    tert-Butyl ether 6294-31-1, Hexyl sulfide 6742-54-7, 1-Phenylundecane
    6765-39-5, 1-Heptadecene 6785-23-5, 1-Cyclopentylundecane 6812-38-0,
    1-Cyclohexylhexadecane 6812-39-1, 1-Cyclopentylhexadecane
                                                                 6863-58-7,
    sec-Butyl ether 6876-18-2, 3-Isopropyl-2-methylhexane
                                                           6876-23-9
    7146-60-3, 2,3-Dimethyloctane 7154-79-2, 2,2,3,3-Tetramethylpentane
    7154-80-5, 3,3,5-Trimethylheptane 7220-26-0, 3-Ethyl-2,4-dimethylhexane
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7289-44-3, Methyl undecyl sulfide 7289-45-4, Methyl tetradecyl sulfide
7309-44-6, Ethyl hexyl sulfide 7372-86-3, 2-Ethyl-6-methylnaphthalene
           7688-21-3 10496-15-8, Hexyl disulfide 10496-16-9, Heptyl
7642-09-3
            10496-18-1, Decyl disulfide 13269-52-8
                                                        13360-61-7,
disulfide
1-Pentadecene 13373-97-2, 1-Eicosanethiol 13475-78-0,
5-Ethyl-2-methylheptane 13475-79-1, 2,4-Dimethyl-3-isopropylpentane
13475-81-5, 2,2,3,3-Tetramethylhexane 13952-84-6, sec-Butylamine
14676-29-0, 3-Ethyl-2-methylheptane 14720-74-2, 2,2,4-Trimethylheptane
15869-80-4, 3-Ethylheptane 15869-85-9, 5-Methylnonane 15869-86-0,
               15869-87-1, 2,2-Dimethyloctane 15869-89-3,
4-Ethyloctane
2,5-Dimethyloctane 15869-92-8, 3,4-Dimethyloctane
                                                       15869-93-9,
3,5-Dimethyloctane 15869-94-0, 3,6-Dimethyloctane
                                                        15869-95-1,
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        (critical consts. of, selected values for)
L23 ANSWER 6 OF 10 CAPLUS COPYRIGHT 2009 ACS on STN
     1988:617106 CAPLUS
     109:217106
OREF 109:35822h,35823a
     Entered STN: 10 Dec 1988
     Enthalpy of formation for 700 major organic compounds Yaws, Carl L.; Chiang, P. Y.
     Lamar Univ., Beaumont, TX, USA
     Chemical Engineering (New York, NY, United States) (1988), 95(13), 81-8
     CODEN: CHEEA3; ISSN: 0009-2460
     Journal
     English
     69-2 (Thermodynamics, Thermochemistry, and Thermal Properties)
     A correlation method for the ideal-gas heat of formation as function of
     temperature was applied and values were calculated for 700 organic compds.
The calculated
     values were compared to the exptl. data and an average deviation of 0.2 kJ/mol
     was found. Consts. for the correlation equation, \Delta Hf = A + BT +
     CT2, with \Delta Hf in kJ/mol and T in {}^{\circ}K, are tabulated for 700
     compds. Values at 298 K are also listed.
     heat formation org compd ideal gas
     Heat of formation
        (calcn. of, of organic compds. in ideal gas state, equation for)
     Organic compounds, properties
     RL: PRP (Properties); FORM (Formation, nonpreparative)
        (heats of formation of, in ideal gas state, equation for calcn. of)
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2,3,5-Trimethylhexane 1070-87-7, 2,2,4,4-Tetramethylpentane 1071-26-7,
2,2-Dimethylheptane 1071-81-4, 2,2,5,5-Tetramethylhexane 1072-05-5
1072-16-8, 2,7-Dimethyloctane 1077-16-3, Hexylbenzene
                                                         1078-71-3,
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                                                    1186-53-4,
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1190-83-6, 2,2,6-Trimethylheptane 1192-18-3 1454-84-8, 1-Nonadecanol
1454-85-9, 1-Heptadecanol 1455-21-6, 1-Nonanethiol 1459-09-2,
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                                                               1613-51-0,
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                   1638-26-2, 1,1-Dimethylcyclopentane 1639-09-4,
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               1640-89-7, Ethylcyclopentane 1678-91-7,
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                                                  1678-93-9,
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1-Cyclohexyloctane 1795-16-0, 1-Cyclohexyldecane 1795-17-1,
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RL: PRP (Properties)
   (heat of formation of, in ideal gas state, equation for calcn. of)
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     2040-95-1, Butylcyclopentane 2040-96-2, Propylcyclopentane
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     2,5-Dimethylheptane 2216-32-2, 4-Ethylheptane
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     1-Eicosene 3452-09-3, 1-Nonyne 3522-94-9, 2,2,5-Trimethylhexane 3698-89-3, Dodecyl methyl sulfide 3698-93-9, Octyl propyl sulfide
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     3741-00-2 3877-15-4, Methyl propyl sulfide 4032-86-4,
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     1-Cyclohexylpentadecane 6163-66-2, tert-Butyl ether 6294-31-1, Hexyl
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     6785-23-5, 1-Cyclopentylundecane 6812-38-0, 1-Cyclohexylhexadecane
     6812-39-1, 1-Cyclopentylhexadecane 6863-58-7, sec-Butyl ether
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     7220-26-0, 3-Ethyl-2,4-dimethylhexane 7289-44-3, Methyl undecyl sulfide
     7289-45-4, Methyl tetradecyl sulfide 7309-44-6, Ethyl hexyl sulfide
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     2,5-Dimethyloctane 15869-92-8, 3,4-Dimethyloctane
                                                             15869-93-9,
                         15869-94-0, 3,6-Dimethyloctane
     3,5-Dimethyloctane
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     4,4-Dimethyloctane 15869-96-2, 4,5-Dimethyloctane 16747-25-4,
     2,2,3-Trimethylhexane 16747-26-5, 2,2,4-Trimethylhexane
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     2,3,3-Trimethylhexane 16747-30-1, 2,4,4-Trimethylhexane 16747-32-3, 3-Ethyl-2,2-dimethylpentane
                                                                    16747-31-2,
     16747-33-4, 3-Ethyl-2,3-dimethylpentane 16747-38-9,
     2,3,3,4-Tetramethylpentane 16747-42-5, 2,2,4,5-Tetramethylhexane
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     17302-02-2, 3,3-Diethylhexane 17302-04-4, 4-Ethyl-4-methylheptane
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     18437-89-3, Butyl hexadecyl sulfide 19313-57-6, Butyl decyl sulfide
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    19484-26-5, 1-Tridecanethiol
    20278-85-7, 2,3,5-Trimethylheptane 20278-87-9, 3,3,4-Trimethylheptane
    20278-88-0, 3,4,4-Trimethylheptane 20291-60-5, Hexyl methyl sulfide
    20291-61-6, Heptyl methyl sulfide 20291-91-2, 3-Ethyl-2,2-dimethylhexane
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    24768-44-3, Ethyl heptyl sulfide 24768-46-5, Heptyl propyl sulfide
    25276-70-4, 1-Pentadecanethiol
                                     26158-99-6, Ethyl pentyl sulfide
    26186-00-5, 1-Heptadecyne
                                26186-01-6, 1-Nonadecyne
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    2,2,4,4-Tetramethylhexane 52896-87-4, 4-Isopropylheptane
                                                                 52896-88-5,
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    52896-90-9, 3-Ethyl-5-methylheptane 52896-91-0, 3-Ethyl-4-methylheptane
    52896-92-1, 2,2,3-Trimethylheptane 52896-93-2, 2,3,3-Trimethylheptane
    52896-95-4, 2,3,4-Trimethylheptane 52896-99-8,
    4-Ethyl-2,2-dimethylhexane 52897-00-4, 3-Ethyl-2,3-dimethylhexane
    52897-01-5, 4-Ethyl-2,3-dimethylhexane
                                             52897-03-7,
     4-Ethyl-2,4-dimethylhexane 52897-04-8, 3-Ethyl-2,5-dimethylhexane
    52897-05-9, 4-Ethyl-3,3-dimethylhexane
                                             52897-06-0,
    3-Ethyl-3,4-dimethylhexane 52897-08-2, 2,2,3,4-Tetramethylhexane
    52897-09-3, 2,2,3,5-Tetramethylhexane 52897-10-6,
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    52897-12-8, 2,3,4,4-Tetramethylhexane 52897-15-1, 2,3,4,5-Tetramethylhexane 52897-16-2, 3,3-Diethyl-2-methylpentane
    52897-17-3, 3-Ethyl-2,2,3-trimethylpentane
                                                52897-18-4,
    3-Ethyl-2, 2, 4-trimethylpentane 52897-19-5,
    3-Ethyl-2,3,4-trimethylpentane 53161-72-1, 1,2-Diiodobutane
    53193-22-9, 1-Heptadecanethiol 53193-23-0, 1-Nonadecanethiol
    54105-66-7, 1-Cyclohexylundecane 59973-07-8, Methyl nonyl sulfide
    59973-08-9, Ethyl nonyl sulfide 62103-66-6, Nonyl propyl sulfide
    62155-09-3, Methyl tridecyl sulfide
                                          62155-10-6, Methyl pentadecyl
             62155-11-7, Heptadecyl methyl sulfide 62155-12-8, Methyl
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                        64919-20-6, Ethyl pentadecyl sulfide
                                                               66271-54-3,
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    66271-81-6, Ethyl tridecyl sulfide 66271-82-7, Dodecyl propyl sulfide
    66271-83-8, Butyl undecyl sulfide
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    66292-32-8, Pentadecyl propyl sulfide 66292-33-9, Butyl tetradecyl
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    66577-31-9, Decyl propyl sulfide 66577-32-0, Butyl nonyl sulfide
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    RL: PRP (Properties)
        (heat of formation of, in ideal gas state, equation for calcn. of)
L23 ANSWER 7 OF 10 CAPLUS COPYRIGHT 2009 ACS on STN
    1988:428388 CAPLUS
    109:28388
OREF 109:4739a,4742a
    Entered STN: 22 Jul 1988
    Heat capacities for 700 compounds
    Yaws, Carl L.; Ni, H. M.; Chiang, P. Y.
    Lamar Univ., Beaumont, TX, USA
    Chemical Engineering (New York, NY, United States) (1988), 95(7), 91-8
    CODEN: CHEEA3; ISSN: 0009-2460
    Journal
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LA English 69-2 (Thermodynamics, Thermochemistry, and Thermal Properties) CC AΒ Consts. (A, B, C, D) for the correlation equation Cp = A + B + T + C+ T2 + D + T3 (where Cp is the heat capacity in J/mol. K and T is the temperature in K) are tabulated for (mostly) 700 organic compds. consts. were obtained by applying a generalized least-squares computer program to data found in the literature. The average deviation claimed for the calculated values (from the original data) is 0.15%. ST heat capacity org compd equation ΙT Heat capacity (calcn. of, of chemical compds., equation for) ΙT Organic compounds, properties RL: PRP (Properties) (heat capacities of, calcn. of) ΙT 50-00-0, Formaldehyde, properties 56-23-5, Carbon tetrachloride, properties 60-29-7, Ethyl ether, properties 62-53-3, Aniline, 64-17-5, Ethyl alcohol, properties 64-18-6, Formic acid, properties 64-19-7, Acetic acid, properties 65-85-0, Benzoic acid, properties 66-25-1, Hexanal 67-56-1, Methanol, properties 67-63-0, properties 2-Propanol, properties 67-64-1, Acetone, properties 67-66-3, properties 67-72-1, Hexachloroethane 71-23-8, Propyl alcohol, 71-36-3, Butyl alcohol, properties 71-41-0, Pentyl alcohol, properties properties 71-43-2, Benzene, properties 74-82-8, Methane, properties 74-83-9, Bromomethane, properties 74-84-0, Ethane, properties 74-85-1Ethylene, properties 74-86-2, Ethyne, properties 74-87-3, Chloromethane, properties 74-88-4, Iodomethane, properties 74-89-5. Methylamine, properties 74-93-1, Methanethiol, properties 74-96-4 74-98-6, Propane, properties 74-99-7, 1-Propyne 75-00-3, Chloroethane 75-01-4, properties 75-02-5, Fluoroethene 75-03-6, Iodoethane 75-04-7, Ethylamine, properties 75-05-8, Acetonitrile, properties 75-07-0, Acetaldehyde, properties 75-09-2, properties 75-10-5, Difluoromethane 75-11-6, Diiodomethane 75-15-0, Carbon disulfide, 75-18-3, Methyl sulfide 75-19-4, Cyclopropane 75-21-8, properties Ethylene oxide, properties 75-26-3, 2-Bromopropane 75-28-5 75-29-6, 2-Chloropropane 75-30-9, 2-Iodopropane 75-33-2, 2-Propanethiol 75-34-3 75-35-4, 1,1-Dichloroethene, properties 75-36-5, Acetyl chloride 75-37-6, 1,1-Difluoroethane 75-38-7 75-43-4, Dichlorofluoromethane 75-45-6, Chlorodifluoromethane 75-46-7, Trifluoromethane 75-47-8, Triiodomethane 75-50-3, Trimethylamine, 75-52-5, Nitromethane, properties 75-56-9, Propylene oxide, properties 75-64-9, tert-Butylamine, properties 75-65-0, properties properties 75-66-1, 2-Methyl-2-propanethiol 75-69-4, Trichlorofluoromethane 75-71-8, Dichlorodifluoromethane 75-72-9, Chlorotrifluoromethane 75-73-0, Carbon tetrafluoride 75-83-2, 2,2-Dimethylbutane 75-85-4, 76-01-7 76-13-1, 1,1,2-Trichlorotrifluoroethane tert-Pentyl alcohol 76-15-3, Chloropentafluoroethane 76-16-4, Hexafluoroethane 76-14-2 78-75-1, 1,2-Dibromopropane 78-76-2, 2-Bromobutane 78-79-5**,** 2-Methyl-1,3-butadiene, properties 78-82-0, Isobutyronitrile 78-86-4. 2-Chlorobutane 78-87-5, 1,2-Dichloropropane 78-92-2, sec-Butyl alcohol 78-93-3, 2-Butanone, properties 79-00-5, 1,1,2-Trichloroethane 79-01-6, Trichloroethene, properties 79-10-7, Acrylic acid, properties 79-24-3, Nitroethane 79-29-8, 2,3-Dimethylbutane 79-34-5, 1,1,2,2-Tetrachloroethane 79-46-9, 2-Nitropropane 86-89-5, 1-Pentylnaphthalene 87-85-4, Hexamethylbenzene 90-12-0, 1-Methylnaphthalene 91-20-3, Naphthalene, properties 91-57-6, 2-Methylnaphthalene 92-52-4, Biphenyl, properties 93-22-1, 2-Pentylnaphthalene 95-47-6, o-Xylene, properties 95-48-7, o-Cresol, properties 95-50-1, o-Dichlorobenzene 95-63-6, 1,2,4-Trimethylbenzene 95-93-2, 1,2,4,5-Tetramethylbenzene 96-14-0, 3-Methylpentane 96-18-4, 1,2,3-Trichloropropane 96-37-7, Methylcyclopentane 98-08-8 Cumene 98-83-9, properties 100-41-4, Ethylbenzene, properties

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110-02-1, Thiophene 110-53-2, 1-Bromopentane 110-54-3, Hexane, properties 110-62-3, Valeraldehyde 110-66-7, 1-Pentanethiol
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RL: PRP (Properties)
   (heat capacity of, equation for calcn. of)
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493-02-7
507-19-7, 2-Bromo-2-methylpropane 507-20-0, 2-Chloro-2-methylpropane
513-35-9, 2-Methyl-2-butene 513-36-0, 1-Chloro-2-methylpropane
513-44-0, 2-Methyl-1-propanethiol 513-53-1, 2-Butanethiol
                                                                   526-73-8,
1,2,3-Trimethylbenzene 527-53-7, 1,2,3,5-Tetramethylbenzene 533-98-2,
1,2-Dibromobutane
                    536-74-3, Ethynylbenzene 538-68-1, Pentylbenzene
540-36-3, p-Difluorobenzene 540-54-5, 1-Chloropropane 540-67-0,
                     540-84-1, 2,2,4-Trimethylpentane
Ethylmethyl ether
                     543-59-9, 1-Chloropentane 544-25-2,
m-Dichlorobenzene
1,3,5-Cycloheptatriene 544-40-1, Butyl sulfide 544-76-3, Hexadecane
554-14-3, 2-Methylthiophene 556-56-9, 3-Iodo-1-propene 557-17-5,
Methyl propyl ether 558-17-8, 2-Iodo-2-methylpropane 558-37-2,
3,3-Dimethyl-1-butene 560-21-4, 2,3,3-Trimethylpentane 562-49-2,
3,3-Dimethylpentane 563-16-6, 3,3-Dimethylhexane 563-45-1,
3-Methyl-1-butene 563-46-2, 2-Methyl-1-butene 563-78-0,
2,3-Dimethyl-1-butene 563-79-1, 2,3-Dimethyl-2-butene
                                                             564-02-3,
2,2,3-Trimethylpentane 565-59-3, 2,3-Dimethylpentane
                                                             565-75-3,
2,3,4-Trimethylpentane 571-58-4, 1,4-Dimethylpaphthalene
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2,3,4-Trimethylpentane
1,5-Dimethylnaphthalene
1,7-Dimethylnaphthalene
1,6-Dimethylnaphthalene
2,6-Dimethylnaphthalene
3/1-30-4, 1,4 Dimethylnaphthalene
573-98-8, 1,2-Dimethylnaphthalene
575-41-7, 1,3-Dimethylnaphthalene
581-40-8, 2,3-Dimethylnaphthalene
582-16-1, 2,7-Dimethylnaphthalene
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                                                                  575-43-9,
                                                                   581-42-0,
                                                                   583-48-2.
3,4-Dimethylhexane 584-94-1, 2,3-Dimethylhexane 589-34-4,
3-Methylhexane 589-43-5, 2,4-Dimethylhexane 589-53-7, 4-Methylheptane
589-81-1, 3-Methylheptane 590-18-1 590-19-2, 1,2-Butadiene 590-35-2,
2,2-Dimethylpentane 590-66-9, 1,1-Dimethylcyclohexane 590-73-8,
2,2-Dimethylhexane 591-50-4, Iodobenzene 591-76-4, 2-Methylhexane
591-93-5, 1,4-Pentadiene 591-95-7, 1,2-Pentadiene 591-96-8,
2,3-Pentadiene 592-13-2, 2,5-Dimethylhexane 592-27-8, 2-Methylheptane
592-41-6, 1-Hexene, properties 592-76-7, 1-Heptene 593-45-3,
Octadecane 593-53-3, Fluoromethane 593-60-2, Bromoethylene
                                                                      593-70-4
594-20-7, 2,2-Dichloropropane 594-36-5, 2-Chloro-2-methylbutane
594-51-4, 2,3-Dibromo-2-methylbutane 594-82-1, 2,2,3,3-Tetramethylbutane
598-23-2, 3-Methyl-1-butyne 598-25-4, 3-Methyl-1,2-butadiene 598-29-8,
                    598-53-8, Methyl isopropyl ether 598-58-3, Methyl
1,2-Diiodopropane
          600-24-8, 2-Nitrobutane 604-88-6, Hexaethylbenzene
Pentaethylbenzene 609-26-7, 3-Ethyl-2-methylpentane 611-14-3,
                  611-15-4, o-Methylstyrene 616-12-6 616-44-4,
o-Ethyltoluene
                    617-78-7, 3-Ethylpentane 619-99-8, 3-Ethylhexane
3-Methylthiophene
620-14-4, m-Ethyltoluene 622-96-8, p-Ethyltoluene 622-97-9,
p-Methylstyrene 624-29-3 624-64-6 624-73-7, 1,2-Diiodoethane
624-89-5, Ethylmethyl sulfide 624-91-9, Methyl nitrite 624-92-0,
Methyl disulfide 625-27-4, 2-Methyl-2-pentene 625-58-1, Ethyl nitrate
625-80-9, Isopropyl sulfide 627-05-4, 1-Nitrobutane 627-13-4, Propyl
nitrate 627-19-0, 1-Pentyne 627-20-3 627-21-4, 2-Pentyne 628-29-5,
Butylmethyl sulfide 628-71-7, 1-Heptyne 629-05-0, 1-Octyne 629-19-6,
Propyl disulfide 629-20-9, 1,3,5,7-Cyclooctatetraene 629-45-8, Butyl
           629-50-5, Tridecane 629-59-4, Tetradecane
disulfide
                                                             629-62-9,
Pentadecane 629-65-2, Heptyl sulfide 629-73-2, 1-Hexadecene
629-74-3, 1-Hexadecyne 629-76-5, 1-Pentadecanol 629-78-7, Heptadecane 629-89-0, 1-Octadecyne 629-92-5, Nonadecane 629-96-9, 1-Eicosanol
629-89-0, 1-Octadecyne
630-08-0, Carbon monoxide, properties
                                           635-81-4,
1,2,4,5-Tetraethylbenzene 638-04-0 638-46-0, Butylethyl sulfide
642-32-0, 1,2,3,4-Tetraethylbenzene 646-04-8 674-76-0 689-97-4,
1-Buten-3-yne 691-37-2, 4-Methyl-1-pentene 691-38-3 693-02-7, 1-Hexyne 693-83-4, Decyl sulfide 693-89-0, 1-Methylcyclopentene
700-12-9 760-20-3, 3-Methyl-1-pentene 760-21-4, 2-Ethyl-1-butene 763-29-1, 2-Methyl-1-pentene 764-93-2, 1-Decyne 765-03-7, 1-Dodecyne 765-10-6, 1-Tetradecyne 765-13-9, 1-Pentadecyne 765-27-5, 1-Eicosyne
766-90-5 821-95-4, 1-Undecene 822-27-5 822-35-5, Cyclobutene
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822-50-4 871-83-0, 2-Methylnonane 872-05-9, 1-Decene 872-10-6, Pentyl sulfide 873-66-5 877-44-1, 1,2,4-Triethylbenzene 921-47
                                                                 921-47-1,
2,3,4-Trimethylhexane 922-28-1, 3,4-Dimethylheptane 922-62-3
926-82-9, 3,5-Dimethylheptane 929-98-6, Nonyl sulfide 939-27-5, 2-Ethylnaphthalene 1067-08-9, 3-Ethyl-3-methylpentane 1067-20-5
                                                             1067-20-5
3,3-Diethylpentane 1068-19-5, 4,4-Dimethylheptane 1068-87-7,
3-Ethyl-2,4-dimethylpentane 1069-53-0, 2,3,5-Trimethylhexane
1070-87-7, 2,2,4,4-Tetramethylpentane 1071-26-7, 2,2-Dimethylheptane
1071-81-4, 2,2,5,5-Tetramethylhexane 1072-05-5, 2,6-Dimethylheptane
1072-16-8, 2,7-Dimethyloctane 1077-16-3, Hexylbenzene 1078-71-3,
1-Phenylheptane 1081-77-2, 1-Phenylnonane 1120-21-4, Undecane
1120-36-1, 1-Tetradecene 1120-62-3, 3-Methylcyclopentene
1-Ethylnaphthalene 1134-62-9, 2-Butylnaphthalene 1186-53-4,
2,2,3,4-Tetramethylpentane 1189-99-7, 2,5,5-Trimethylheptane
1190-83-6, 2,2,6-Trimethylheptane 1192-18-3 1454-84-8, 1-Nonadecanol
1454-85-9, 1-Heptadecanol 1455-21-6, 1-Nonanethiol 1459-09-2,
1-Phenylhexadecane 1459-10-5 1551-21-9, Isopropyl methyl sulfide
1574-41-0 1613-46-3, Butylpropyl sulfide 1613-51-0, Thiacyclohexane
1630-77-9, cis-1,2-Difluoroethene 1630-78-0, trans-1,2-Difluoroethene
1634-04-4, Methyl tert-butyl ether
                                     1634-09-9, 1-Butylnaphthalene
1638-26-2, 1,1-Dimethylcyclopentane
                                      1639-09-4, 1-Heptanethiol
1640-89-7, Ethylcyclopentane 1678-91-7, Ethylcyclohexane 1678-92-8, Propylcyclohexane 1678-93-9, Butylcyclohexane 1679-07-8,
                    1679-09-0, 2-Methyl-2-butanethiol 1712-64-7, 1741-83-9, Methylpentyl sulfide 1759-58-6
Cyclopentanethiol
Isopropyl nitrate
1759-81-5, 4-Methylcyclopentene 1795-15-9, 1-Cyclohexyloctane
           1795-17-1, 1-Cyclohexyldodecane 1795-18-2,
1795-16-0
1-Cyclohexyltetradecane 1795-20-6 1795-21-7 1795-22-8
                                                               1795-26-2
1795-27-3
RL: PRP (Properties)
   (heat capacity of, equation for calcn. of)
2004-70-8 2027-19-2, 2-Propylnaphthalene 2040-95-1, Butylcyclopentane
2040-96-2, Propylcyclopentane 2051-30-1, 2,6-Dimethyloctane
                                                                   2079-95-0,
1-Tetradecanethiol 2131-18-2 2189-60-8, 1-Phenyloctane 2207-01-4
2207-03-6
            2207-04-7 2213-23-2, 2,4-Dimethylheptane
                                                          2216-30-0,
2,5-Dimethylheptane 2216-32-2, 4-Ethylheptane 2216-33-3,
                2216-34-4, 4-Methyloctane
3-Methyloctane
                                              2243-98-3, 1-Undecyne
2437-56-1, 1-Tridecene 2532-58-3
                                     2613-61-8, 2,4,6-Trimethylheptane
2690-08-6
           2765-18-6, 1-Propylnaphthalene 2851-83-4
1-Cyclopentylnonane 2883-02-5 2885-00-9, 1-Octadecanethiol
2917-26-2, 1-Hexadecanethiol
                               3074-71-3, 2,3-Dimethylheptane
3074-75-7, 4-Ethyl-2-methylhexane
                                    3074-76-8, 3-Ethyl-3-methylhexane
                                    3129-90-6, Isothiocyanic acid
3074-77-9, 3-Ethyl-4-methylhexane
3178-29-8, 4-Propylheptane 3221-61-2, 2-Methyloctane 3452-07-1,
1-Eicosene 3452-09-3, 1-Nonyne
                                   3522-94-9, 2,2,5-Trimethylhexane
                         3698-94-0, Ethyloctyl sulfide 3698-95-1,
           3698-93-9
3698-89-3
                     3741-00-2
                                  3877-15-4, Methyl propyl sulfide
Methyloctyl sulfide
4032-86-4, 3,3-Dimethylheptane 4032-92-2, 2,4,4-Trimethylheptane
4032-93-3, 2,3,6-Trimethylheptane 4032-94-4, 2,4-Dimethyloctane
           4110-44-5, 3,3-Dimethyloctane 4110-50-3, Ethylpropyl sulfide
4050-45-7
            4292-92-6, Pentylcyclohexane
4292-75-5
                                             4457-00-5
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           4753-80-4, Thiacycloheptane
4669-01-6
                                          5171-84-6,
3,3,4,4-Tetramethylhexane
                            5332-52-5, 1-Undecanethiol
                                                           5408-86-6,
                                              5634-30-0
2,3-Dibromobutane 5617-41-4
                                 5617-42-5
                                                           5881-17-4,
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               5911-04-6, 3-Methylnonane 6006-33-3
3-Ethyloctane
6006-95-7 6163-66-2, tert-Butyl ether 6294-31-1, Hexyl sulfide 6742-54-7, 1-Phenylundecane 6765-39-5, 1-Heptadecene 6785-23-5
6812-38-0 6812-39-1 6863-58-7, sec-Butyl ether 6876-18-2 67146-60-3, 2,3-Dimethyloctane 7154-79-2 7154-80-5 7220-26-0,
                                                                   6876-23-9
3-Ethyl-2,4-dimethylhexane 7289-44-3
                                           7289-45-4, Methyltetradecyl
sulfide 7309-44-6, Ethylhexyl sulfide 7372-86-3,
2-Ethyl-6-methylnaphthalene 7642-09-3 7688-21-3
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10496-16-9
               10496-18-1 13269-52-8
                                        13360-61-7, 1-Pentadecene
    13373-97-2, 1-Eicosanethiol 13475-78-0, 5-Ethyl-2-methylheptane
    13475-79-1 13475-81-5, 2,2,3,3-Tetramethylhexane 13952-84-6,
                    14676-29-0, 3-Ethyl-2-methylheptane
    sec-Butylamine
                                                        14720-74-2,
    2,2,4-Trimethylheptane 15869-80-4, 3-Ethylheptane 15869-85-9,
    5-Methylnonane 15869-86-0, 4-Ethyloctane 15869-87-1,
    2,2-Dimethyloctane
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    3,4-Dimethyloctane 15869-93-9, 3,5-Dimethyloctane
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    3,6-Dimethyloctane 15869-95-1, 4,4-Dimethyloctane 15869-96-2,
    4,5-Dimethyloctane 16747-25-4, 2,2,3-Trimethylhexane 16747-26-5,
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    2,4,4-Trimethylhexane 16747-31-2, 3,3,4-Trimethylhexane
                                                            16747-32-3,
    3-Ethyl-2,2-dimethylpentane 16747-33-4, 3-Ethyl-2,3-dimethylpentane
    16747-38-9, 2,3,3,4-Tetramethylpentane 16747-42-5,
    2,2,4,5-Tetramethylhexane 16747-44-7, 2,2,3,3,4-Pentamethylpentane
    16747-45-8, 2,2,3,4,4-Pentamethylpentane 16789-46-1,
    3-Ethyl-2-methylhexane 16900-07-5, Butyloctyl sulfide
                                                            16900-08-6,
                         16967-04-7, Butylhexyl sulfide 17059-55-1
    Butyldodecyl sulfide
    17301-94-9, 4-Methylnonane 17302-01-1, 3-Ethyl-3-methylheptane
    17302-02-2 17302-04-4
                           17348-59-3, Isopropyl tert-butyl ether
    18435-45-5, 1-Nonadecene
                             18437-89-3 19313-57-6
                                                       19313-61-2,
    Decylethyl sulfide 19398-77-7, 3,4-Diethylhexane
                                                      19484-26-5,
    1-Tridecanethiol 20278-84-6, 2,4,5-Trimethylheptane 20278-85-7,
    2,3,5-Trimethylheptane 20278-87-9, 3,3,4-Trimethylheptane
                                                                20278-88-0,
    3,4,4-Trimethylheptane 20291-60-5, Hexylmethyl sulfide 20291-61-6,
    Heptylmethyl sulfide 20291-91-2 20291-95-6, 2,2,5-Trimethylheptane
    22438-39-7, Decylmethyl sulfide 24768-42-1, Butylpentyl sulfide
    24768-43-2
                 24768-44-3, Ethylheptyl sulfide 24768-46-5, Heptylpropyl
    sulfide 25276-70-4, 1-Pentadecanethiol 26158-99-6, Ethylpentyl sulfide
    26186-00-5, 1-Heptadecyne 26186-01-6, 1-Nonadecyne 26186-02-7,
    1-Tridecyne 27563-68-4, Hexadecylmethyl sulfide 31032-94-7
    36653-82-4, 1-Hexadecanol 38842-05-6, 1,2,3,5-Tetraethylbenzene
    40289-98-3 40813-84-1 41947-84-6 42205-08-3
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    Pentylpropyl sulfide 51750-65-3, 2,2,4,4-Tetramethylhexane
                                                                 52896-87-4,
    4-Isopropylheptane 52896-88-5
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    3-Ethyl-4-methylheptane 52896-92-1, 2,2,3-Trimethylheptane
                                                                 52896-93-2,
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    2,3,3-Trimethylheptane
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    52897-03-7
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    2,2,3,4-Tetramethylhexane
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    3-Ethyl-2,2,3-trimethylpentane 52897-18-4
                                               52897-19-5,
    3-Ethyl-2,3,4-trimethylpentane 53161-72-1 53193-22-9,
                        53193-23-0, 1-Nonadecanethiol 54105-66-7
    1-Heptadecanethiol
    59973-07-8, Methylnonyl sulfide 59973-08-9 62103-66-6 62155-09-3
    62155-10-6, Methylpentadecyl sulfide
                                        62155-11-7
                                                     62155-12-8
                66271-54-3
                            66271-55-4
                                        66271-81-6
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    64919-20-6
                 66292-31-7, Ethylhexadecyl sulfide 66292-32-8 66292-33-9
    66271-83-8
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    RL: PRP (Properties)
        (heat capacity of, equation for calcn. of)
L23 ANSWER 8 OF 10 CAPLUS COPYRIGHT 2009 ACS on STN
    1976:121128 CAPLUS
    84:121128
OREF 84:19661a,19664a
    Entered STN: 12 May 1984
    O,S'-Dialkyl-S-hydrocarbylthioalkyl dithiophosphates
    Oswald, Alexis A.; Valint, Paul L., Jr.
    Exxon Research and Engineering Co., USA
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CODEN: USXXAM
          Patent
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       English
          C07F; A01N
IC
INCL 260948000
            23-8 (Aliphatic Compounds)
            Section cross-reference(s): 5
FAN.CNT 4
           PATENT NO.
                                                         KIND DATE
                                                                                                      APPLICATION NO.
                                                         ____
        US 3927148
                                                          A
                                                                        19751216 US 1973-377874
                                                                                                                                                              19730709
PRAI US 1969-821117
                                                          A1
                                                                            19690501
                                                          A3
           US 1971-173267
                                                                            19710819
CLASS
  PATENT NO. CLASS PATENT FAMILY CLASSIFICATION CODES
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  US 3927148
                                                        C07F; A01N
                                        INCL
                                                         260948000
                                                        C07F0009-165 [ICM]; C07F0009-00 [ICM,C*]; A01N0009-36
                                        IPCI
                                                         [ICS]
                                         IPCR
                                                         A01N0057-00 [I,C*]; A01N0057-12 [I,A]; C07F0009-00
                                                         [I,C*]; C07F0009-165 [I,A]
                                                          558/183.000; 558/184.000; 558/187.000; 987/209.000
                                         ECLA
                                                         A01N057/12; C07F009/165A1+M
            Five (RO) 2P(S) S(CH2) nCH(SR1) R2 (R = Et, hexadecyl; n = 0, 1, 3; R1 = Et, hexadecyl; n = 0, 1, 3; R1 = Et, hexadecyl; n = 0, 1, 3; R1 = Et, hexadecyl; n = 0, 1, 3; R1 = Et, hexadecyl; n = 0, 1, 3; R1 = Et, hexadecyl; n = 0, 1, 3; R1 = Et, hexadecyl; n = 0, 1, 3; R1 = Et, hexadecyl; n = 0, 1, 3; R1 = Et, hexadecyl; n = 0, 1, 3; R1 = Et, hexadecyl; n = 0, 1, 3; R1 = Et, hexadecyl; n = 0, 1, 3; R1 = Et, hexadecyl; n = 0, 1, 3; R1 = Et, hexadecyl; n = 0, 1, 3; R1 = Et, hexadecyl; n = 0, 1, 3; R1 = Et, hexadecyl; n = 0, 1, 3; R1 = Et, hexadecyl; n = 0, 1, 3; R1 = Et, hexadecyl; n = 0, 1, 3; R1 = Et, hexadecyl; n = 0, 1, 3; R1 = Et, hexadecyl; n = 0, 1, 3; R1 = Et, hexadecyl; n = 0, 1, 3; R1 = Et, hexadecyl; n = 0, 1, 3; R1 = Et, hexadecyl; n = 0, 1, 3; R1 = Et, hexadecyl; n = 0, 1, 3; R1 = Et, hexadecyl; n = 0, 1, 3; R1 = Et, hexadecyl; n = 0, 1, 3; R1 = Et, hexadecyl; n = 0, 1, 3; R1 = Et, hexadecyl; n = 0, 1, 3; R1 = Et, hexadecyl; n = 0, 1, 3; R1 = Et, hexadecyl; n = 0, 1, 3; R1 = Et, hexadecyl; n = 0, 1, 3; R1 = Et, hexadecyl; n = 0, 1, 3; R1 = Et, hexadecyl; n = 0, 1, 3; R1 = Et, hexadecyl; n = 0, 1, 3; R1 = Et, hexadecyl; n = 0, 1, 3; R1 = Et, hexadecyl; n = 0, 1, 3; R1 = Et, hexadecyl; n = 0, 1, 3; R1 = Et, hexadecyl; n = 0, 1, 3; R1 = Et, hexadecyl; n = 0, 1, 3; R1 = Et, hexadecyl; n = 0, 1, 3; R1 = Et, hexadecyl; n = 0, 1, 3; R1 = Et, hexadecyl; n = 0, 1, 3; R1 = Et, hexadecyl; n = 0, 1, 3; R1 = Et, hexadecyl; n = 0, 1, 3; R1 = Et, hexadecyl; n = 0, 1, 3; R1 = Et, hexadecyl; n = 0, 1, 3; R1 = Et, hexadecyl; n = 0, 1, 3; R1 = Et, hexadecyl; n = 0, 1, 3; R1 = Et, hexadecyl; n = 0, 1, 3; R1 = Et, hexadecyl; n = 0, 1, 3; R1 = Et, hexadecyl; n = 0, 1, 3; R1 = Et, hexadecyl; n = 0, 1, 3; R1 = Et, hexadecyl; n = 0, 1, 3; R1 = Et, hexadecyl; n = 0, 1, 3; R1 = Et, hexadecyl; n = 0, 1, 3; R1 = Et, hexadecyl; n = 0, 1, 3; R1 = Et, hexadecyl; n = 0, 1, 3; R1 = Et, hexadecyl; n = 0, 1, 3; R1 = Et, hexadecyl; n = 0, 1, 3; R1 = Et, hexadecyl; n = 0, 1, 3; R1 = Et, hexadecyl; n = 0, 1, 3; R1 = Et, hexadecyl; n = 0, 
AB
            4-chlorophenyl, octyl; R2 = Me, H) were O-dealkylated and S-alkylated with
            R3Br (R3 = Pr, Et, dodecyl) to give the resp. RO(R3S)P(O)S(CH2)nCH(SR1)R2
            (I). The addition reaction of 13 RO(R1S)P(O)S(CH2)nCH:CHR2 (R = Et, Me, Pr;
            R1 = Pr, CH2CHMe2, Bu, CHMe2; n = 0, 1; R2 = Me, Et, CMe3) with R3SH (R3 = R1 = Pr, R3 = Pr, R3 = Pr, R3 = Pr, R3 
            Me, Et, CHMe2, Pr, hexyl) gave the resp. RO(R1S)P(O)S(CH2)nCH2CH(SR3)R2
            (II). The I and II demonstrated pesticidal activity.
ST
            alkylthioalkyl dialkyl dithiophosphate pesticide; alkenyl dithiophosphate
            addn alkanethiol
            Pesticides
ΙT
                   (O, S-dialkyl S-alkylthioalkyl dithiophosphates)
ΙT
            Dealkylation
                   (O-, of O,O-dialkyl S-alkylthioalkyl dithiophosphates, S-alkylation of
                   products from)
TT
            Alkylation
                   (S-, of O-alkyl-S-(alkylthioalkyl)dithiophosphoric acids with alkyl
                   bromides)
ΙT
            786-19-6 22911-14-4
                                                                   57342-34-4
            RL: RCT (Reactant); RACT (Reactant or reagent)
                   (O-Dealkylation and S-alkylation of)
            298-02-2 17346-57-5 57342-43-5
ΤТ
            RL: RCT (Reactant); RACT (Reactant or reagent)
                    (O-dealkylation and S-alkylation of)
            74-96-4 106-94-5 143-15-7
ΤТ
            RL: RCT (Reactant); RACT (Reactant or reagent)
                    (S-alkylation of O-alkyl-S-(alkylthioalkyl)dithiophosphoric acids with)
ΙT
            111-85-3 4860-03-1
            RL: RCT (Reactant); RACT (Reactant or reagent)
                    (S-alkylation of O-alkyl-S-alkenyldithiophosphoric acid derivative with)
                                     27941-98-6
TΤ
            2917-26-2
            RL: RCT (Reactant); RACT (Reactant or reagent)
                   (addition reaction of, with O,S-dialkyl S-alkenyl dithiophosphate
derivative)
                                                                                107-03-9 111-31-9
                                 75-08-1 75-33-2
           74-93-1
            RL: RCT (Reactant); RACT (Reactant or reagent)
                    (addition reaction of, with O,S-dialkyl S-alkenyl dithiophosphates)
            27564-69-8 27564-71-2 27564-72-3 27564-73-4 27564-77-8
ΤТ
```

SO

U.S., 15 pp.

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57342-37-7 57342-38-8 57342-39-9 57342-40-2
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (addition reaction of, with alkanethiols)
     57342-36-6P
IΤ
     RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT
     (Reactant or reagent)
        (preparation and S-alkylation of, with dodecyl bromide)
ΙT
     57342-45-7P
     RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT
     (Reactant or reagent)
        (preparation and S-alkylation of, with hexadecyl chloride)
TT
     57342-47-9P
     RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT
     (Reactant or reagent)
        (preparation and S-alkylation of, with octyl chloride)
     57342-42-4P
ΤT
     RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT
     (Reactant or reagent)
        (preparation and addition reaction of, with hexadecanethiol)
     57342-41-3P
ΤT
     RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT
     (Reactant or reagent)
        (preparation and addition reaction of, with trichlorobenzenethiol)
ΙT
     32039-91-1P 32064-96-3P 32064-97-4P 32064-98-5P 32064-99-6P
                                32065-02-4P 32065-03-5P 32065-04-6P 32065-07-9P 32065-08-0P 32065-10-4P 57342-32-2P 57342-33-3P 58588-84-4P
                 32065-01-3P
     32065-00-2P
     32065-05-7P 32065-06-8P
57341-50-1P 57342-31-1P
     RL: BAC (Biological activity or effector, except adverse); BSU (Biological
     study, unclassified); SPN (Synthetic preparation); BIOL (Biological
     study); PREP (Preparation)
        (preparation and pesticidal activity of)
L23 ANSWER 9 OF 10 CAPLUS COPYRIGHT 2009 ACS on STN
    1975:592556 CAPLUS
AN
    83:192556
DN
OREF 83:30261a,30264a
ED
   Entered STN: 12 May 1984
   Pesticidal O,S'-dialkyl S-alkylthioalkyl dithiophosphates
TI
   Oswald, Alexis A.; Valint, Paul L., Jr.
ΤN
     Exxon Research and Engineering Co., USA
PΑ
SO
    U.S., 16 pp.
     CODEN: USXXAM
DT
    Patent
LA
    English
    C07F; A01N
IC
INCL 260949000
     23-8 (Aliphatic Compounds)
CC
     Section cross-reference(s): 5
FAN.CNT 4
                        KIND
                                           APPLICATION NO. DATE
     PATENT NO.
                               DATE
     _____
                         ____
                                _____
                                             _____
                                            US 1973-377872 19730709
                         A
PI US 3904710 A
PRAI US 1969-821117 A1
US 1971-173267 A3
PΙ
   US 3904710
                                19750909
                                19690501
                                19710819
CLASS
             CLASS PATENT FAMILY CLASSIFICATION CODES
 PATENT NO.
 _____
                 IC
 US 3904710
                        C07F; A01N
                 INCL
                        260949000
                        C07F0009-165 [ICM]; C07F0009-00 [ICM,C*]; A01N0009-36
                 IPCI
                        [ICS]
                 IPCR
                        A01N0057-00 [I,C*]; A01N0057-14 [I,A]; C07F0009-00
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[I,C*]; C07F0009-165 [I,A]
                        558/187.000; 558/183.000; 558/184.000; 987/209.000
                 NCL
                       A01N057/14; C07F009/165A1+M
                 ECLA
     Esters RSQSP(S)(OR1)2 (R = Et, octyl, 4-ClC6H4; Q = Cl-4 straight-chain or
AΒ
     branched alkylene; R1 = Et, hexadecyl) were O-dealkylated with amine
     catalysts and the products were S-alkylated with R2Br to give five
     RSQSP(0)(OR1)SR2 (R2 = Et, Pr, dodecyl) which exhibited pesticidal,
     insecticidal, and miticidal activity. Twelve RCH(SR3)CH2SP(0)(OR1)SR2 (R
     = C1-4 alkyl, R1 = C1-8 alkyl, R2 = C3-16 alkyl, R3 = C1-6 alkyl), which
     also demonstrated the above properties, were prepared from
     RCH: CHSP(O) (OR1) SR2 and R3SH.
     alkyl dithiophosphate pesticide insecticide; miticide alkyl
     dithiophosphate; alkylthioalkyl dithiophosphate pesticide miticide;
     dealkylation catalytic trialkyl dithiophosphate; alkylation dialkyl
     dithiophosphate sulfur
    Acaricides
ΤT
     Insecticides
     Pesticides
        (O, S-dialkyl S-alkylthioalkyl dithiophosphates)
ΙT
     Dealkylation catalysts
        (O-, amines, for O,O-dialkyl S-alkylthioalkyl dithiophosphates)
ΙT
     Alkylation
        (S-, of O-alkyl S-alkylthioalkyl dithiophosphates with alkyl bromides)
ΙT
     Addition reaction
        (of O, S-dialkyl S-alkenyl dithiophosphates with alkanethiols)
     298-04-4
                786-19-6 17346-57-5 22911-14-4 57342-34-4 57342-43-5
ΙT
     57583-99-0
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (O-dealkylation of, catalysts for)
ΙT
     111-85-3
                4860-03-1
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (S-alkylation of O-alkyl S-alkenyl dithiophosphates with)
     74-96-4 106-94-5
                      143-15-7
ΤT
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (S-alkylation of O-alkyl S-alkylthioalkyl dithiophosphates with)
ΤТ
     74-93-1
               75-08-1
                        75-33-2 107-03-9
                                             111-31-9 2917-26-2
     27941-98-6
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (addition reaction of, with O,S-dialkyl S-alkenyl dithiophosphates)
ΙT
     27564-69-8
                  27564-71-2
                               27564-72-3
                                            27564-73-4 27564-77-8
                  57342-38-8
     57342-37-7
                               57342-39-9
                                            57342-40-2
                                                         57342-41-3
     57342-42-4
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (addition reaction of, with alkanethiols)
     75-50-3, uses and miscellaneous
                                       280-57-9 7664-41-7, uses and
ΤT
     miscellaneous
     RL: CAT (Catalyst use); USES (Uses)
        (catalyst, for O-dealkylation of O,O-dialkyl S-alkylthioalkyl
        dithiophosphates)
ΙT
     57342-36-6P
                   57342-45-7P
                                 57342-47-9P
     RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT
     (Reactant or reagent)
        (preparation and S-alkylation of)
ΙT
     32039-91-1P
                   32064-96-3P
                                 32064-97-4P
                                               32064-99-6P
                                                             32065-00-2P
     32065-01-3P
                   32065-02-4P
                                 32065-03-5P
                                                             32065-05-7P
                                               32065-04-6P
     32065-06-8P
                   32065-07-9P
                                 32065-08-0P
                                                             32065-10-4P
                                               32065-09-1P
     57341-50-1P
                   57342-31-1P
                                 57342-32-2P
                                               57342-33-3P
                                                             57517-30-3P
     RL: BAC (Biological activity or effector, except adverse); BSU (Biological
     study, unclassified); SPN (Synthetic preparation); BIOL (Biological
     study); PREP (Preparation)
        (preparation and pesticidal activity of)
```

```
L23 ANSWER 10 OF 10 CAPLUS COPYRIGHT 2009 ACS on STN
AN
     1961:12985 CAPLUS
DN
     55:12985
OREF 55:2478g-i,2479a-g
     Entered STN: 22 Apr 2001
ED
ΤI
     S-Alkylmercaptosuccinic acids as solid derivatives of olefins, alkyl
     bromides, and mercaptans
ΑU
     Hendrickson, Joe G.; Hatch, Lewis F.
CS
     Univ. of Texas, Austin
SO
     Journal of Organic Chemistry (1960), 25, 1747-52
     CODEN: JOCEAH; ISSN: 0022-3263
DT
     Journal
LA
     Unavailable
CC
     10B (Organic Chemistry: Aliphatic Compounds)
AΒ
     Solid S-alkylmercaptosuccinic acids were prepared from olefins, mercaptans,
     and alkyl bromides and their m.ps. and solubilities studied as a function
     of structure of the alkyl group. These properties varied with structure
     in a predictable manner. The acids were satisfactory solid derivs. for
     primary olefins and mercaptans and both primary and secondary alkyl
     bromides because of the ease with which the reaction could be effected,
     the good yields obtained, and the ease of purification. They had the
     added advantage of being acids; thus their neutralization equivs. could be
     obtained for confirmatory characterization. Mercaptosuccinic acid (I) (2
     g.) and 3 ml. MeOH heated until the acid had completely dissolved, the
     solution cooled, 1 ml. olefin plus 0.10 g. Bz202 added, the tube stoppered,
     shaken 5 min., left at room temperature, the crystals washed with H2O, and
     treated with 25 ml. 6N HCl gave 1.3-1.9 g. products. The crystals were
     recovered by vacuum filtration, dried 12 hrs. at room temperature, 1 g. of the
     derivative dissolved in 10-15 ml. Et20, then pentane added, the mixture
     filtered, the crystals discarded, pentane added to the filtrate, and the
     crystals separated Di-Na maleate (20 ml., 1.0M), 2 ml. alc., 1 ml. mercaptan,
     and chips were refluxed 2-4 hrs., the mixture cooled, the lower layer
separated,
     diluted with 10 ml. concentrated HCl, the mercaptan derivative precipitated,
and purified in
     the same manner as the products from the I-olefin reaction. The yield
     usually was in the range 0.8-2.0 g. I (1.00 ml.), 2 ml. PrOH, 1 ml. alkyl
     bromide, 25 ml. 1.33N KOH, and chips were refluxed 4-24 hrs., the aqueous
     layer extracted with pentane, 10 ml. concentrated HCl added to the aqueous
laver, and
     the precipitated material recrystd. as usual. The S-alkylmercaptosuccinic
acids
     were titrated with 0.07N KOH to phenolphthalein end point in the presence
     of 5 ml. EtOH and 40 ml. H2O; the higher mol. weight derivs. were titrated in
     warm solution because of their limited solubility The following results were
thus
     obtained (R of RSCH(CO2H)CH2CO2H, olefin, % yield of olefin, m.p., % yield
     of mercaptan, m.p., % yield of bromide and m.p. given): Pr, -, -, -, -, -,
     78, 118.4-18.8°; 1-Bu, -, -, -, 43, 103.7-4.0°, -, -; 2-Bu,
     -, -, -, -, -, 60, 134.9-5.1°; iso-Bu, -, -, -, -, 41,
     120.9-1.4; tert-Bu, -, -, -, -, -, -, -; 1-pentyl, 1-pentene, 85, 107.3-7.6°, 100, 107.7-8.0°, 60, 107.0-7.6°;
     2-pentyl, -, -, -, -, 50, 134.8-5.4°; 3-pentyl, -, -, -, -, 39, 153.8-4.1°; 2-methylbutyl, 2-methyl-1-butene, 85,
     122.3-2.6°, -, -, -, -; isopentyl, -, -, -, -, -, 65, 115.6-16.0°; 1,2-dimethylpropyl, 2-methyl-2-butene, 75,
     153.7-4.0°, -, -, -, -; 1-hexyl, 1-hexene, 100, 95.4-5.7°, 79, 96.0-6.2°, 91, 96.3-6.5%; 2-hexyl, -, -, -, -, 31, 123.9-5.0°; 3-hexyl, -, -, -, -, 24, 143.4-3.5°;
     3-methylpentyl, -, -, -, 84, 111.9-12.3°; 4-methylpentyl,
     4-methyl-1-pentene, 92, 102.6-2.9°, -, -, -, -; 1-isopropylpropyl,
     2-methyl-2-pentene, -, -, 60, 152.1-2.6°, -, -; 2-ethylbutyl, -, -,
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-, -, -, 75, 132.4-2.8°; 1-heptyl, 1-heptene, 88,
103.4-3.9°, 82, 105.8-6.2°, -, -; 2-heptyl, -, -, -, -, -, 31, 128.0-9.1°; 3-heptyl, -, -, -, -, 21, 144.9-5.4°;
4,4-di-methylpentyl, 4,4-dimethyl-1-pentene, 90, 119.0-19.5°, -, -,
-, -; 1-methyl-2-ethylbutyl, 3-ethyl-2-pentene, 45, 148.9-9.9°, -,
-, -, -; 1-octyl, 1-octene, 96, 96.1-6.6°, -, -, -, -; 2-octyl, -,
-, -, -, 50, 128.0-9.0°; 3-octyl, 2-octene, 94,
142.9-3.5°, -, -, -, 2-ethyl-hexyl, 2-ethyl-1-hexene, 81, 101.9-2.7°, -, -, 59, 103.7-4.2°; 2-cyclohexylethyl, -, -,
-, -, -, 47, 126.3-6.8°; 1-nonyl, -, -, -, 85, 105.0-6.0°,
43, 105.0-5.5°; 3-phenylpropyl, -, -, -, 98, 114.9-15.7°, -,
-; 1-decyl, 1-decene, 82, 93.5-3.8°, -, -, -, -; 1-undecyl, -, -,
-, -, -, 37, 110.4-10.6°; 1-tetradecyl, 1-tetradecene, 76,
104.0-4.8°, -, -, -, 1-hexadecyl, 1-hexadecene, 84,
105.0-5.8°, -, -, -; 1-octadecyl, -, -, -, 100, 102-3°, -, -; cyclopentyl, cyclopentene, 75, 142.8-3.1°, -, -, -;
cyclohexyl, cyclohexene, 83, 150.5-1.1°, -, -, 9,
149.0-9.9°; 2-methylcyclohexyl, 2-methylcyclohexene, 73,
187.0-7.3^{\circ}, -, -, -, 4-methylcyclohexyl, -, -, 6,
148.7-9.3°, -, -; benzyl, -, -, -, 97, 188.4-8.9, -, -; p-chlorophenyl, -, -, -, -, -, -, -; 3-chloropropyl, allyl chloride, 35, 108.5-8.8°, -, -, -, -. The
following solubilities of RSCH(CO2H)CH2CO2H in PhMe at 50.0° were
obtained (R and g./100 g. PhMe given): 1-Pr, 0.30; 2-Bu, 0.46; iso-Bu,
0.87; 2-methylbutyl, 4.00; 2-methyl-2-butyl, 0.14; 2-ethylbutyl, 2.59;
octadecyl, 3.70; cyclohexyl, 0.25; 2-cyclohexylethyl, 1.90;
3-chloropropyl, 0.67.
Alkyl bromides
Olefins
   (identification of)
Thiols
   (separation and determination of)
110-83-8, Cyclohexene
   (detection of)
78-76-2, Butane, 2-bromo-
                              78-77-3, Propane, 1-bromo-2-methyl-
100-53-8, \alpha-Toluenethiol 106-94-5, Propane, 1-bromo-
107-05-1, Propene, 3-chloro- 107-81-3, Pentane, 2-bromo-
                                                                   107-82-4,
Butane, 1-bromo-3-methyl- 108-85-0, Cyclohexane, bromo- 109-67-1,
1-Pentene 109-79-5, 1-Butanethiol 110-53-2, Pentane, 1-bromo-
110-66-7, 1-Pentanethiol 111-25-1, Hexane, 1-bromo- 111-31-9,
                 111-66-0, 1-Octene 111-67-1, 2-Octene 142-29-0,
1-Hexanethiol
Cyclopentene
                 513-35-9, 2-Butene, 2-methyl- 557-35-7, Octane, 2-bromo-
563-46-2, 1-Butene, 2-methyl- 591-49-1, Cyclohexene, 1-methyl-
592-41-6, 1-Hexene 592-76-7, 1-Heptene 625-27-4, 2-Pentene, 2-methyl-
629-73-2, 1-Hexadecene 691-37-2, 1-Pentene, 4-methyl- 693-58-3,
Nonane, 1-bromo- 693-67-4, Undecane, 1-bromo- 762-62-9, 1-Pentene,
4,4-dimethyl- 816-79-5, 2-Pentene, 3-ethyl- 872-05-9, 1-Decene
1120-36-1, 1-Tetradecene 1455-21-6, 1-Nonanethiol
                                                           1632-16-2, 1-Hexene,
2-ethyl- 1639-09-4, 1-Heptanethiol 1647-26-3, Cyclohexane, (2-bromoethyl)- 1809-10-5, Pentane, 3-bromo- 1974-04-5, Heptane,
2-bromo- 1974-05-6, Heptane, 3-bromo- 2885-00-9, 1-Octadecanethiol 3377-86-4, Hexane, 2-bromo-
                                                      3377-87-5, Hexane,
18908-66-2, Heptane,
3-bromo- 3814-34-4, Pentane, 3-(bromomethyl)-
3-(bromomethyl)- 24734-68-7, 1-Propanethiol, 3-phenyl- 51116-73-5, Pentane, 1-bromo-3-methyl- 60260-87-9, Cyclohexanethiol, 4-methyl-
   (identification of)
5413-66-1P, Succinic acid, (pentylthio) - 6188-77-8P, Succinic acid,
(octylthio) - 22119-10-4P, Succinic acid, (benzylthio) - 26819-75-0P,
Succinic acid, (isobutylthio) - 26819-76-1P, Succinic acid,
(isopentylthio) - 45015-91-6P, Succinic acid, (propylthio) -
45084-17-1P, Succinic acid, (butylthio) - 60713-01-1P, Succinic acid,
(tetradecylthio) - 60713-02-2P, Succinic acid, (hexadecylthio) -
```

ΤТ

ΤТ

ΙT

60745-27-9P, Succinic acid, (decylthio) - 65594-35-6P, Succinic acid, (hexylthio) - 85927-34-0P, Succinic acid, (octadecylthio) -98431-24-4P, Succinic acid, (3-chloropropylthio) - 99174-55-7P, Succinic acid, (1-ethyl-2-methylpropylthio) - 99183-70-7P, Succinic acid, (1,2-dimethylpropylthio) - 99183-71-8P, Succinic acid, (1-ethylpropylthio) - 99974-58-0P, Succinic acid, (cyclohexylthio) -100048-63-3P, Succinic acid, (4,4-dimethylpentylthio) - 100048-64-4P, Succinic acid, (2-ethyl-1-methylbutylthio) - 100048-65-5P, Succinic acid, (1-ethylpentylthio) - 100048-66-6P, Succinic acid, (heptylthio) -100048-67-7P, Succinic acid, (1-methylhexylthio) - 100145-30-0P, Succinic acid, (cyclopentylthio) - 100250-93-9P, Succinic acid, (1-methylheptylthio) - 100315-91-1P, Succinic acid, (2-cyclohexylethylthio) - 100538-68-9P, Succinic acid, (nonylthio) -100613-26-1P, Succinic acid, (3-phenylpropylthio) - 103204-54-2P, Succinic acid, [2-ethylbutylthio] - 103204-55-3P, Succinic acid, [3-methylpentylthio] - 103205-84-1P, Succinic acid, [1-ethylbutylthio]-103205-85-2P, Succinic acid, [1-methylpentylthio] - 103260-48-6P, Succinic acid, [2-methylbutylthio] - 103263-72-5P, Succinic acid, [1-methylbutylthio] - 104177-65-3P, Succinic acid, [4-methylcyclohexylthio]-104178-83-8P, Succinic acid, [2-methylcyclohexylthio] - 105906-88-5P, Succinic acid, [2-ethylhexylthio] - 105910-67-6P, Succinic acid, [1-ethylhexylthio] -114098-60-1P, Succinic acid, (isohexylthio) - 120089-21-6P, Succinic acid, (sec-butylthio) -131731-38-9P, Succinic acid, undecylthio-RL: PREP (Preparation) (preparation of) 70-49-5, Succinic acid, mercapto-

 \Rightarrow s (106-94-5 or 2885-00-9 or 2917-26-2 or 22811-02-5 or 10220-46-9) and (silver or ag)

REG1stRY INITIATED

(S-alkyl derivs.)

ΤТ

Substance data SEARCH and crossover from CAS REGISTRY in progress... Use DISPLAY HITSTR (or FHITSTR) to directly view retrieved structures.

L25 50 L24

REG1stRY INITIATED
Substance data SEARCH and crossover from CAS REGISTRY in progress...
Use DISPLAY HITSTR (or FHITSTR) to directly view retrieved structures.

L27 6 L26

REG1stRY INITIATED
Substance data SEARCH and crossover from CAS REGISTRY in progress...
Use DISPLAY HITSTR (or FHITSTR) to directly view retrieved structures.

L29 1621 L28

REG1stRY INITIATED

Substance data SEARCH and crossover from CAS REGISTRY in progress... Use DISPLAY HITSTR (or FHITSTR) to directly view retrieved structures.

L31 2011 L30

REG1stRY INITIATED

Substance data SEARCH and crossover from CAS REGISTRY in progress... Use DISPLAY HITSTR (or FHITSTR) to directly view retrieved structures.

L33 4963 L32

379472 SILVER 342330 AG

L34 510 (L33 OR L31 OR L29 OR L27 OR L25) AND (SILVER OR AG)

=> s polish####### or tarnish#### or detarnish####

117992 POLISH#######

3049 TARNISH####

9 DETARNISH#####

L35 120792 POLISH####### OR TARNISH#### OR DETARNISH####

=> 134 and 135

L34 IS NOT A RECOGNIZED COMMAND

The previous command name entered was not recognized by the system. For a list of commands available to you in the current file, enter "HELP COMMANDS" at an arrow prompt (=>).

=> s 134 and 135

L36 26 L34 AND L35

=> d 1-26 all

L36 ANSWER 1 OF 26 CAPLUS COPYRIGHT 2009 ACS on STN

```
DN
     143:351549
ΕD
     Entered STN: 05 Oct 2005
ΤI
     Water-based sulfur-containing composition chemical mechanical
     polishing of nonferrous metals
IN
     Johns, Peter Gamon; Harrison, Clare Elizabeth
PA
     Middlesex Silver Co. Limited, UK
SO
     Brit. UK Pat. Appl., 29 pp.
     CODEN: BAXXDU
DT
     Patent
     English
LA
     ICM C23F011-16
IC
     ICS C23F011-00
CC
     57-7 (Ceramics)
     Section cross-reference(s): 56
FAN.CNT 1
                                               APPLICATION NO.
     PATENT NO.
                           KIND DATE
                                                                           DATE
                                    _____
                                                 _____
                           ____
     GB 2412666
GB 2412666
                                               GB 2004-7163
                            A
                                   20051005
                                                                            20040330
PΙ
     GB 2412666 B 20081008
AU 2005229275 A1 20051013
CA 2559989 A1 20051013
WO 2005095675 A1 20051013
                                                 AU 2005-229275
                                                                            20050324
                                                 CA 2005-2559989
WO 2005-GB50043
                                               CA 2005-2001
WO 2005-GB50043
                                                                            20050324
                                                                           20050324
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A 20070713 IN 2006-DN5356

A 20061116 MX 2006-10964

A1 20071206 US 2007-594477

A 20040330

W 2005022
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OS
    MARPAT 143:351549
    A composition and associated method of manufacture of a water based
AΒ
composition comprising a
     treatment agent selected from an alkanethiol, alkyl thioglycollate, and
     dialkyl sulfide or dialkyl disulfide. The composition also includes at least
     one of an amphoteric, non-ionic or cationic surfactant, where the
     treatment agent is directly dissolved or dispersed the water containing the
     amphoteric, non-ionic or cationic surfactant. The composition is particularly
     useful for the treatment of Ag-Cu-Ge alloy, copper, brass, and
     nickel. A solid polishing medium can also be included in the
     composition, for example, silica or precipitated chalk, alumina, or silica.
ST
     chalk alumina silica alkanethiol thioglycollate chem mech
     polishing copper
ΙT
     Thiols, processes
     RL: CPS (Chemical process); PEP (Physical, engineering or chemical
     process); TEM (Technical or engineered material use); PROC (Process); USES
     (Uses)
        (alkanethiol; water-based sulfur-containing composition chemical mech.
        polishing of metals)
ΙT
     Disulfides
     RL: CPS (Chemical process); PEP (Physical, engineering or chemical
     process); TEM (Technical or engineered material use); PROC (Process); USES
     (Uses)
        (alkyl; water-based sulfur-containing composition chemical mech. polishing
        of metals)
ΙT
     Chalk
     Diatomite
     RL: TEM (Technical or engineered material use); USES (Uses)
        (as abrasive; water-based sulfur-containing composition chemical mech.
        polishing of metals)
ΙT
     Surfactants
        (cationic; water-based sulfur-containing composition chemical mech.
        polishing of metals)
ΙT
     Polishing
        (chemical-mech.; water-based sulfur-containing composition chemical mech.
        polishing of metals)
ΤТ
     Polishing materials
        (paste; water-based sulfur-containing composition chemical mech. polishing
        of metals)
ΤТ
     Thioethers
     RL: CPS (Chemical process); PEP (Physical, engineering or chemical
     process); TEM (Technical or engineered material use); PROC (Process); USES
     (Uses)
        (water-based sulfur-containing composition chemical mech. polishing of
        metals)
ΤТ
     1344-28-1, Alumina, uses
     RL: TEM (Technical or engineered material use); USES (Uses)
        (abrasive; water-based sulfur-containing composition chemical mech.
        polishing of metals)
     9004-82-4, Sodium laureth sulfate
TT
     RL: MOA (Modifier or additive use); USES (Uses)
        (anionic surfactant; water-based sulfur-containing composition chemical
mech.
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polishing of metals)
ΤТ
     7631-86-9, Silica, uses
     RL: TEM (Technical or engineered material use); USES (Uses)
        (as abrasive; water-based sulfur-containing composition chemical mech.
        polishing of metals)
IT
     36574-66-0D, N-coco acyl derivs.
     RL: MOA (Modifier or additive use); USES (Uses)
        (cocamidopropyl betaine, surfactant; water-based sulfur-containing
composition
        chemical mech. polishing of metals)
     7440-02-0, Nickel, processes
                                    7440-50-8, Copper, processes
ΤТ
     12597-71-6, Brass, processes
                                    74969-69-0
     RL: PEP (Physical, engineering or chemical process); PYP (Physical
     process); PROC (Process)
        (polished substrate; water-based sulfur-containing composition chemical
        mech. polishing of metals)
ΙT
     62-56-6, Thiourea, uses 2885-00-9, Octadecyl mercaptan
     RL: TEM (Technical or engineered material use); USES (Uses)
        (polishing composition component; water-based sulfur-containing composition
        chemical mech. polishing of metals)
ΤT
     2917-26-2, Hexadecyl mercaptan
     RL: MOA (Modifier or additive use); USES (Uses)
        (surfactant; water-based sulfur-containing composition chemical mech.
        polishing of metals)
     68-11-1D, alkyl esters
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     (Uses)
        (water-based sulfur-containing composition chemical mech. polishing of
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RE.CNT
             THERE ARE 6 CITED REFERENCES AVAILABLE FOR THIS RECORD
(1) Anon; EP 0492487 A1 CAPLUS
(2) Anon; GB 0956927 A
(3) Anon; GB 1117510 A
(4) Anon; US 3503883 A
(5) Anon; US 3518098 A
(6) Anon; US 5650385 A CAPLUS
L36 ANSWER 2 OF 26 CAPLUS COPYRIGHT 2009 ACS on STN
ΑN
     2004:1087508 CAPLUS
DN
     142:489405
ED
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ΤТ
     Reduction of silver tarnishing and protection against
     subsequent corrosion
ΑU
     Bernard, M. C.; Dauvergne, E.; Evesque, M.; Keddam, M.; Takenouti, H.
     UPR 15 of CNRS "Laboratoire Interfaces et Systemes Electrochimiques",
CS
     Universite P&M Curie, Paris, 75252, Fr.
     Corrosion Science (2005), 47(3), 663-679
SO
     CODEN: CRRSAA; ISSN: 0010-938X
PΒ
     Elsevier Ltd.
DT
     Journal
LA
     English
     72-2 (Electrochemistry)
CC
     Section cross-reference(s): 28, 56, 66
     The kinetics of tarnishing formation was examined on a
     polished silver dipped in a 10 mM Na2S. The recovery of
     an initial bright surface was then obtained by cathodic reduction of the
     tarnish layer in a 5% sesqui-carbonate solution  Two protection
     methods to prevent a further formation of a dark deposit were tested: an
     electrodeposited poly(amino-triazole) film and the surface treatment in
     hexadecane-thiol. The protection by poly(amino-triazole) is not reliable
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for all nuances of silver. In contrast, the film formed with
     hexadecane-thiol showed satisfactory properties. The formations of
     tarnish and protective films were examined by electrochem. methods,
     the reflectance measurements, and the quartz crystal microbalance.
     silver tarnishing protection polyaminotriazole thiol
ST
     surface treatment electroredn; hexadecanethiol film silver
     tarnishing protection microbalance
ΙT
     Optical reflection
        (by silver during tarnishing reduction in sodium
        carbonate-bicarbonate solution)
     Electric potential
TΤ
        (during silver tarnishing in Na2S solution)
ΙT
     Polymerization
        (electrochem.; of 3-amino-1,2,4-triazole on silver for
        protection of tarnishing)
ΤТ
     Corrosion kinetics
        (kinetics of tarnishing formation on polished
        silver dipped in 10 mM Na2S)
TΤ
     Adsorption
        (of hexadecanethiol on silver surface for protection of
        tarnishing)
ΙT
     Cyclic voltammetry
        (of silver in sodium acetate methanol solution containing
        amino-triazole)
     Reduction, electrochemical
ΙT
        (of silver tarnishing and protection against
        subsequent corrosion)
ΙT
     Voltammetry
        (of silver tarnishing reduction in sodium
        carbonate-bicarbonate solution)
ΤТ
     Surface treatment
        (protection of silver tarnishing using
        hexadecane-thiol)
    Microbalances
ΙT
        (quartz crystal; study of silver tarnishing reduction
        in sodium carbonate-bicarbonate solution using)
ΙT
     Corrosion prevention
       Tarnishing
        (reduction of silver tarnishing and protection against
        subsequent corrosion)
ΙT
     67-56-1, Methanol, uses
                              127-09-3, Sodium acetate
     RL: NUU (Other use, unclassified); USES (Uses)
        (cyclic voltammetry of silver in sodium acetate methanol
        solution containing amino-triazole)
     61-82-5, 3-Amino-1,2,4-triazole
ΤT
     RL: CPS (Chemical process); PEP (Physical, engineering or chemical
     process); RCT (Reactant); PROC (Process); RACT (Reactant or reagent)
        (electropolymn. on silver for protection of
        tarnishing)
     1313-82-2, Sodium sulfide, reactions
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     RL: CPS (Chemical process); PEP (Physical, engineering or chemical
     process); RCT (Reactant); PROC (Process); RACT (Reactant or reagent)
        (kinetics of tarnishing formation on polished
        silver dipped in 10 mM Na2S)
     2917-26-2, Hexadecane-thiol
     RL: NUU (Other use, unclassified); USES (Uses)
        (protection of silver tarnishing using)
ΤТ
     7440-22-4, Silver, uses
     RL: CPS (Chemical process); DEV (Device component use); PEP (Physical,
     engineering or chemical process); RCT (Reactant); PROC (Process); RACT
     (Reactant or reagent); USES (Uses)
        (reduction of silver tarnishing and protection against
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subsequent corrosion)
ΙT
    151313-83-6P, Poly(3-amino-1,2,4-triazole)
    RL: NUU (Other use, unclassified); PNU (Preparation, unclassified); PREP
    (Preparation); USES (Uses)
       (silver protection of tarnishing using film of)
IT
    533-96-0, Sodium sesqui-carbonate
    RL: NUU (Other use, unclassified); USES (Uses)
       (voltammetry of silver tarnishing reduction in sodium
       carbonate-bicarbonate solution)
             THERE ARE 6 CITED REFERENCES AVAILABLE FOR THIS RECORD
RE
(1) Burleigh, T; Corrosion 2001, V57(12), P1066 CAPLUS
(2) Degrigny, C; Corrosion Australasia 1993, V18, P16 CAPLUS
(3) Degrigny, C; J Int Inst Conservat Historic Artistic Works 1995, P170
(4) Evesque, M; Electrochim Acta 2004, V49, P2939
(5) Keddam, M; 15th International Corrosion Congress, CD-Rom Proceedings 2002,
   701
(6) Trachli, B; Corros Sci 2002, V44, P997 CAPLUS
L36 ANSWER 3 OF 26 CAPLUS COPYRIGHT 2009 ACS on STN
ΑN
    2004:1051670 CAPLUS
DN
    142:26523
    Entered STN: 08 Dec 2004
ED
TΙ
    Silver-germanium-copper alloy for decorative utensils with a
    tarnish-preventing treatment
ΙN
    Johns, Peter Gamon
    Middlesex Silver Co. Limited, UK; Cole, Paul Gilbert
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SO
    Brit. UK Pat. Appl., 26 pp.
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IC
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20090115

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AT 2004-735594

IN 2005-DN5033

20040601

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AT 420980

IN 2005DN05033 A

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MX 2005012991 A 20060720 PRAI GB 2003-12693 A 20030603
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 JP 2007535616
                IPCI
                      C22C0005-06 [I,A]; C23C0022-02 [I,A]; C23G0005-02
                      [I,A]; C23G0005-00 [I,C*]; C22F0001-14 [N,A];
                      C22F0001-00 [N,A]
                IPCR
                      C22C0005-06 [I,C]; C22C0005-06 [I,A]; C22F0001-00
                      [N,C]; C22F0001-00 [N,A]; C22F0001-14 [N,C];
                      C22F0001-14 [N,A]; C23C0022-02 [I,C]; C23C0022-02
                      [I,A]; C23F0011-10 [I,C*]; C23F0011-16 [I,A];
                      C23G0005-00 [I,C]; C23G0005-02 [I,A]
                      C23F011/16; C23F011/16B
                FTERM 4K026/AA01; 4K026/BA01; 4K026/BB01; 4K026/BB08;
                      4K026/CA02; 4K053/PA01; 4K053/PA13; 4K053/QA07;
                      4K053/RA08; 4K053/RA54; 4K053/SA02; 4K053/SA06;
                      4K053/ZA01
                AT 420980
                IPCR C23F0011-10 [I,C*]; C23F0011-16 [I,A]
               ECLA C23F011/16; C23F011/16B
 IN 2005DN05033 IPCI C22C0005-06 [ICM, 7]
AΒ
    The decorative alloy contains 93.5-95.5% Aq, 0.5-3% Ge by weight,
    and Cu as the balance, optionally with 1-40 ppm of B as the grain refiner.
    The typical alloy contains Ag 94.5, Cu 4.3, and Ge 1.2%, and is
    suitable for strip manufacture by continuous casting followed by cold rolling
    with intermediate annealing. The polished surface of manufactured
    Ag-alloy articles is treated for tarnish resistance with
    an alkanethiol, alkyl thioglycollate, dialkyl sulfide, or dialkyl
    disulfide, especially stearyl mercaptan, cetyl mercaptan (octadecyl mercaptan),
    stearyl thioglycollate, or cetyl thioglocollate. The S-containing mols. are
    optionally dissolved in: (a) organic solvent (especially Pr bromide), and
applied
    as a polish or impregnated into a cleaning cloth; or (b) organic
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solvent modified by adding concentrated aqueous soap or detergent. The
resulting
    mixture are optionally diluted with water for the tarnish-preventing
     treatment.
     silver copper germanium alloy utensil tarnish
ST
     prevention thiol treatment
ΙT
     Metalworking
        (Ag-alloy; Ag-Ge-Cu alloy for decorative
        polished utensils with tarnish-preventing treatment)
ΙT
     Tarnishing
        (prevention, on Ag alloy; Ag-Ge-Cu alloy for
        decorative polished utensils with tarnish
        -preventing treatment)
ΙT
     Detergents
        (tarnish prevention with, on Ag alloy; Ag
        -Ge-Cu alloy for decorative polished utensils with
        tarnish-preventing treatment)
TT
     Thioethers
     Thiols, processes
     RL: CPS (Chemical process); PEP (Physical, engineering or chemical
     process); PROC (Process)
        (tarnish prevention with, on Ag alloy; Ag
        -Ge-Cu alloy for decorative polished utensils with
        tarnish-preventing treatment)
ΙT
     7440-42-8, Boron, uses
     RL: MOA (Modifier or additive use); USES (Uses)
        (Ag alloy containing; Ag-Ge-Cu alloy for decorative
        polished utensils with tarnish-preventing treatment)
ΙT
     802919-79-5
     RL: TEM (Technical or engineered material use); USES (Uses)
        (alloying of; Ag-Ge-Cu alloy for decorative polished
        utensils with tarnish-preventing treatment)
     802919-80-8
TT
     RL: TEM (Technical or engineered material use); USES (Uses)
        (decorative; Ag-Ge-Cu alloy for decorative polished
        utensils with tarnish-preventing treatment)
ΤТ
     68-11-1D, alkylated
     RL: CPS (Chemical process); PEP (Physical, engineering or chemical
     process); PROC (Process)
        (tarnish prevention with; Ag-Ge-Cu alloy for
        decorative polished utensils with tarnish
        -preventing treatment)
ΙT
     106-94-5, n-Propyl bromide
     RL: TEM (Technical or engineered material use); USES (Uses)
        (tarnish prevention with; Ag-Ge-Cu alloy for
        decorative polished utensils with tarnish
        -preventing treatment)
RE.CNT 8
              THERE ARE 8 CITED REFERENCES AVAILABLE FOR THIS RECORD
RE
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(2) Goddard; GB 1130540 A
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(4) Johns; WO 02095082 A2 CAPLUS
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(7) Metaleurop; GB 2255348 A CAPLUS
(8) Murphey; US 2841501 A
L36 ANSWER 4 OF 26 CAPLUS COPYRIGHT 2009 ACS on STN
AN
     2004:847649 CAPLUS
DN
    141:353637
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ED

Entered STN: 15 Oct 2004

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Pretreatment of Ag-alloy surface with organosulfur compounds for
      tarnishing prevention
      Johns, Peter Gammon; Harrison, Clare Elizabeth
ΤN
PA
      Middlesex Silver Co. Limited, UK
      PCT Int. Appl., 43 pp.
SO
      CODEN: PIXXD2
DT
      Patent
LA
      English
IC
      ICM C23F011-16
CC
      56-6 (Nonferrous Metals and Alloys)
FAN.CNT 1
      PATENT NO.
                            KIND DATE
                                                  APPLICATION NO.
      WO 2004087996
                             A1 20041014 WO 2004-GB1373
PΤ
                                                                              20040330
          W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BW, BY, BZ, CA, CH,
               CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, EG, ES, FI, GB, GD,
               GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC,
               LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NA, NI,
               NO, NZ, OM, PG, PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, SY,
          NO, NZ, OM, PG, PH, PL, P1, RO, RO, SC, SD, SE, SG, SK, SL, S1, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, YU, ZA, ZM, ZW RW: BW, GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZM, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM, AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IT, LU, MC, NL, PL, PT, RO, SE, SI, SK, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN,
               TD, TG
                                                 AU 2004-225693
CA 2004-2520807
EP 2004-724313
      AU 2004225693
                            A1 20041014
A1 20041014
A1 20060104
                             A1
                                                                              20040330
      CA 2520807
                                                                              20040330
      EP 1611267
                                                                               20040330
          R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT,
               IE, SI, LT, LV, FI, RO, MK, CY, AL, TR, BG, CZ, EE, HU, PL, SK
                            A 20060531 CN 2004-80011375 20040330
      CN 1780937
                             Τ
                                    20061012 JP 2006-506057
                                                                              20040330
      JP 2006523266
IN 2005DN04346 A 20070831 IN 2005-DN4346

MX 2005010452 A 20060510 MX 2005-10452

US 20070039665 A1 20070222 US 2005-551476

PRAI GB 2003-7290 A 20030331

WO 2004-GB1373 W 20040330
                                                                              20050926
                                                                              20050928
                                                                               20050929
CLASS
 PATENT NO. CLASS PATENT FAMILY CLASSIFICATION CODES
 WO 2004087996
                  ICM C23F011-16
                    IPCI C23F0011-16 [ICM, 7]; C23F0011-10 [ICM, 7, C*]
                    IPCR C23F0011-10 [I,C*]; C23F0011-16 [I,A]
                    ECLA C23F011/16; C23F011/16B
                            C23F0011-16 [ICM, 7]; C23F0011-10 [ICM, 7, C*]
 AU 2004225693
                    IPCI
                            C23F0011-10 [I,C*]; C23F0011-16 [I,A]
                    IPCR
                            C23F011/16; C23F011/16B
                    ECLA
                            C23F0011-16 [ICM,7]; C23F0011-10 [ICM,7,C*]
 CA 2520807
                    IPCI
                            C23F0011-10 [I,C*]; C23F0011-16 [I,A]
                    IPCR
                            C23F011/16; C23F011/16B
                    ECLA
                            C23F0011-16 [ICM,7]; C23F0011-10 [ICM,7,C*]
 EP 1611267
                    IPCI
                    IPCR
                            C23F0011-10 [I,C*]; C23F0011-16 [I,A]
                    ECLA
                            C23F011/16; C23F011/16B
                            C23F0011-16 [I,A]; C23F0011-10 [I,C*]
 CN 1780937
                    IPCI
                            C23F011/16; C23F011/16B
                    ECLA
                            C23F0011-00 [I,A]; C22C0005-06 [I,A]; C22C0005-08 [I,A]
 JP 2006523266
                    IPCI
                    IPCR
                            C23F0011-00 [I,C]; C23F0011-00 [I,A]; C22C0005-06
                            [I,C]; C22C0005-06 [I,A]; C22C0005-08 [I,A];
                            C23F0011-10 [I,C*]; C23F0011-16 [I,A]
                    FTERM 4K062/AA01; 4K062/BB21; 4K062/BC22; 4K062/FA16
 IN 2005DN04346 IPCI C23F0011-16 [ICM,7]; C23F0011-10 [ICM,7,C*]
 MX 2005010452 IPCI
                            C23F0011-16 [ICM, 7]; C23F0011-10 [ICM, 7, C*]
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ТΤ

ECLA C23F011/16; C23F011/16B US 20070039665 IPCI C23G0001-00 [I,A]; C23C0022-58 [I,A]; C23C0022-05 [I,C*] 148/271.000; 134/002.000 NCL The Ag alloys containing minor Ge (especially Ag-Cu-Ge alloys) AB to decrease the fire stain discoloration are pretreated on the surface with an alkanethiol, alkyl thioglycollate, dialkyl sulfide, or dialkyl disulfide to prevent tarnishing. The treatment with organosulfur compds. is suitable for manufactured Aq-alloy articles to prevent tarnished appearance during transit and the subsequent extended display without special packaging. The Aq -alloy surface is optionally treated with aqueous solution containing an alkanethiol, alkyl thioglycollate, dialkyl sulfide, or dialkyl disulfide, as well as a mixture of anionic surfactant and amphoteric or nonionic surfactant to solubilize the treatment agent. The typical ternary alloy contains Ag 80-96, Cu 1-19.9, and Ge 0.1-5%. ST silver copper germanium alloy tarnishing prevention organosulfur Surfactants ΤT (anionic, in tarnishing prevention; Ag-alloy surface treated with organosulfur compds. for tarnishing prevention) ΙT Surfactants (in tarnishing prevention; Ag-alloy surface treated with organosulfur compds. for tarnishing prevention) ΙT Surfactants (nonionic, in tarnishing prevention; Ag-alloy surface treated with organosulfur compds. for tarnishing prevention) ΙT Tarnishing (prevention of; Aq-alloy surface treated with organosulfur compds. for tarnishing prevention) ΤТ Thioethers Thiols, uses RL: TEM (Technical or engineered material use); USES (Uses) (tarnishing prevention by; Ag-alloy surface treated with organosulfur compds. for tarnishing prevention) ΙT 7440-56-4, Germanium, uses RL: MOA (Modifier or additive use); USES (Uses) (Ag alloys containing, tarnishing prevention on; Ag-alloy surface treated with organosulfur compds. for tarnishing prevention) ΙT 106-94-5, n-Propyl bromide RL: TEM (Technical or engineered material use); USES (Uses) (solvent, in tarnishing prevention; Ag-alloy surface treated with organosulfur compds. for tarnishing prevention) 2885-00-9, Octadecyl mercaptan 2917-26-2, Cetyl ΤТ mercaptan RL: CPS (Chemical process); PEP (Physical, engineering or chemical process); PROC (Process) (tarnishing prevention by; Ag-alloy surface treated with organosulfur compds. for tarnishing prevention) 39282-03-6, Sterling silver 103221-24-5 ΤТ 476614-10-5 476614-12-7 476614-13-8 RL: CPS (Chemical process); PEP (Physical, engineering or chemical process); PROC (Process) (tarnishing prevention on; Ag-alloy surface treated with organosulfur compds. for tarnishing prevention) 9080-17-5, Ammonium polysulfide ΤТ

RL: CPS (Chemical process); PEP (Physical, engineering or chemical

process); PROC (Process)

(test solution with, for tarnishing; Ag-alloy surface treated with organosulfur compds. for tarnishing prevention)

RE.CNT 9 THERE ARE 9 CITED REFERENCES AVAILABLE FOR THIS RECORD RE

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- L36 ANSWER 5 OF 26 CAPLUS COPYRIGHT 2009 ACS on STN
- AN 2004:396797 CAPLUS
- DN 141:113055
- ED Entered STN: 17 May 2004
- TI The formation of self-assembling membrane of hexadecane-thiol on silver to prevent the tarnishing
- AU Evesque, Magali; Keddam, Michel; Takenouti, Hisasi
- CS Laboratoires Interface et Systemes Electrochimiques, UPR15 du CNRS, Pierre et Marie Curie University, Paris, 75252, Fr.
- SO Electrochimica Acta (2004), 49(17-18), 2937-2943 CODEN: ELCAAV; ISSN: 0013-4686
- PB Elsevier Science B.V.
- DT Journal
- LA English
- CC 72-6 (Electrochemistry)
 Section cross-reference(s): 56
- AB Artifacts in Ag suffer from tarnishing when exposed to atms. polluted by sulfide. The authors found the optimum conditions to form an efficient, invisible and protective film against the tarnishing appearance on Ag in 0.5M NaCl solution containing 10 mM Na2S. This solution corresponds to a highly aggressive medium not only by the coupling of 2 aggressive agents, but also by a high concentration of S2-(320)
- ppm). The Ag surface was 1st degreased carefully by successive dippings in 3 organic solvents, EtOH, acetone and hexane, followed by a slight surface activation in H2SO4 solution Then, a Ag specimen was immersed in an isoPrOH solution with 0.15M C16H33SH during 1 h at 30°. The kinetics of tarnishing was tracked by reflectance, quartz-microbalance measurements, and electrochem. impedance spectroscopy. The protection of hexadecane-thiol reaches 90% in terms of reflectance, after 1 h of corrosion test, i.e., no alteration by visual inspection. The thiol film has a double structure, an inner self-assembling membrane of 1 or 2 monolayers and an outer-layer with some tenths micrometers. This surface film limits markedly the diffusion of dissolved oxygen to the electrode surface, thus slows down the rate of Ag sulfide (Ag2S) formation.
- ST self assembling membrane formation hexadecanethiol silver tarnishing prevention
- IT Adsorbed substances

(corrosion of silver with and without adsorbed hexadecanethiol in NaCl containing Na2S in tarnishing prevention study)

IT Microbalances

(electrochem. quartz crystal; in corrosion and characterization study of hexadecanethiol adsorbed film on silver in NaCl containing Na2S)

IT Adsorption

(hexadecanethiol by Ag in isoPrOH solution containing hexadecanethiol) ΤТ Electric impedance (in corrosion study of hexadecanethiol adsorbed film on silver in NaCl containing Na2S) ΙT (of silver with and without adsorbed hexadecanethiol in NaCl containing Na2S in tarnishing prevention study) ΙT (prevention by hexadecanethiol film formed by immersion of Aq in isoPrOH solution containing hexadecanethiol) 1313-82-2, Sodium sulfide (Na2S), uses 7647-14-5, Sodium chloride, uses RL: CPS (Chemical process); NUU (Other use, unclassified); PEP (Physical, engineering or chemical process); PRP (Properties); PROC (Process); USES (Uses) (elec. impedance and electrochem. quartz crystal microbalance study of hexadecanethiol adsorbed film on silver in NaCl containing Na2S in tarnishing prevention study) 2917-26-2, Hexadecanethiol ΤТ RL: MOA (Modifier or additive use); PEP (Physical, engineering or chemical process); PRP (Properties); PYP (Physical process); PROC (Process); USES (Uses) (formation of self-assembling membrane of hexadecanethiol on silver to prevent tarnishing in electrochem. quartz crystal microbalance and impedance study) 7440-22-4, Silver, properties RL: CPS (Chemical process); PEP (Physical, engineering or chemical process); PRP (Properties); PROC (Process) (tarnishing prevention by hexadecanethiol film formed by immersion of Ag in isoPrOH solution containing hexadecanethiol) THERE ARE 6 CITED REFERENCES AVAILABLE FOR THIS RECORD RE.CNT (1) Burleigh, D; Corrosion 2001, V57, P1066 (2) Burleigh, T; Corrosion 2002, V58, P49 CAPLUS (3) Kartlucke, D; Galvanotechnik 1992, V83, P1918 (4) Laibinis, P; Thin Films 1998, V24, P1 CAPLUS (5) Lee, J; Metall Mater Trans B 2001, V32, P895 (6) Trachli, B; Corros Sci 2002, V44, P997 CAPLUS L36 ANSWER 6 OF 26 CAPLUS COPYRIGHT 2009 ACS on STN ΑN 2002:49041 CAPLUS DN 136:187483 ED Entered STN: 18 Jan 2002 TΙ Self-assembled monolayers of perfluoroalkyl amideethanethiols, fluoroalkylthiols, and alkylthiols for the prevention of silver tarnish Burleigh, T. D.; Shi, C.; Kilic, S.; Kovacik, S.; Thompson, T.; Enick, R. ΑU CS Department of Materials Science and Engineering, University of Pittsburgh, Pittsburgh, PA, 15261, USA Corrosion (Houston, TX, United States) (2002), 58(1), 49-56 SO CODEN: CORRAK; ISSN: 0010-9312 PΒ NACE International DT Journal English LA 56-10 (Nonferrous Metals and Alloys) CC Section cross-reference(s): 42 AΒ Self-assembled monolayers (SAM) of perfluoroalkyl amideethanethiols. F(CF2)nCONH(CH2)2SH (n = 6, 7, or 8), inhibit the corrosion of silver by hydrogen sulfide (H2S) in air. Unlike conventional hydrocarbon thiols used to protect silver from corrosion, these

fluorinated amidethiols have a very low mercaptan odor, impart

fluorocarbon wettability properties to the silver surfaces, and exhibit intermol. assocns. via hydrogen bonding of the amide functionality. These fluorinated thiols were synthesized by reacting fluoroalkyl acid chloride with 2-aminoethanethiol, or by reacting Me fluoroalkanoate with 2-aminoethanethiol. SAM were formed by immersing silver coupons in 0.01, 0.1, and 1 wt% solns. of the fluorinated amidethiol in propanol (CH3CH2CH2OH), or by applying a thin film of the thiol solution that rapidly evaporated Electrochem. impedance spectroscopy (EIS) was used to evaluate the thickness and integrity of the monolayers. The thin films of evaporating thiol solution yielded rapid monolayer formation as a result of the increasing concentration of the thiol in the solution on the silver during the evaporation of the propanol. Accelerated tarnish tests were performed in a chamber that exposed the silver to air, water vapor, and 1 ppm H2S for 7 h to 24 h at 313 K. The tarnish resistances associated with a fluoroalkyl thiol (1= 1H, 2H, 2H- perfluorodecyl-1-thiol) and hexadecanethiol were also determined The best tarnish resistance was attained with the hexadecanethiol, and the perfluoroalkylamide ethanethiol yielded better corrosion resistance results than the fluoroalkylthiol. tarnishing protection silver hexadecanethiol Corrosion prevention Films Self-assembly Tarnishing (self-assembled monolayers for the prevention of silver tarnish) Thiols, reactions RL: RCT (Reactant); TEM (Technical or engineered material use); RACT (Reactant or reagent); USES (Uses) (self-assembled monolayers for the prevention of silver tarnish) 7440-22-4, Silver, processes RL: CPS (Chemical process); PEP (Physical, engineering or chemical process); PRP (Properties); RCT (Reactant); TEM (Technical or engineered material use); PROC (Process); RACT (Reactant or reagent); USES (Uses) (self-assembled monolayers for the prevention of silver tarnish) 2917-26-2, Hexadecanethiol 7783-06-4, Hydrogen sulfide, processes RL: CPS (Chemical process); PEP (Physical, engineering or chemical process); RCT (Reactant); TEM (Technical or engineered material use); PROC (Process); RACT (Reactant or reagent); USES (Uses) (self-assembled monolayers for the prevention of silver tarnish) 95612-22-9 115281-11-3 192137-69-2 RL: TEM (Technical or engineered material use); USES (Uses) (self-assembled monolayers for the prevention of silver tarnish) RE.CNT 19 THERE ARE 19 CITED REFERENCES AVAILABLE FOR THIS RECORD (1) Braach-Maksvytis, V; J Amer Chem Soc 2000, V122, P9544 CAPLUS (2) Burleigh, T; Corrosion 2001, V57(12), P1066 CAPLUS (3) Burleigh, T; J Electrochem Soc 1991, V138(8), PL34 CAPLUS (4) Enick, R; US 6183815 2001 CAPLUS (5) Jennings, G; Colloids Surf A 1996, V116, P105 CAPLUS (6) Kartluke, V; Galvanotech 1992, V83(6), P1918 (7) Lenk, T; Langmuir 1994, V10, P4610 CAPLUS

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ΙT

ΙT

ΤТ

ΤТ

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L36 ANSWER 7 OF 26 CAPLUS COPYRIGHT 2009 ACS on STN
ΑN
     2001:905959 CAPLUS
DN
     136:89158
     Entered STN: 16 Dec 2001
ED
     Tarnish protection of silver using a hexadecanethiol
ΤТ
     self-assembled monolayer and descriptions of accelerated tarnish
     tests
ΑU
     Burleigh, T. D.; Gu, Y.; Donahey, G.; Vida, M.; Waldeck, D. H.
CS
     Department of Materials Science and Engineering, University of Pittsburgh,
     Pittsburgh, PA, 15261, USA
SO
     Corrosion (Houston, TX, United States) (2001), 57(12), 1066-1074
     CODEN: CORRAK; ISSN: 0010-9312
PΒ
     NACE International
DT
     Journal
     Enalish
LA
     56-10 (Nonferrous Metals and Alloys)
CC
AΒ
     A four-step procedure was developed for depositing a hexadecanethiol
     self-assembled monolayer (SAM) onto the surface of silver to
     provide tarnish resistance. The four steps may be characterized
     as cleaning, etching, monolayer self-assembly, and rinsing. A key
     observation in this work is that an optimal deposition time exists for a
     given concentration of the hexadecanethiol. For example, a 2 vol% solution (2
mL
     hexadecanethiol in 98 mL trichloroethylene) required 30 min to 60 min for
     optimum coating formation. The quality of the coatings was characterized
     using water drop contact angle measurements and electrochem. impedance
     spectroscopy (EIS). In addition, two tarnish tests were developed
     specifically for this project. One test was a laboratory bench vapor test that
     could tarnish silver, copper, or brass to a dark color
     within a few hours. A second test exposed the silver to a
     stream of a sulfide-containing foam and could tarnish silver
     to black within several minutes. The degree of tarnishing from
     these two tests was quantified by measuring the% reflectance of the
     surface using visible light.
ST
    tarnish protection silver hexadecanethiol
ΙT
     Tarnishing
        (tarnish protection of silver using a
        hexadecanethiol self-assembled monolayer and descriptions of
        accelerated tarnish tests)
ΙT
     7440-22-4, Silver, processes
     RL: CPS (Chemical process); PEP (Physical, engineering or chemical
     process); PRP (Properties); RCT (Reactant); PROC (Process); RACT (Reactant
     or reagent)
        (tarnish protection of silver using a
        hexadecanethiol self-assembled monolayer and descriptions of
        accelerated tarnish tests)
ΤТ
     2917-26-2, Hexadecanethiol
     RL: CPS (Chemical process); PEP (Physical, engineering or chemical
     process); RCT (Reactant); PROC (Process); RACT (Reactant or reagent)
        (tarnish protection of silver using a
        hexadecanethiol self-assembled monolayer and descriptions of
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accelerated tarnish tests)

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- 2000:83978 CAPLUS ΑN
- DN 132:172522
- Entered STN: 04 Feb 2000 ED
- ΤI Electrochemical Cleaning of Surface-Confined Carbon Contamination in Self-Assembled Monolayers on Polycrystalline Ag and Au
- ΑU Schoenfisch, Mark H.; Ross, Azalia M.; Pemberton, Jeanne E.
- CS Department of Chemistry, University of Arizona, Tucson, AZ, 85721, USA
- SO Langmuir (2000), 16(6), 2907-2914 CODEN: LANGD5; ISSN: 0743-7463
- ΡВ American Chemical Society
- DTJournal
- LA English
- CC 72-2 (Electrochemistry)
 - Section cross-reference(s): 66, 73
- A protocol for electrochem. cleaning of carbon-contaminated alkanethiol AB SAMs at mech. polished (MP) Ag surfaces was characterized by surface Raman spectroscopy and electrochem. Vibrational information in the $\nu \text{(C-S)}\text{, }\nu \text{(C-C)}\text{, }\nu \text{(C-H)}\text{, and }\delta \text{(C-H)}$ regions is particularly useful in elucidating the degree of order and amount of contamination in propanethiol, dodecanethiol, and octadecanethiol monolayers before and after neg. potential exposure in several aqueous electrolytes. Specifically, Raman spectra indicate that electrochem. cleaning of alkanethiol SAMs at potentials neg. of the thiolate reduction removes carbonaceous species and greatly increases the film order near the sulfur headgroup.
- electrochem cleaning surface confined carbon contamination self assembled monolayer; polycryst silver gold self assembled monolayer carbon contamination electrocleaning; Raman spectra alkanethiol elf assembled monolayer gold silver
- ΙT Thiols (organic), properties RL: PEP (Physical, engineering or chemical process); PRP (Properties); PROC (Process)

(electrochem. cleaning of surface-confined carbon contamination in alkanethiol self-assembled monolayers on polycryst. Ag and Au) Self-assembled monolayers (electrochem. cleaning of surface-confined carbon contamination in self-assembled monolayers on polycryst. Ag and Au) Desorption (electrochem.; of alkanethiol with contamination removal: electrochem. cleaning of surface-confined carbon contamination in self-assembled monolayers on polycryst. Ag) Cleaning (electrochem.; of surface-confined carbon contamination in self-assembled monolayers on polycryst. Ag and Au) Electric potential (neg.; in alkanethiol desorption and contamination removal: electrochem. cleaning of surface-confined carbon contamination in self-assembled monolayers on polycryst. Ag and Au) Cyclic voltammetry (of Ru(NH3)63+ in KCl at alkanethiol self-assembled monolayers on Ag before and after neg. potential application and electrochem. cleaning) Raman spectra (of dodecanethiol and octadecanethiol and propanethiol self-assembled monolayers on Ag and Au: electrochem. cleaning of surface-confined carbon contamination in propanethiol self-assembled monolayers on polycryst. Ag and Au) 7447-40-7, Potassium chloride (KCl), uses RL: NUU (Other use, unclassified); PRP (Properties); USES (Uses) (cyclic voltammetry of Ru(NH3)63+ in KCl at alkanethiol self-assembled monolayers on Ag before and after neg. potential application and electrochem. cleaning) 18943-33-4, Hexaammineruthenium(3+) RL: PRP (Properties); RCT (Reactant); RACT (Reactant or reagent) (cyclic voltammetry of Ru(NH3)63+ in KCl at alkanethiol self-assembled monolayers on Ag before and after neg. potential application and electrochem. cleaning) 7681-49-4, Sodium fluoride, uses RL: NUU (Other use, unclassified); PRP (Properties); USES (Uses) (electrochem. cleaning of surface-confined carbon contamination in alkanethiol self-assembled monolayers on polycryst. Ag in solution of) 1322-36-7, Dodecanethiol RL: PEP (Physical, engineering or chemical process); PRP (Properties); PROC (Process) (electrochem. cleaning of surface-confined carbon contamination in dodecanethiol self-assembled monolayers on polycryst. Ag and Au) 2885-00-9, Octadecanethiol RL: PEP (Physical, engineering or chemical process); PRP (Properties); PROC (Process) (electrochem. cleaning of surface-confined carbon contamination in octadecanethiol self-assembled monolayers on polycryst. Ag and Au) 79869-58-2, Propanethiol RL: PEP (Physical, engineering or chemical process); PRP (Properties); PROC (Process) (electrochem. cleaning of surface-confined carbon contamination in propanethiol self-assembled monolayers on polycryst. Ag and Au)

7440-22-4, Silver, uses 7440-57-5, Gold, uses

RL: DEV (Device component use); PRP (Properties); USES (Uses)

(electrochem. cleaning of surface-confined carbon contamination in

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self-assembled monolayers on polycryst. Ag and Au)
     7440-44-0, Carbon, properties
ΤТ
     RL: OCU (Occurrence, unclassified); PEP (Physical, engineering or chemical
     process); PRP (Properties); OCCU (Occurrence); PROC (Process)
        (electrochem. cleaning of surface-confined carbon contamination in
        self-assembled monolayers on polycryst. Ag and Au)
RE.CNT
              THERE ARE 36 CITED REFERENCES AVAILABLE FOR THIS RECORD
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L36
ΑN
     2000:83975 CAPLUS
     132:199505
DN
     Entered STN: 04 Feb 2000
ED
ΤI
     Sequestration of carbonaceous species within alkanethiol self-assembled
     monolayers on Ag by Raman spectroscopy
ΑU
     Taylor, Chad E.; Schoenfisch, Mark H.; Pemberton, Jeanne E.
CS
     Department of Chemistry, University of Arizona, Tucson, AZ, 85721, USA
     Langmuir (2000), 16(6), 2902-2906
SO
     CODEN: LANGD5; ISSN: 0743-7463
PΒ
     American Chemical Society
DT
     Journal
LA
     English
CC
     66-4 (Surface Chemistry and Colloids)
     Raman spectra of hydrogenated CnSH (where, n = 3-5, 8, 9, 12, and 18) and
AB
     C8D17SH SAMs at mech. polished (MP) Ag indicate
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monolayer contamination by a small polyarom. hydrocarbon (PAH).

contaminant source at the unmodified MP Ag surface is identified using Raman spectroscopy, and thus, the contaminant is believed to be placed at this surface during the mech. polishing procedure. Notably, the PAH contaminant is not completely removed by either solvent dissoln. or alkanethiol adsorption, suggesting that it is strongly bound, and more significantly, sequestered within the alkanethiol SAM. Controlled incorporation of pyrene into dodecanethiol SAMs demonstrates that doping of alkanethiol SAMs may be possible for certain systems.

- ST alkanethiol self assembled monolayer silver PAH contaminant sequestration
- IT Polycyclic compounds

Polycyclic compounds

RL: MOA (Modifier or additive use); USES (Uses)

(aromatic hydrocarbons; sequestration of PAH contaminant within alkanethiol self-assembled monolayers on Ag)

IT Aromatic hydrocarbons, uses

Aromatic hydrocarbons, uses

RL: MOA (Modifier or additive use); USES (Uses)

(polycyclic; sequestration of PAH contaminant within alkanethiol self-assembled monolayers on Ag)

IT Self-assembled monolayers

(sequestration of PAH contaminant within alkanethiol self-assembled monolayers on Aq)

IT Thiols (organic), properties

RL: PEP (Physical, engineering or chemical process); PRP (Properties); PROC (Process)

(sequestration of PAH contaminant within alkanethiol self-assembled monolayers on Ag)

IT Complexation

(sequestration; of PAH contaminant within alkanethiol self-assembled monolayers on Ag)

IT 129-00-0, Pyrene, processes

RL: MOA (Modifier or additive use); PEP (Physical, engineering or chemical process); PROC (Process); USES (Uses)

(model contaminant; sequestration of PAH contaminant within dodecanol
self-assembled monolayers on Ag)

IT 7440-22-4, Silver, processes

RL: NUU (Other use, unclassified); PEP (Physical, engineering or chemical process); PROC (Process); USES (Uses)

(sequestration of PAH contaminant within alkanethiol self-assembled monolayers on)

IT 109-79-5, Butanethiol 110-66-7, Pentanethiol 1322-36-7, Dodecanethiol 2885-00-9, Octadecanethiol 79869-58-2, Propanethiol 94805-33-1, Octanethiol

RL: PEP (Physical, engineering or chemical process); PRP (Properties); PROC (Process)

(sequestration of PAH contaminant within alkanethiol self-assembled monolayers on Ag)

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L36 ANSWER 10 OF 26 CAPLUS COPYRIGHT 2009 ACS on STN
AN 1999:819549 CAPLUS
DN
   132:67594
   Entered STN: 30 Dec 1999
ED
ΤI
    Aliphatic alcohol for inhibiting tarnish formation in cleaning
     of silver surfaces with ether stabilized, n-propyl bromide
     solvent systems
ΙN
    Shubkin, Ronald L.
PA
    Albemarle Corp., USA
SO
    PCT Int. Appl., 14 pp.
    CODEN: PIXXD2
DT
    Patent
LA
    English
    ICM C23G005-028
IC
     ICS C11D007-50
CC
     56-10 (Nonferrous Metals and Alloys)
FAN.CNT 1
    PATENT NO.
                        KIND
                                           APPLICATION NO. DATE
                                DATE
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     WO 9967445
                                           WO 1999-US12965
PΙ
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                                19991229
                                                                    19990609
         W: CA, JP, KR
         RW: AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL,
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     CA 2333496
                          Α1
                                19991229
                                            CA 1999-2333496
                                                                    19990609
     EP 1090164
                                20010411
                                           EP 1999-928514
                          Α1
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                             20031217
     EP 1090164
                         В1
         R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT,
             IE, FI
JP 2002519506 T 20020702
AT 256767 T 20040115
PRAI US 1998-104898 A 19980625
                                           JP 2000-556083
                                                                    19990609
                                           AT 1999-928514
                                                                   19990609
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WO 1999-US12965 W 19990609 CLASS CLASS PATENT FAMILY CLASSIFICATION CODES PATENT NO. ____ _____ WO 9967445 ICM C23G005-028 ICS C11D007-50 IPCI C23G0005-028 [ICM, 6]; C23G0005-00 [ICM, 6, C*]; C11D0007-50 [ICS, 6] IPCR C11D0007-22 [I,C*]; C11D0007-26 [I,A]; C11D0007-30 [I,A]; C11D0007-32 [I,A]; C11D0007-50 [I,C*]; C11D0007-50 [I,A]; C11D0011-00 [I,C*]; C11D0011-00 [I,A]; C23G0005-00 [I,C*]; C23G0005-028 [I,A] ECLA C11D007/26A; C11D011/00B2D8; C11D011/00B10; C23G005/028; C11D007/26C; C11D007/32C; C11D007/50A; C11D011/00B2D2 CA 2333496 C23G0005-028 [ICM, 6]; C23G0005-00 [ICM, 6, C*]; IPCI C11D0007-50 [ICS,6] C11D0007-22 [I,C*]; C11D0007-26 [I,A]; C11D0007-30 IPCR [I,A]; C11D0007-32 [I,A]; C11D0007-50 [I,C*]; C11D0007-50 [I,A]; C11D0011-00 [I,C*]; C11D0011-00 [I,A]; C23G0005-00 [I,C*]; C23G0005-028 [I,A] ECLA C11D007/26A; C11D007/26C; C11D007/32C; C11D007/50A; C11D011/00B2D2; C11D011/00B2D8; C11D011/00B10; C23G005/028 EP 1090164 IPCI C23G0005-028 [ICM,6]; C23G0005-00 [ICM,6,C*]; C11D0007-50 [ICS, 6] C11D0007-22 [I,C*]; C11D0007-26 [I,A]; C11D0007-30 IPCR [I,A]; C11D0007-32 [I,A]; C11D0007-50 [I,C*]; C11D0007-50 [I,A]; C11D0011-00 [I,C*]; C11D0011-00 [I,A]; C23G0005-00 [I,C*]; C23G0005-028 [I,A] ECLA C11D007/26A; C11D007/26C; C11D007/32C; C11D007/50A; C11D011/00B2D2; C11D011/00B2D8; C11D011/00B10; C23G005/028 C23G0005-028 [ICM,7]; C23G0005-00 [ICM,7,C*]; JP 2002519506 IPCI C11D0007-26 [ICS,7]; C11D0007-30 [ICS,7]; C11D0007-22 [ICS, 7, C*]; C11D0007-50 [ICS, 7] C11D0007-22 [I,C*]; C11D0007-26 [I,A]; C11D0007-30 IPCR [I,A]; C11D0007-32 [I,A]; C11D0007-50 [I,C*]; C11D0007-50 [I,A]; C11D0011-00 [I,C*]; C11D0011-00 [I,A]; C23G0005-00 [I,C*]; C23G0005-028 [I,A] AT 256767 IPCI C23G0005-028 [ICM, 7]; C23G0005-00 [ICM, 7, C*]; C11D0007-50 [ICS, 7] ECLA C11D007/26A; C11D007/26C; C11D007/32C; C11D007/50A; C11D011/00B2D2; C11D011/00B2D8; C11D011/00B10; C23G005/028 AB The tarnishing of Ag surface in cleaning with an ether-stabilized Pr bromide solution is inhibited by adding a saturated aliphatic C1-10 alc. at 0.1-15%. The resulting bath is suitable for cleaning of Aq-coated electronic parts, and typically contains the Pr bromide at ≥ 80 , ether (especially 1,3-dioxolane) as the stabilizer and metal passivator at 1.5-2.5, and the alc. (typically 1-propanol) 1.5-3.5%. The stabilized bath for vapor-type cleaning of Ag-coated steel strip contains 1,3-dioxolane 1.50, 1,2-epoxybutane 0.50, nitromethane 0.50, and 1-propanol 3.50%, vs. dark tarnish in 10-min test at 71° in the similar bath containing 4.0% 1.3-dioxolane with no 1-propanol. ST silver tarnish prevention cleaning propyl bromide solvent; aliph alc inhibitor tarnish silver cleaning bath

Alcohols, uses

RL: MOA (Modifier or additive use); USES (Uses)

(aliphatic, C1-10, cleaning bath containing; aliphatic alc. as tarnish

ΙT

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inhibitor in stable solvent bath for cleaning of silver
        surface)
ΤТ
     Ethers, uses
     RL: MOA (Modifier or additive use); USES (Uses)
        (cyclic, stabilizer, cleaning bath containing; aliphatic alc. as
        tarnish inhibitor in stable solvent bath for cleaning of
        silver surface)
ΙT
     Tarnishing
        (on silver; aliphatic alc. as tarnish inhibitor in
        stable solvent bath for cleaning of silver surface)
ΤТ
     Electric contacts
        (silver-coated, cleaning of; aliphatic alc. as tarnish
        inhibitor in stable solvent bath for cleaning of silver
        surface)
     67-63-0, 2-Propanol, uses
                                71-23-8, 1-Propanol, uses
ΤT
                                                            71-36-3,
     1-Butanol, uses 75-52-5, Nitromethane, uses 75-65-0, tert-Butanol,
            75-85-4, 1,1-Dimethylpropan-1-ol 78-83-1, 2-Methylpropan-1-ol,
            78-92-2, 2-Butanol 106-88-7, 1,2-Epoxybutane 137-32-6,
     2-Methylbutan-1-ol 598-75-4, 1,2-Dimethylpropan-1-ol
                                                             646-06-0,
     1,3-Dioxolane
     RL: MOA (Modifier or additive use); USES (Uses)
        (cleaning bath containing; aliphatic alc. as tarnish inhibitor in
        stable solvent bath for cleaning of silver surface)
ΙT
     106-94-5, n-Propyl bromide
     RL: TEM (Technical or engineered material use); USES (Uses)
        (cleaning bath; aliphatic alc. as tarnish inhibitor in stable
        solvent bath for cleaning of silver surface)
ΙT
     7440-22-4, Silver, processes
     RL: PEP (Physical, engineering or chemical process); PROC (Process)
        (cleaning of; aliphatic alc. as tarnish inhibitor in stable
        solvent bath for cleaning of silver surface)
RE.CNT 12
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AN
     131:352890
DN
     Entered STN: 25 Nov 1999
ED
ΤI
     Acetylene compound for inhibiting tarnish formation when
     cleaning silver with ether-stabilized, n-propyl bromide-based
     solvent systems and cleaning electronic parts
     Shubkin, Ronald L.
IN
     Albemarle Corporation, USA
PA
SO
     U.S., 4 pp.
     CODEN: USXXAM
DT
     Patent
LA
     English
TC
     ICM C11D003-24
     ICS C11D003-43
INCL 510412000
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CC 46-6 (Surface Active Agents and Detergents)
Section cross-reference(s): 76

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PI	US CA	5990071		A A1 A1	19991123 19991229 19991229		US CA	1998- 1999-	1048	72 534		19 19	9980 9990 9990	625 609
		W: CA, RW: AT, PT,	BE, CH		, DK, ES,	FI,	FR	, GB,	GR,	IE,	IT,	LU,	MC,	NL,
	EP	1090163			20010411 , ES, FR,									
		IE,	FΙ											
	US WO	20025195 1998-104 1999-US1	872	A	20020702 19980625 19990609		JP	2000-	5560	84		19	9990	609
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US 5990071		ICM ICS	C11D003-24 C11D003-43											
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			NCL	510/412	C23G0005-0 .000; 252, .000; 510,	/364	.00	0; 51	0/17	5.00	0; 53	10/25	55.0	
			ECLA		50A6; C1									
CA 2333534		IPCI	C23G000	5-028 [ICM 7-50 [ICM		; C	23G00	05-0	0 []	СМ, 6	,C*];	;		
			IPCR	C11D000° [N,A]; C C11D000°	7-22 [N,C C11D0007-2 7-50 [I,A]	*]; 28 []; C	N,A 11D]; C1 0011-	1D00 00 [07-5 I,C*	0 [I]; C	,C*]; 11D0(;)11-	
			ECLA		/50A6; C1									M11D;
WO 9967446		IPCI	C23G0005-028 [ICM,6]; C23G0005-00 [ICM,6,C*]; C11D0007-50 [ICS,6]											
			IPCR	C11D000° [N,A]; C C11D000°	7-22 [N,C) C11D0007-2 7-50 [I,A] C23G0005-0	*]; 28 []; C	N,A 11D]; C1 0011-	1D00 00 [07-5 I,C*	0 [I,]; Ci	,C*]; 11D0(;)11-	
			ECLA		/50A6; C1									M11D;
EP 3	1090)163	IPCI	C23G000	5-028 [ICM 7-50 [ICS,		; C	23G00	05-0	[] O	СМ, 6	,C*];	;	
			IPCR	C11D000° [N,A]; C C11D000°	7-22 [N,C 011D0007-2 7-50 [I,A]	*]; 28 []; C	N,A 11D]; C1 0011-	1D00 00 [07-5 I,C*	0 [I]; C	,C*]; 11D0(;)11-	
			ECLA		C23G0005-0 /50A6; C13									M11D;
JP 2	2002519507		IPCI	C23G000	5-028 [ICM 7-50 [ICS,		; C	23G00	05-0	0 []	СМ, 7	,C*];	;	
			IPCR	C11D000° [N,A]; (C11D000°	7-22 [N,C C11D0007-2 7-50 [I,A] C23G0005-0	*]; 28 []; C	N,A 11D]; C1 0011-	1D00 00 [07-5 I,C*	0 [I,]; Ci	,C*]; 11D0(;)11-	

```
Ag tarnishing is inhibited when using
AΒ
     ether-stabilized, Pr bromide-based cleaning compns. by including a small
     amount of ≥1 C3-8 acetylenic hydrocarbon or halohydrocarbon. Adding
     4% dioxolane, and 3-butyne-2-ol to stabilized Pr bromide cleaner resulted
     in no tarnishing of Ag-plated steel coupons.
ST
     acetylenic compd tarnish inhibitor solvent cleaner; propyl
     bromide solvent cleaner silver; ether stabilized solvent cleaner
ΙT
     Cleaning solvents
        (acetylene compound for inhibiting tarnish formation when
        cleaning silver with ether-stabilized, Pr bromide-based
        solvent systems)
ΤТ
     Tarnishing
        (inhibiting; acetylene compound for inhibiting tarnish
        formation when cleaning silver with ether-stabilized, Pr
        bromide-based solvent systems)
     106-94-5, n-Propyl bromide
ΤТ
     RL: TEM (Technical or engineered material use); USES (Uses)
        (solvent; acetylene compound for inhibiting tarnish formation
        when cleaning silver with ether-stabilized, Pr bromide-based
        solvent systems)
ΤT
     96-48-0, \cdot \gamma.-Butyrolactone
                                  109-99-9, Tetrahydrofuran, uses
     110-88-3, Trioxane, uses
                              123-91-1, 1,4-Dioxane, uses
     1,3-Dioxolane
     RL: MOA (Modifier or additive use); USES (Uses)
        (stabilizer; acetylene compound for inhibiting tarnish
        formation when cleaning silver with ether-stabilized, Pr
        bromide-based solvent systems)
     106-96-7, Propargyl bromide 107-19-7, Propargyl alcohol
ΙT
                                                                115-19-5,
     2-Methyl-3-butyn-2-ol
                            624-65-7, Propargyl chloride
                                                           693-02-7, 1-Hexyne
     764-01-2, 2-Butyn-1-ol
                              927-74-2, 3-Butyn-1-ol
                                                      928-49-4, 3-Hexyne
     2028-63-9, 3-Butyn-2-ol
     RL: MOA (Modifier or additive use); USES (Uses)
        (tarnish inhibitor; acetylene compound for inhibiting
        tarnish formation when cleaning silver with
        ether-stabilized, Pr bromide-based solvent systems)
RE.CNT 9
             THERE ARE 9 CITED REFERENCES AVAILABLE FOR THIS RECORD
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(2) Anon; JP 6119700 1986
(3) Anon; JP 62-7798 1987 CAPLUS
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(5) Flynn; US 5814595 1998 CAPLUS
(6) Flynn; US 5827812 1998 CAPLUS
(7) Hartzler; US 3758503 1973 CAPLUS
(8) Henry; US 5403507 1995 CAPLUS
(9) Oshima; US 5492645 1996 CAPLUS
L36 ANSWER 12 OF 26 CAPLUS COPYRIGHT 2009 ACS on STN
    1999:734696 CAPLUS
ΑN
     132:86245
DN
ED
     Entered STN: 19 Nov 1999
     New normal-propyl bromide based cleaning technology for the electronics
ΤI
     industry
     Chang, Suae-chen; Shubkin, Ronald L.
ΑU
CS
     Albemarle Corporation, Baton Rouge, LA, USA
SO
     Circuit World (1999), 25(4), 17-21
    CODEN: CIWODV; ISSN: 0305-6120
PΒ
    MCB University Press
DT
    Journal; General Review
    English
LA
CC
    76-0 (Electric Phenomena)
```

A review with 5 refs. Precision cleaning with solvent systems based on Pr

AΒ

bromide (nPB) has become an important component of the circuit board fabrication process. The nPB-based cleaners have proved themselves valuable alternatives to the once popular chlorocarbons and hydrochlorocarbons. These latter solvents have been largely banned or restricted because of toxicol. and/or environmental considerations. Pr bromide has nearly identical phys. and cleaning properties to 1,1,1-trichloroethane (1,1,1-T or TCA), but it has a low ozone depletion potential (ODP) and a low global warming potential (GWP). A growing body of evidence shows that nPB-based solvents are safe, effective, and cost-efficient alternatives for precision cleaning applications. New formulations have now been developed specifically for the electronics industry. The first challenge is the efficient removal of ionic residues left by certain types of solder flux. The second challenge is the prevention of tarnish on silver leads used on some circuit boards. Case histories and a discussion of relevant toxicol., environmental, and regulatory considerations are included.

ST review propyl bromide cleaning electronic industry

ΙT Cleaning

Electronics

(normal-Pr bromide based cleaning technol. for electronics industry)

ΤT 106-94-5, Propyl bromide

RL: NUU (Other use, unclassified); USES (Uses)

(normal-Pr bromide based cleaning technol. for electronics industry)

THERE ARE 5 CITED REFERENCES AVAILABLE FOR THIS RECORD RE.CNT

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- L36 ANSWER 13 OF 26 CAPLUS COPYRIGHT 2009 ACS on STN
- ΑN 1999:236323 CAPLUS
- DN 131:10750
- EDEntered STN: 16 Apr 1999
- Effect of surface roughness on the self-assembly of octadecanethiol ΤI monolayer onto polycrystalline noble metal surfaces
- ΑU Subramanian, R.; Lakshminarayanan, V.
- Raman Research Institute, Bangalore, 560 080, India CS
- SO Current Science (1999), 76(5), 665-669 CODEN: CUSCAM; ISSN: 0011-3891
- PB Current Science Association
- DΤ Journal
- LA English
- CC 72-2 (Electrochemistry)
- Section cross-reference(s): 66
- The role of surface roughness on defect formation in ocatadecanethiol AΒ monolayer covered surfaces of gold, silver and copper was studied using cyclic voltammetry and scanning tunneling microscopy. The adsorption of alkanethiol on surfaces subjected to various pre-treatments like mech. polishing using different grades of emery and alumina indicates that a surface polished with 0.05 μm alumina has a significantly greater number of defect sites, thereby allowing access to redox species, compared to a surface treated with coarse emery. Scanning tunneling microscopic studies reveal that for a given area, a 'smooth' alumina polished surface has a number of closely spaced corrugations compared to a surface roughened using a coarse emery. exists a direct correlation between surface roughness and barrier efficiency.
- ST surface roughness self assembly octadecanethiol monolayer polycryst noble

metal; cyclic voltammetry octadecanethiol monolayer gold silver copper surface pretreatment; STM surface roughness effect octadecanethiol monolayer noble metal Adsorbed monolayers Self-assembly (effect of surface roughness on self-assembly of octadecanethiol monolayer onto copper or silver or gold surfaces studied by cyclic voltammetry and scanning tunneling microscopy) Cyclic voltammetry Scanning tunneling microscopy Surface roughness (effect of surface roughness on self-assembly of octadecanethiol monolayer onto polycryst. noble metal surfaces studied by cyclic voltammetry and scanning tunneling microscopy) Noble metals RL: PEP (Physical, engineering or chemical process); PRP (Properties); PROC (Process) (effect of surface roughness on self-assembly of octadecanethiol monolayer onto polycryst. noble metal surfaces studied by cyclic voltammetry and scanning tunneling microscopy) Polishing (of copper or silver or gold in study of effect of surface roughness on self-assembly of octadecanethiol monolayer onto copper or silver or gold surfaces studied by cyclic voltammetry and scanning tunneling microscopy) Adsorption (of octadecanethiol by copper or silver or gold surfaces and effect of surface roughness studied by cyclic voltammetry and scanning tunneling microscopy scanning tunneling microscopy) 7440-22-4, Silver, uses 7440-50-8, Copper, uses 7440-57-5, Gold, uses RL: DEV (Device component use); PEP (Physical, engineering or chemical process); PRP (Properties); PROC (Process); USES (Uses) (effect of surface roughness on self-assembly of octadecanethiol monolayer onto copper or silver or gold surfaces studied by cyclic voltammetry and scanning tunneling microscopy) 2885-00-9, Octadecanethiol RL: PEP (Physical, engineering or chemical process); PRP (Properties); PROC (Process) (effect of surface roughness on self-assembly of octadecanethiol monolayer onto polycryst. noble metal surfaces studied by cyclic voltammetry and scanning tunneling microscopy) RE.CNT 25 THERE ARE 25 CITED REFERENCES AVAILABLE FOR THIS RECORD (1) Abd El Haleem, S; J Electroanal Chem 1981, V117, P309 CAPLUS (2) Bain, C; J Am Chem Soc 1988, V110, P5897 CAPLUS (3) Chidsey, C; Langmuir 1990, V6, P682 CAPLUS (4) Creager, S; Langmuir 1992, V8, P854 CAPLUS (5) Dirske, T; Electrochim Acta 1989, V34, P647 (6) Feng, Y; J Electrochem Soc 1997, V144, P55 CAPLUS (7) Finklea, H; Electroanalytical Chemistry A Series of Advances 1996, V19, P124 (8) Finklea, H; Langmuir 1990, V6, P371 CAPLUS (9) Finklea, H; Langmuir 1993, V9, P3660 CAPLUS (10) Guo, L; Langmuir 1994, V10, P4588 CAPLUS (11) Hamelin, A; J Electroanal Chem 1990, V295, P291 CAPLUS (12) Ishibashi, M; Electrochim Acta 1996, V41, P241 CAPLUS (13) Laibinis, P; J Am Chem Soc 1991, V113, P7152 CAPLUS

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ΙT

ΙT

ΙT

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ΙT

RE

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- (18) Sabatini, E; J Phys Chem 1987, V91, P6663
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- (20) Trasatti, S; Pure Appl Chem 1991, V63, P711 CAPLUS
- (21) Ulman, A; An Introduction to Ultrathin Organic Films From Langmuir-Blodgett to Self-Assembly 1991
- (22) Walczak, M; J Am Chem Soc 1991, V113, P2370 CAPLUS
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- (24) Xu, J; J Colloid Interface Sci 1995, V176, P138 CAPLUS
- (25) Yamamoto, Y; J Electrochem Soc 1993, V140, P436 CAPLUS
- L36 ANSWER 14 OF 26 CAPLUS COPYRIGHT 2009 ACS on STN
- AN 1998:269960 CAPLUS
- DN 129:45750
- OREF 129:9485a,9488a
- ED Entered STN: 13 May 1998
- TI Air Stability of Alkanethiol Self-Assembled Monolayers on Silver and Gold Surfaces
- AU Schoenfisch, Mark H.; Pemberton, Jeanne E.
- CS Department of Chemistry, University of Arizona, Tucson, AZ, 85721, USA
- SO Journal of the American Chemical Society (1998), 120(18), 4502-4513 CODEN: JACSAT; ISSN: 0002-7863
- PB American Chemical Society
- DT Journal
- LA English
- CC 66-4 (Surface Chemistry and Colloids) Section cross-reference(s): 72, 73
- AB Surface Raman spectroscopy, electrochem., and XPS were used to study the effects of air exposure on the stability of self-assembled monolayers (SAM) formed from alkanethiols on mech. polished, smooth Ag and Au surfaces. Raman spectra exhibited oxidized S modes after only hours of air exposure. XPS of the S 2p region provided addnl. evidence of S oxidation Cyclic voltammetry of Ru(NH3)63+ indicated that oxidized alkanethiol SAM retain blocking characteristics toward electron transfer, even after exposure of the oxidized SAM-surface to a solubilizing solvent. Control expts. suggested O3 as the primary oxidant in ambient laboratory air which causes rapid oxidation of the thiolate moiety. Results have important ramifications for the general use of SAM in many proposed applications.
- ST alkanethiol self assembled monolayer air stability; silver surface alkanethiol monolayer air stability; gold surface alkanethiol monolayer air stability; ozone oxidn alkanethiol monolayer
- IT Thiols (organic), properties
 RL: PEP (Physical, engineering or chemical process); PRP (Properties);
 PROC (Process)
 - (laboratory air exposure and alkyl chain length effect on stability of alkanethiol self-assembled monolayers on silver and gold surfaces in absence of light)
- IT Adsorbed monolayers
 - (self-assembled; laboratory air exposure and alkyl chain length effect on stability of alkanethiol self-assembled monolayers on silver and gold surfaces in absence of light)
- IT Oxidation
 - (surface; laboratory air exposure and alkyl chain length effect on stability of alkanethiol self-assembled monolayers on silver and gold surfaces in absence of light)
- IT 18943-33-4
 - RL: NUU (Other use, unclassified); PEP (Physical, engineering or chemical process); PROC (Process); USES (Uses)
 - (laboratory air exposure and alkyl chain length effect on stability of alkanethiol self-assembled monolayers on silver and gold surfaces in absence of light)
- IT 108-98-5, Thiophenol, properties 1322-36-7, Dodecanethiol

```
2885-00-9, Octadecanethiol
                                 79869-58-2, Propanethiol
     RL: PEP (Physical, engineering or chemical process); PRP (Properties);
     PROC (Process)
        (laboratory air exposure and alkyl chain length effect on stability of
        alkanethiol self-assembled monolayers on silver and gold
        surfaces in absence of light)
ΙT
     10028-15-6, Ozone, reactions
     RL: OCU (Occurrence, unclassified); RCT (Reactant); OCCU (Occurrence);
     RACT (Reactant or reagent)
        (oxidant; laboratory air exposure and alkyl chain length effect on stability
        of alkanethiol self-assembled monolayers on silver and gold
        surfaces in absence of light)
     7440-22-4, Silver, properties
                                      7440-57-5, Gold, properties
IΤ
     RL: NUU (Other use, unclassified); PRP (Properties); USES (Uses)
        (polycryst.; laboratory air exposure and alkyl chain length effect on
        stability of alkanethiol self-assembled monolayers on silver
        and gold surfaces in absence of light)
              THERE ARE 43 CITED REFERENCES AVAILABLE FOR THIS RECORD
RE.CNT
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   Allied Topics 1996
L36 ANSWER 15 OF 26 CAPLUS COPYRIGHT 2009 ACS on STN
   1993:8126 CAPLUS
DN
   118:8126
OREF 118:1671a,1674a
    Entered STN: 10 Jan 1993
    Curable organopolysiloxane compositions for metals and their cured
    products
IN
    Kawate, Yasutoshi; Aramata, Mikio; Noguchi, Naoya
    Shin-Etsu Chemical Industry Co., Ltd., Japan
PΑ
    Jpn. Kokai Tokkyo Koho, 4 pp.
SO
    CODEN: JKXXAF
DT
    Patent
    Japanese
LA
    ICM C08L083-06
IC
    ICS C08K005-37
    38-3 (Plastics Fabrication and Uses)
CC
    Section cross-reference(s): 55, 56
FAN.CNT 1
                      KIND
                             DATE
                                         APPLICATION NO.
    PATENT NO.
                       ____
    JP 04253769
                       A
                              19920909
                                         JP 1991-35408
                                                               19910205
    JP 2762172
                       B2 19980604
PRAI JP 1991-35408
                              19910205
CLASS
PATENT NO. CLASS PATENT FAMILY CLASSIFICATION CODES
 ______
JP 04253769
              ICM C08L083-06
                ICS
                      C08K005-37
                IPCI C08L0083-06 [ICM,5]; C08L0083-00 [ICM,5,C*];
                      C08K0005-37 [ICS,5]; C08K0005-00 [ICS,5,C*]
                IPCR C08K0005-37 [I,A]; C08K0005-00 [I,C*]; C08K0005-54
                      [I,A]; C08K0005-5419 [I,A]; C08L0083-00 [I,C*];
                      C08L0083-04 [I,A]; C08L0083-06 [I,A]
AΒ
    The title compns. which effectively inhibit sulfurization of metals
    contain organopolysiloxanes whose both terminals are blocked with OH,
    ≥2 hydrolyzable group-containing organic Si compds. or their hydrolyzates,
    crosslinking catalysts, and organomercaptans. Thus, OH-terminated
    dimethylpolysiloxane (20,000 cSt) was mixed with vinyltributanoximesilane,
    dibutyltin dioctoate, stearyl mercaptan, dimethylpolysiloxane (100 cSt),
    and silica-based fillers under anhydrous condition, spread on Ag
    -plated plates, then cured at 20-50% relative humidity for 48 h to give
    sample specimens, which when left in 1% H2S-containing dry air for 14 days
    completely inhibit sulfurization of the surfaces of the plates.
ST
    curable organopolysiloxane compn organomercaptan; silicon compd
    organopolysiloxane compn curable; mercaptan organopolysiloxane compn
    curable; sulfurization metal inhibiting organopolysiloxane compn
ΙT
    Sulfurization and Sulfidization
        (inhibition of, by organopolysiloxane compns. containing organomercaptans)
ΙT
    Tarnishing
        (prevention of, of silver, silicone coatings for)
    Coating materials
ΙT
        (tarnish-preventing, silicones, for silver)
    Siloxanes and Silicones, uses
ΤT
    RL: USES (Uses)
        (di-Me, hydroxy-terminated, vinyltributanoximesilane-crosslinked,
       containing organomercaptans, with sulfurization-inhibiting effects on
```

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```
metals)
     2224-33-1
ΤТ
     RL: MOA (Modifier or additive use); USES (Uses)
        (crosslinking agents, for hydroxy-terminated dimethylpolysiloxanes)
     4731-77-5, Dibutyltin dioctoate
ΤТ
     RL: CAT (Catalyst use); USES (Uses)
        (crosslinking catalysts, for organopolysiloxane compns.)
ΙT
     2885-00-9, Stearyl mercaptan 31494-22-1, Oleyl mercaptan
     RL: USES (Uses)
        (organopolysiloxane compns. containing, for inhibition of metal
        sulfurization)
     7704-34-9
ΤТ
     RL: USES (Uses)
        (sulfurization and Sulfidization, inhibition of, by organopolysiloxane
        compns. containing organomercaptans)
     7440-22-4, Silver, uses
ΤТ
     RL: USES (Uses)
        (tarnishing-preventing silicone coatings for)
L36 ANSWER 16 OF 26 CAPLUS COPYRIGHT 2009 ACS on STN
ΑN
    1992:536001 CAPLUS
DN
     117:136001
OREF 117:23503a,23506a
    Entered STN: 04 Oct 1992
    Aqueous emulsion for temporary protection of silver and copper
    surfaces against tarnishing
ΙN
    Grossmann, Hermann
PA
    Doduco GmbH und Co. Dr. Eugen Duerrwaechter, Germany
    Eur. Pat. Appl., 6 pp.
SO
    CODEN: EPXXDW
DT
    Patent
   German
LA
    ICM C23F011-16
TC
    56-10 (Nonferrous Metals and Alloys)
CC
FAN.CNT 1
    PATENT NO.
                       KIND DATE
                                         APPLICATION NO. DATE
                        ____
                                          _____
    EP 492487
                       A1 19920701
                                         EP 1991-121903
PΙ
                        B1 19960320
     EP 492487
    DE 4041596 A1 19920702 DE 1990-4041596 ES 2086471 T3 19960701 DC 1000
        R: DE, ES, FR, GB, IT, NL
                                                                 19901222
                              19960701 ES 1991-121903
                                                                 19911220
PRAI DE 1990-4041596 A
DE 1991-4124955 A
                              19901222
                              19910727
CLASS
            CLASS PATENT FAMILY CLASSIFICATION CODES
PATENT NO.
 EP 492487
               ICM C23F011-16
                IPCI C23F0011-16 [ICM,5]; C23F0011-10 [ICM,5,C*]
                IPCR
                       C23F0011-10 [I,C*]; C23F0011-16 [I,A]
                ECLA
                       C23F011/16B
                       C23F0011-12 [ICM,5]; C23F0011-16 [ICS,5]; C23F0011-10
 DE 4041596
                IPCI
                       [ICS, 5, C*]; C09K0015-06 [ICA, 5]; C09K0015-12 [ICA, 5];
                       C09K0015-00 [ICA, 5, C*]; B01F0017-42 [ICA, 5];
                       B01F0017-38 [ICA,5]
                       C23F0011-10 [I,C*]; C23F0011-16 [I,A]
                IPCR
                ECLA
                       C23F011/16B
                IPCI
 ES 2086471
                       C23F0011-16 [ICM,6]; C23F0011-10 [ICM,6,C*]
                IPCR C23F0011-10 [I,C*]; C23F0011-16 [I,A]
                ECLA C23F011/16B
     The emulsion of pH 1-10 (preferably 2-4) comprises a hydrophobic inhibitor
AB
```

of a C \geq 12 thioalc. with \geq 1 SH group and its ester 0.05-50

(preferably 2-20), emulsifier 0.05-50 (2-20), and an anionic or nonionic surfactant ≤ 2 (0.05-1 g/L). The emulsifier comprises an alkoxylated and preferably ethoxylated branched C4-20 alc., an alkyl or alkylphenyl ether of polyethylene glycol. Ag, Cu, and their alloys are treated with the emulsion at >T (m.p. of inhibitor), rinsed with H2O at <T, and dried with hot air. An example emulsion of pH 3 and suitable for treatment of Ag and Ag alloys contains octadecanethiol 0.5-30, polyethylene glycol alkyl ether 0.5-30, and SDS ≤ 1 q/L H2O. tarnishing inhibitor silver copper; thiol SDS tarnishing inhibitor silver; SDS thiol tarnishing inhibitor copper; polyethylene glycol ether tarnishing inhibitor Thiols, uses RL: USES (Uses) (corrosion inhibitors, for copper and silver, with emulsifiers of alkyl or alkylphenyl ether of polyethylene glycol) Tarnishing (of silver and copper alloys, aqueous emulsion for prevention of) Corrosion inhibitors (thiols, with emulsifiers of alkyl or alkyl Ph ether of polyethylene glycol) Alcohols, compounds RL: PROC (Process) (C8-16, ethoxylated, corrosion inhibitor emulsion containing, thiol, for copper and silver and their alloys) copper alloy, base silver alloy, base RL: RCT (Reactant); RACT (Reactant or reagent) (tarnishing of, thiol inhibitor for) 25322-68-3D, Polyethylene glycol, alkyl and alkylphenyl ethers 151-21-3, uses RL: PROC (Process) (corrosion inhibitor emulsion containing, thiol, for copper and silver and their alloys) 2885-00-9, Octadecanethiol RL: PROC (Process) (corrosion inhibitors, for copper and silver, with emulsifiers of alkyl or alkylphenyl ether of polyethylene glycol) 7440-22-4, Silver, reactions 7440-50-8, Copper, reactions RL: RCT (Reactant); RACT (Reactant or reagent) (tarnishing of, thiol inhibitor for) L36 ANSWER 17 OF 26 CAPLUS COPYRIGHT 2009 ACS on STN 1991:607232 CAPLUS 115:207232 OREF 115:35345a,35348a Entered STN: 15 Nov 1991 Surface Raman scattering of self-assembled monolayers formed from 1-alkanethiols: behavior of films at gold and comparison to films at silver Bryant, Mark A.; Pemberton, Jeanne E. Dep. Chem., Univ. Arizona, Tucson, AZ, 85721, USA Journal of the American Chemical Society (1991), 113(22), 8284-93 CODEN: JACSAT; ISSN: 0002-7863 Journal Enalish 22-3 (Physical Organic Chemistry) Surface Raman scattering is used to study self-assembled monolayers formed from a series of Me(CH2)nSH (I; n = 3-5, 7, 8, 11, 17) at mech. polished and electrochem. roughened Au surfaces. Defect structure in these films is examined by use of the relative intensities of peaks due

ΙT

ΤТ

ΤT

ΙT

TT

ΙT

ΤТ

ΑN DN

ΕD

ΤI

ΑU

CS

DT

LA

CC

AΒ

to trans and gauche conformations in the $\nu(\text{C-S})$ and $\nu(\text{C-C})$ frequency regions. Surface selection rules for Raman spectroscopy are used to estimate orientation of the I layers at Au. The orientation proposed on the basis of the Raman spectral data is consistent with those previously reported on the basis of other measurements at Au surfaces. This orientation is compared to that previously determined for these films at Ag, which have a chain tilt from the surface normal less than the 30° previously reported for Au. The C-S bond is perpendicular to the Ag surface, but largely parallel to the surface at Au. Differences in the spectra of short-chain I from smooth and rough surfaces are attributed to disordering of the film at the roughened Au surface, which occurs predominantly near the S end of the mol. on rough Au surfaces.

ST gold surface Raman selfassembled alkanethiol monolayer; silver surface Raman selfassembled alkanethiol film

IT Raman spectra

(of alkanethiols in self-assembled monolayers at roughened or smooth gold surfaces)

IT Molecular association

Molecular orientation

(of alkanethiols on roughened or smooth gold surfaces)

IT Molecular vibration

(of alkanethiols, effect of self-assembled monolayers formed on roughened or smooth gold surfaces on)

IT Conformation and Conformers

(of alkenethiols in self-assembled monolayers on roughened or smooth gold surfaces)

IT Films

(self-assembled monolayer, of alkanethiols on roughened or smooth gold surfaces)

IT Thiols, properties

RL: PRP (Properties)

(surface self-assembled monolayers of, on roughened or smooth gold surfaces, Raman spectra in relation to conformation and orientation in)

IT 7440-57-5P, Gold, preparation

RL: PREP (Preparation)

(self-assembled monolayers of alkanethiols on)

IT 109-79-5, Butanethiol 110-66-7, Pentanethiol 111-31-9, Hexanethiol 111-88-6, 1-Octanethiol 112-55-0, 1-Dodecanethiol 1455-21-6, Nonanethiol 2885-00-9, 1-Octadecanethiol

RL: PRP (Properties)

(self-assembled monolayers of, on gold surfaces, Raman spectra of)

- L36 ANSWER 18 OF 26 CAPLUS COPYRIGHT 2009 ACS on STN
- AN 1991:216967 CAPLUS
- DN 114:216967
- OREF 114:36393a,36396a
- ED Entered STN: 31 May 1991
- TI Surface Raman scattering of self-assembled monolayers formed from 1-alkanethiols at silver [electrodes]
- AU Bryant, Mark A.; Pemberton, Jeanne E.
- CS Dep. Chem., Univ. Arizona, Tucson, AZ, 85721, USA
- SO Journal of the American Chemical Society (1991), 113(10), 3629-37 CODEN: JACSAT; ISSN: 0002-7863
- DT Journal
- LA English
- CC 73-3 (Optical, Electron, and Mass Spectroscopy and Other Related Properties)
- AB Surface Raman scattering is used to study self-assembled monolayers formed from a series of 1-alkanethiols (1-butanethiol, 1-dodecanethiol, 1-octadecanethiol) at both electrochem. roughened and mech. polished polycryst. Ag electrodes. The spectra obtained

```
at both surfaces are similar in all spectral regions. Defect structure in
     these films is investigated using the relative amts. of trans and gauche
     conformers in the \nu \, \text{(C-S)} and \nu \, \text{(C-C)} frequency regions. These
     monolayer films are most ordered in the cases of 1-butanethiol and
     1-octadecanethiol and least ordered in the case of 1-dodecanethiol. This
     behavior correlates with the ordering observed in the bulk 1-alkanethiols.
     Surface selection rules are used to determine mol. orientation at Aq.
     surface Raman alkanethiol silver electrode; thiol alkane surface
    Raman silver electrode; butanethiol monolayer silver
     surface Raman; dodecanethiol monolayer silver surface Raman;
     octadecanethiol monolayer silver surface Raman
ΤТ
     Surface
       (Raman scattering of alkanethiols at silver electrode)
ΙT
    Electrodes
        (silver, surface Raman scattering of self-assembled
       monolayers formed from alkanethiols at)
     Thiols, properties
ΤT
     RL: PRP (Properties)
        (surface Raman scattering of self-assembled monolayers formed from, at
       silver electrodes)
ΤT
    Raman spectra
        (surface scattering of self-assembled monolayers formed from
       alkanethiols at silver electrodes)
ΙT
     7440-22-4, Silver, properties
     RL: PRP (Properties)
        (surface Raman scattering of self-assembled monolayers formed from
       alkane thiols at electrodes of)
ΙT
     109-79-5, 1-Butanethiol 112-55-0, 1-Dodecanethiol 2885-00-9,
     1-Octadecanethiol
     RL: PRP (Properties)
       (surface Raman scattering of self-assembled monolayers formed from, at
       silver electrodes)
L36 ANSWER 19 OF 26 CAPLUS COPYRIGHT 2009 ACS on STN
   1991:86953 CAPLUS
DN
    114:86953
OREF 114:14759a,14762a
ED Entered STN: 09 Mar 1991
TI Protection of silver parts from tarnishing
PA Blasberg-Oberflaechentechnik G.m.b.H., Germany
SO Ger. Offen., 3 pp.
    CODEN: GWXXBX
DT
   Patent
LA
    German
    ICM C23C022-03
TC
    ICS H01R043-00
ICA H01R013-629
     56-6 (Nonferrous Metals and Alloys)
     Section cross-reference(s): 76
FAN.CNT 1
     PATENT NO.
                       KIND
                               DATE
                                         APPLICATION NO. DATE
                        ____
                               _____
                                          _____
PI DE 3905850 A1 19900830
PRAI DE 1989-3905850 19890224
                                         DE 1989-3905850
                                                                 19890224
CLASS
             CLASS PATENT FAMILY CLASSIFICATION CODES
PATENT NO.
 _____
                ____
 DE 3905850
               ICM
                      C23C022-03
                ICS
                      H01R043-00
                      H01R013-629
                ICA
                IPCI
                      C23C0022-03 [ICM,5]; C23C0022-02 [ICM,5,C*];
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H01R0043-00 [ICS,5]; H01R0013-629 [ICA,5]

AB The parts (e.g. elec. contacts) are treated with a solution of a long-chain mercapto compound (stearyl mercaptan, cetyl mercaptan) in C4-5 glycol and/or glycol ether (1-methoxy-2-propanol, Bu glycol, and/or methoxybutanol) with or without subsequent rinsing with solvent, water, or warm aqueous detergent. Successful protection of Ag parts in H2S atmospheric by treatment with the invention solution was demonstrated.

ST silver protection stearyl mercaptan; cetyl mercaptan silver protection; hydrogen sulfide silver protection mercaptan

IT Electric contacts

(silver tarnishing of, in hydrogen sulfide atmospheric, treatment with cetyl or stearyl mercaptan for prevention of)

IT 7783-06-4, Hydrogen sulfide, uses and miscellaneous

RL: USES (Uses)

(tarnishing by, of silver, in sulfide atmospheric,

treatment with cetyl or stearyl mercaptan for prevention of)

IT 7440-22-4, Silver, reactions

RL: RCT (Reactant); RACT (Reactant or reagent)

(tarnishing of, in hydrogen sulfide atmospheric, treatment with cetyl or stearyl mercaptan for prevention of)

IT 2885-00-9, Stearyl mercaptan 2917-26-2, Cetyl mercaptan

KIND DATE

RL: USES (Uses)

(treatment with glycol or glycol ether of, of silver, for tarnishing prevention)

L36 ANSWER 20 OF 26 CAPLUS COPYRIGHT 2009 ACS on STN

AN 1985:118206 CAPLUS

DN 102:118206

OREF 102:18526h, 18527a

ED Entered STN: 06 Apr 1985

TI Tarnish inhibitors for gold and silver

PA Alps Electric Co., Ltd., Japan

SO Jpn. Kokai Tokkyo Koho, 5 pp.

CODEN: JKXXAF

DT Patent

LA Japanese

IC ICM C23F011-16 ICS C23F011-12

PATENT NO.

CC 56-10 (Nonferrous Metals and Alloys)

Section cross-reference(s): 76

FAN.CNT 1

PI JP 59215490 JP 61055596		A 19841205 JP 1983-89183 19830 B 19861128	1523					
PRAI JP 1983-891	83	19830523						
CLASS								
PATENT NO.	CLASS	PATENT FAMILY CLASSIFICATION CODES						
			-					
JP 59215490 ICM		C23F011-16						
	ICS	C23F011-12						
	IPCI	C23F0011-16 [ICM,3]; C23F0011-12 [ICS,3]; C23F00	11-10					
		[ICS, 3, C*]						
	IPCR	C23F0011-00 [I,C*]; C23F0011-00 [I,A]; C23F0011-	-10					
		[I,C*]; C23F0011-10 [I,A]						

APPLICATION NO. DATE

AB The inhibitors contain linear alkyl mercaptan, poly(oxyethylene) nonylphenol ether, and iso-Pr alc. The inhibitors prevent discoloration of Au, Ag, or their alloys, and do not increase the elec. resistance. Thus, a phosphor bronze sheet coated with Ag 3μ thick was treated with a tarnish inhibitor composed of

```
n-hexadecyl mercaptan [2917-26-2] 2 + 10-4,
    poly(oxyethylene) nonylphenol ether [9016-45-9] (ethylene oxide 9 mol%)
    0.07, iso-Pr alc. 5%, and balance water. The sheet exposed in an atmospheric
at.
    20° containing H2S 0.05 ppm showed no change in the surface condition,
    with decreased discoloration and elec. contact resistance.
ST
    gold silver tarnishing inhibitor; hexadecyl mercaptan
    tarnish inhibitor; tetradecyl mercaptan tarnish
    inhibitor; polyoxyethylene nonylphenol ether tarnish inhibitor;
    isopropanol tarnish inhibitor silver
ΙT
    Tarnishing
       (inhibitors, for gold and silver)
ΙT
    Corrosion inhibitors
       (tarnishing, for gold and silver)
ΙT
    2079-95-0 2917-26-2 9016-45-9
    RL: USES (Uses)
       (in tarnish inhibitor, for gold and silver)
    7440-02-0, uses and miscellaneous
ΤТ
    RL: USES (Uses)
        (tarnish inhibitor for gold on brass plated with)
ΤT
    12597-71-6, uses and miscellaneous
    RL: USES (Uses)
       (tarnish inhibitor for gold- or silver-plated)
ΙT
    12767-50-9
    RL: USES (Uses)
       (tarnish inhibitor for silver-plated)
ΙT
    7440-22-4, uses and miscellaneous
    RL: USES (Uses)
       (tarnish inhibitors for)
    7440-57-5, uses and miscellaneous
ΤT
    RL: USES (Uses)
       (tarnishing inhibitors for)
L36 ANSWER 21 OF 26 CAPLUS COPYRIGHT 2009 ACS on STN
   1978:157165 CAPLUS
DN
   88:157165
OREF 88:24723a,24726a
   Entered STN: 12 May 1984
   Prevention of tarnishing on silver or its alloy
    products
IN
    Kawana, Yasuo; Ara, Takeo
PA
    Alps Electric Co., Ltd., Japan
SO
    Jpn. Kokai Tokkyo Koho, 6 pp.
    CODEN: JKXXAF
DТ
    Patent
LA
    Japanese
IC
    C23F007-00
    56-5 (Nonferrous Metals and Alloys)
    Section cross-reference(s): 76
FAN.CNT 1
                                        APPLICATION NO. DATE
    PATENT NO.
                       KIND
                              DATE
     _____
                       ____
                              _____
                                         ______
    JP 52111430
JP 56001396
                      А
                              19770919
                                         JP 1976-27793
                                                           19760315
                       В
                              19810113
PRAI JP 1976-27793 A
                              19760315
CLASS
PATENT NO. CLASS PATENT FAMILY CLASSIFICATION CODES
               ____
               IC
 JP 52111430
                      C23F007-00
                IPCI C23F0007-00; C23F0011-12; C23F0011-16; C23F0011-10 [C*]
                IPCR C23C0022-05 [I,C*]; C23C0022-60 [I,A]; C23F0011-00
                      [I,C*]; C23F0011-00 [I,A]; C23F0011-10 [I,C*];
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C23F0011-12 [I,A]; C23F0011-16 [I,A]; H01H0001-00 [I,C*]; H01H0001-00 [I,A]

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AB
    A mixture containing mercapto compound, alkaline compound, alc., and organotin
laurate.
     e.g., Bu2Sn dilaurate, is used to prevent tarnish on Ag
     , and stabilizes elec. contact resistance. Thus, the mixture contained
     lauryl mercaptan [112-55-0] 2.0, cetyl mercaptan [2917-26-2]
     0.5, 2-naphthalenethiol [91-60-1] 0.1, benzotriazole [95-14-7] 0.1,
     dioctyltin dilaurate [3648-18-8] 1.0, polyoxyethylene alkyl ether 1.0,
     polyoxyethylene alkyl ester 1.0, 28% NH4OH solution 20, EtOH 10, and water
     64.3%. The tarnish rating ofg Ag with the coating was
     .apprx.1 after exposing 60 h to a H2S-NH3 atmospheric vs. .apprx.10 after
     treatment 25 h with a com. mixture
ST
    silver tarnish preventing mixt; elec contact
    silver tarnish prevention
IT
    Coating materials
        (Coating materials for tarnish prevention of silver)
ΤТ
     Tarnishing
       (of silver, coating for prevention of)
     91-60-1 95-14-7 112-55-0 2917-26-2 3648-18-8
ΤT
     RL: USES (Uses)
        (in tarnish-preventing coating for silver)
ΙT
     7440-22-4, reactions
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (tarnishing of, coatings for prevention of)
L36 ANSWER 22 OF 26 CAPLUS COPYRIGHT 2009 ACS on STN
    1970:405261 CAPLUS
AN
DN
     73:5261
OREF 73:899a,902a
ED Entered STN: 12 May 1984
   Liquid silver polishing agents
ΤI
   Thornton, James C.; Nixon, C. P.; Cox, Bernard C.
ΤN
PA
    Goddard, J., and Sons Ltd.
SO
    Ger. Offen., 13 pp.
    CODEN: GWXXBX
DT
    Patent
   German
LA
IC
    C11D007-34
CC
    46 (Surface Active Agents and Detergents)
FAN.CNT 1
                                      APPLICATION NO. DATE
                   KIND DATE
    PATENT NO.
                              19700416 DE 1969-1932524 19690626
19701231 GB 1968-31214 19680629
    DE 1932524
PΤ
                        A
                        A
     GB 1217414
                              19691201 BE 1969-734968
     BE 734968
                        Α
                                                               19690623
BE 734908 A 19691201 BE 1969-734968
NL 6909896 A 19691231 NL 1969-9896
FR 2011801 A5 19700306 FR 1969-22017
PRAI GB 1968-31214 A 19680629
                                                                 19690627
                                                                 19690630
CLASS
 PATENT NO. CLASS PATENT FAMILY CLASSIFICATION CODES
 _____
                       ______
                IC
 DE 1932524
                       C11D007-34
                 IPCI
                       C11D0007-34; C11D0007-22 [C*]
                       C11D0007-02 [I,C*]; C11D0007-08 [I,A]; C11D0007-22
                 TPCR
                       [I,C*]; C11D0007-34 [I,A]; C23F0011-10 [I,C*];
                       C23F0011-16 [I,A]; C23G0001-02 [I,C*]; C23G0001-10
                       [I,A]
 GB 1217414
                IPCI
                       C11D0007-34; C11D0007-22 [C*]
                       C11D0007-02 [I,C*]; C11D0007-08 [I,A]; C11D0007-22
                 IPCR
                       [I,C*]; C11D0007-34 [I,A]; C23G0001-02 [I,C*];
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C23G0001-10 [I,A]

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C23F011/16B; C11D007/08; C11D007/34; C23G001/10
                ECT.A
BE 734968
                TPCT
                       C11D0007-34; C11D0007-22 [C*]
NL 6909896
                IPCI
                       C11D0007-34 [ICM]; C11D0007-22 [ICM, C*]; C23G0001-02
                       [ICS]
                IPCR
                       C11D0007-02 [I,C*]; C11D0007-08 [I,A]; C11D0007-22
                       [I,C*]; C11D0007-34 [I,A]; C23F0011-10 [I,C*];
                       C23F0011-16 [I,A]; C23G0001-02 [I,C*]; C23G0001-10
                       C23F011/16B; C11D007/08; C11D007/34; C23G001/10
                ECLA
FR 2011801
                IPCI
                       C23G0001-00 [ICM]
                IPCR
                       C11D0007-02 [I,C*]; C11D0007-08 [I,A]; C11D0007-22
                       [I,C*]; C11D0007-34 [I,A]; C23F0011-10 [I,C*];
                       C23F0011-16 [I,A]; C23G0001-02 [I,C*]; C23G0001-10
                       [I,A]
                       C23F011/16B; C11D007/08; C11D007/34; C23G001/10
                ECLA
    Liquid Ag-polishing agents with tarnish
AΒ
    -resistant activities are described. They contain an acid stronger than
    H2S, at least 1% CS(NH2)2 or H2NCSNHNH2, forming a soluble Ag
    complex, 0.5-2% SH-containing compound, e.g. n-C16H31SH, n-C18H37SH, or stearyl
    or cetyl thioglycolate, forming a transparent, colorless protective layer
    on Ag surfaces, and an emulsifier. Thus, a solution contained
    H2NCSNH2 4.9%, H2SO4 (d. 1.84) 0.81%, HCl (d. 1.16) 0.38%, ethoxylated
    aliphatic C12-18 amine 0.83%, stearyl mercaptan 0.88%, Solvay Blue PFN 125
    0.01%, PrOH 0.011%, rest H2O. The Ag objects were cleaned by
    immersion in an Al basket in the above solution with formation of a galvanic
    cell.
    silver polishing tarnishproofing; polishing
ST
    silver tarnishproofing; tarnishproofing silver
    polish
    62-56-6, uses and miscellaneous 2885-00-9
ΤТ
    RL: USES (Uses)
       (polishing materials containing, for silver)
    7440-22-4, uses and miscellaneous
ΤТ
    RL: USES (Uses)
        (polishing materials for, sulfur compound-containing)
L36 ANSWER 23 OF 26 CAPLUS COPYRIGHT 2009 ACS on STN
AN
    1969:39127 CAPLUS
DN
    70:39127
OREF 70:7349a,7352a
   Entered STN: 12 May 1984
TΙ
    Silver polish containing thio compounds
IN
    Schlegel, Hans; Straub, Ewald; Bauer, Martin
PA
    Wuerttembergische Metallwarenfabrik
SO
    Ger., 2 pp.
    CODEN: GWXXAW
DT
    Patent
LA
    German
IC
    C23F; C23G
CC
    46 (Surface Active Agents and Detergents)
FAN.CNT 1
                                         APPLICATION NO. DATE
    PATENT NO.
                       KIND
                               DATE
                        ____
                               _____
                                           _____
PI DE 1282414 B
PRAI DE 1963-W35841 A
                               19681107 DE 1963-W35841
                                                            19631218
                               19631218
CLASS
PATENT NO. CLASS PATENT FAMILY CLASSIFICATION CODES
              IC C23F; C23G
DE 1282414
                IPCR C11D0007-22 [I,C*]; C11D0007-34 [I,A]; C23F0003-00
                       [I,C*]; C23F0003-04 [I,A]; C23F0011-10 [I,C*];
                       C23F0011-16 [I,A]
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Ag articles can be passivated by immersion or spraying with organic
AΒ
    solvents containing ≤50% alkyl thioglycolates having 8 C atoms or by
    polishing with polishers containing 0.5-10% alkyl
    thioglycolates. The polisher can be prepared from kieselguhr 5,
    siliceous chalk 3, Me cellulose 1, and n-hexadecyl thioglycolate 1 kg./50
    1. distilled H2O.
ST
    silver polish thio compd; thio compd Aq
    polish
ΙT
    Polishing materials
       (alkyl mercaptoacetate-containing, for silver)
    7440-22-4, uses and miscellaneous
ΙT
    RL: USES (Uses)
       (polishing material for, alkyl mercaptoacetate-containing)
ΙT
    22811-02-5
    RL: USES (Uses)
       (polishing materials containing, for silver)
L36 ANSWER 24 OF 26 CAPLUS COPYRIGHT 2009 ACS on STN
    1967:484494 CAPLUS
ΑN
DN 67:84494
OREF 67:15951a,15954a
ED
    Entered STN: 12 May 1984
    Metal cleaning
    Kroll, Harry; Therrien, Alderic R., Jr.; Bennett, Phyllis W.
    Phillip A. Hunt Chemical Corp.
PA
SO
    U.S., 3 pp.
    CODEN: USXXAM
   Patent
DT
LA English
INCL 106003000
CC 56 (Nonferrous Metals and Alloys)
FAN.CNT 1
                                       APPLICATION NO.
                                                             DATE
    PATENT NO.
                KIND DATE
    _____
                             _____
                                        _____
                      ____
                             19670711 US 1964-378993
PI US 3330672
                                                            19640629
CLASS
           CLASS PATENT FAMILY CLASSIFICATION CODES
PATENT NO.
 ______
US 3330672
              INCL 106003000
               IPCR C23F0011-10 [I,C*]; C23F0011-10 [I,A]; C23G0001-00
                      [I,C*]; C23G0001-00 [I,A]
                      106/003.000; 106/008.000; 106/014.130; 106/014.150;
                      106/014.210; 106/014.420; 106/014.430; 106/014.440;
                      148/271.000; 252/390.000; 252/395.000
    Ag and metals chemical similar to Ag are treated with a
AB
    composition that removes tarnish, cleanses the metal surface,
    protects the Ag against further tarnish, and imparts a
    high surface lustre. The active ingredients of the composition are 0.1-10.0
    parts of mercapto esters and 0.1-10.0 parts of amine salts. The structure
    of the mercapto esters is: HS(CH2)nCO2R where n is 1 or 2 and R is alkyl
    of 12-18 C. The structure of the amine salts is: [R1R2R3N+H]X- where R1
    is alkyl of 8-20 C and R2 and R3 are H, Me, and Et, and X is the anion
    derived from a low mol. weight, aqueous soluble organic acid as acetic,
formic, citric,
    malic, maleic, fumaric, etc. A polishing abrasive may be added
    to the composition in 0.1-10.0 parts. The active ingredients may be used with
    a liquid carrier or a finely divided solid abrasive carrier. The liquid
    carrier may be water, iso-PrOH, or chlorinated hydrocarbon. The abrasive
    carrier may be pumice, rouge, diatomaceous earth, CaCO3, or any usual
    polishing abrasive.
    CLEANING COMPN AG; SILVER CLEANING COMPN
ST
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ΤТ

Tallow

```
RL: PRP (Properties)
        (amines from, acetates, compns. containing, for tarnish removal
       from silver)
    Tarnish
IΤ
       (removal of, from silver, compns. for)
    Metals, uses and miscellaneous
ΙT
    RL: USES (Uses)
       (tarnish removal from, compns. for)
    2190-04-7 3746-39-2 10220-46-9 17369-34-5 17369-37-8
ΙT
    RL: USES (Uses)
       (compns. containing, for tarnish removal from silver)
    7440-22-4P, uses and miscellaneous
ΤТ
    RL: PREP (Preparation); USES (Uses)
        (tarnish removal from, compns. for)
L36 ANSWER 25 OF 26 CAPLUS COPYRIGHT 2009 ACS on STN
    1967:465804 CAPLUS
AN
DN 67:65804
OREF 67:12443a,12446a
   Entered STN: 12 May 1984
ED
    Metal cleaning, polishing or protecting preparations
ΙN
    Ford, Ian A. M.; Cox, Bernard C.; Thornton, James C.
    Goddard, J., and Sons Ltd.
SO
    Brit., 3 pp.
    CODEN: BRXXAA
    Patent
DT
   English
LA
ΙC
    C23G
CC
    46 (Surface Active Agents and Detergents)
FAN.CNT 1
                 KIND DATE
                                        APPLICATION NO.
                                                               DATE
    PATENT NO.
    _____
                              _____
                                         _____
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                              19670601 GB 1963-25235
    GB 1070383
                                                              19630625
PΤ
                                         DE
    DE 1519159
    US 3518098
                              19700630
                                         US
                                                                19640623
CLASS
PATENT NO. CLASS PATENT FAMILY CLASSIFICATION CODES
 _____
              IC C23G
GB 1070383
                IPCI C23G
                IPCR C09G0001-00 [I,C*]; C09G0001-02 [I,A]; C11D0003-34
                       [I,C*]; C11D0003-34 [I,A]; C23F0011-10 [I,C*];
                       C23F0011-16 [I,A]; C23G0001-02 [I,C*]; C23G0001-10
                      [I,A]
 US 3518098
                      C09G0001-00 [I,C*]; C09G0001-02 [I,A]; C11D0003-34
                IPCR
                       [I,C*]; C11D0003-34 [I,A]; C23F0011-10 [I,C*];
                       C23F0011-16 [I,A]; C23G0001-02 [I,C*]; C23G0001-10
                       [I,A]
                       106/003.000; 106/008.000; 106/014.130; 148/271.000;
                NCL
                       252/395.000
AΒ
    A cleaning, polishing, or protective preparation is provided for
    metal surfaces containing Ag, Cu, or Ni. The preparation may be a solid, liquid, paste, powder, or semi-solid, or it may be in the form of a
    surface-treating compound mixed with a protective medium consisting
    preferably of esters derived from thioglycolic or mercaptopropionic acid
    and a C12-22 aliphatic alc. Thus a suitable preparation consists of
    polishing powder 20, detergent paste 40, stearyl
    mercaptopropionate 2.5, H3PO4 0.2, and H2O up to 100. The paste is prepared
    by warming the mercaptopropionate with an equal weight of detergent paste and
    the resulting liquid added to the dry ingredients with stirring to form a
    stiff paste. Afterward, the liquid content is added slowly while
    stirring.
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METAL CLEANER COATING POLISH; NICKEL CLEANER POLISH;
    COPPER CLEANER POLISH; SILVER CLEANER POLISH
     ; CLEANER METAL; POLISH METAL; COATING METAL
    Metals, uses and miscellaneous
IΤ
     RL: USES (Uses)
        (detergents and polishing materials containing octadecyl
        mercaptopropionate or thioglycollate for)
ΙT
     Detergents, preparation
      Polishing materials
        (octadecyl mercaptopropionate or octadecyl thioglycollate-containing, for
                                         7440-22-4, uses and miscellaneous
ΤТ
     7440-02-0, uses and miscellaneous
     7440-50-8, uses and miscellaneous
     RL: USES (Uses)
        (detergents and polishing materials containing octadecyl
        mercaptopropionate or thioglycollate for)
     10220-46-9
ΤТ
                28986-42-7
     RL: USES (Uses)
        (detergents and polishing materials containing, for metals)
L36 ANSWER 26 OF 26 CAPLUS COPYRIGHT 2009 ACS on STN
ΑN
     1964:485188 CAPLUS
DN
     61:85188
OREF 61:14899c-d
    Entered STN: 22 Apr 2001
    Antitarnish silver polish
ТΤ
ΑU
    Glickman, Charles S.
    Manufacturing Chemist (1930-1963) (1964), 35(9), 57,59
SO
    CODEN: MACSAS; ISSN: 0368-8313
DT
    Journal
LA
    Unavailable
CC
     52 (Coatings, Inks, and Related Products)
     The use of octadecyl thioglycolate (I) as the main ingredient in the
AB
     formulation of satisfactory antitarnish silver polishes
     is discussed; its alkane portion is H2O-insol. and forms a solid film,
     whereas its SH group is active enough to give a suitable resistance to
     tarnishing. I (25% by weight in iso-PrOH) can be used as a
     concentrate containing diatomaceous earth (mixed with Me cellulose or
     bentonite as a suspending agent), a surfactant, and pine oil.
ΤT
    Polishing materials
        (from octadecyl thioglycolate, Ag antitarnishing)
ΙT
     Tarnishing
        (of silver in Br-KBr solns., prevention of, octadecyl
        thioglycolate polishes for)
     Acetic acid, mercapto-, octadecyl ester
ΤТ
        (tarnishing-inhibiting silver polishes
        from)
     10220-46-9
ΙT
        (Derived from data in the 7th Collective Formula Index (1962-1966))
ΙT
     7440-22-4, Silver
        (tarnish-inhibiting polishes for, from octadecyl
        thioglycolate)
=> logoff y
COST IN U.S. DOLLARS
                                                 SINCE FILE
                                                                  TOTAL
                                                      ENTRY
                                                               SESSION
FULL ESTIMATED COST
                                                     112.28
                                                               262.55
DISCOUNT AMOUNTS (FOR QUALIFYING ACCOUNTS)
                                                 SINCE FILE
                                                                 TOTAL.
                                                      ENTRY
                                                              SESSION
CA SUBSCRIBER PRICE
                                                                 -30.34
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-21.32

ST